## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2008 question paper

## 0620 CHEMISTRY

0620/05

Paper 5 (Practical Test), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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| Page 2   | Mark Scheme  | Syllabus                |
|--|--|-------------------------|
|  | IGCSE – May/June 2008  | 0620                    |
| Experiment 1   | (a) green precipitate (1)  | Syllabus A. Day er 0620 |
| Experiment 2   | <b>2(d)</b> brown/orange/rust precipitate (1)                    | Tag                     |
| Table of resu  | lts  |                         |
| Experiment initial and fina                                | 1 al volume boxes correctly completed (1)                        |                         |
| Experiment initial and fina                                | 2 al volume boxes correctly completed (2)                        |                         |
| differences c  | ompleted correctly (1)   |                         |
| comparable to Supervisor's results ± 5 cm <sup>3</sup> (2) |  |                         |
| (e) (pale) gr  | een (1)  | [1]                     |
| (f) (turns) ye   | [1]  |                         |
| (g) (i) Expe   | eriment 1 (1) ecf  | [1]                     |
| (ii) more  | e in Experiment 1/greater volume (1)                             | [1]                     |
| (iii) solu   | tion <b>A</b> more concentrated/stronger than <b>B</b> (1) ×2 (1 | ) [2]                   |
| (h) twice the  | volume value for Experiment 2 (1) cm³/ml (1)                     | [2]                     |
| (i) change   | e.g. repeat titrations (1) or use a burette/pipette              |                         |
| explanat   | ion e.g. average reading more accurate (1) instead               | of m/cylinder [2]       |
| ( <b>j) (i)</b> iron(                                      | (II) present (1)   | [1]                     |
| (ii) iron(   | (III) (1)  | [1]                     |

1

[Total: 20]

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| Page 3 | Mark Scheme           | Syllabus |
|--------|-----------------------|----------|
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## 2 Tests on solid T

| <b>(a)</b> wh                                | ite (1)               | see Supervisor's results            | Tag     |
|--|-----------------------|-------------------------------------|---------|
| <b>(b)</b> cor                               | ndensation (1)        | solid goes black/chars (1)          |         |
| gas ignites/catches fire or flame (1)        |                       |                                     | [max 2] |
| (c) (i)                                      | orange/yellow (1)     | pH6/5 (1) see Supervisor's results  | [2]     |
| (ii)   | white (1)             | precipitate (1)                     | [2]     |
| (iii)  | reddish (1)           | brown on heating (1)                | [2]     |
| Tests o                                      | n solid <b>V</b>      |                                     |         |
| (d) green (1)                                |                       |                                     | [1]     |
| (e) solid changes colour to copper/brown (1) |                       |                                     | [1]     |
| <b>(f)</b> sm                                | ells acidic/vinegar o | similar (1)                         | [1]     |
| (i)  | green (1)             | pH 7/6 (1) see Supervisor's results | [2]     |
| (ii)   | blue (1)              | precipitate (1)                     | [2]     |
| (iii)  | green (1)             |                                     | [1]     |
| (g) cal                                      | cium present (1)      |                                     | [1]     |
| (h) cop                                      | oper present (1)      | ethanoic acid/organic salt (1)      | [2]     |

[Total: 20]

[Total for paper: 40]