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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2008 question paper

0620 CHEMISTRY

0620/05

Paper 5 (Practical Test), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

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Syllabus 0620

1	Experiment 5 observations rapid/vigorous o w t t e (1) bubbles/fizz (1)			
	rapid/vigorous o.w.t.t.e (1) bubbles/fizz (1) blue precipitate (1) [m			
	Table of results initial boxes correctly completed (1) final boxes correctly completed (1) differences correctly completed (1) comparable to Supervisor's results (1)	[5]		
	(a) all 4 bars correctly drawn (3), -1 for each incorrect labelled (1)	[4]		
	(b) (i) solid A/Experiment 1 (1)	[1]		
	(ii) temperature increased/heat/energy given out (1)	[1]		
	(c) Experiment 3/solid C or from results (1)	[1]		
	(d) (i) double the value or 2 × value from table (1) e.c.f. applies	[1]		
	(ii) half the value or half the value from table (1)	[1]		
	(iii) more/larger volume of water (1) twice as much (for solid to dissolve in) o.w.t.t.e. (1)	[2]		
	(e) acid present (1) carbonate present (1) carbon dioxide (1)			
	[Total:	20]		
2	(a) solution K blue/green solution L yellow/orange/brown (1)	[1]		
	 (b) check pHs from Supervisor's results pH of solution K 1-6 pH of solution L 1-6 (1) 	[1]		
	tests on solution K			
	(c) (i) blue (1) precipitate (1)	[2]		
	(ii) deep/royal blue (1) solution/precipitate dissolves (1)	[2]		
	(iii) no reaction/change/nothing (1) not lighter	[1]		
	(iv) white (1) precipitate (1)	[2]		

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[2]

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tests on solution L

(iv) white (1) precipitate (1) not lighter/paler

(a) (ı)	(rusty) brown (1) precipitate (1)	So
(ii)	brown precipitate (1) not red	[1]
(iii)	no reaction/change/nothing (1)	[1]

(e)	acidic ((1))	11	

(f) chloride/
$$Cl^-(1)$$

[Total: 20]

[Total for paper: 40]