

## **Cambridge International Examinations**

Cambridge International General Certificate of Secondary Education

CHEMISTRY 0620/63

Paper 6 Alternative to Practical

October/November 2016

MARK SCHEME
Maximum Mark: 40

## **Published**

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Page 2	Mark Scheme	Syllabus	Paper
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Question	Answer	Mark
1(a)	(clamp / retort) stand trough	1 1
1(b)	to absorb/hold/keep/soak up/contain the paraffin oil	1
1(c)	M1 bromine (aqueous/in cyclohexane) M2 turns colourless/decolourised	1
1(d)	to prevent suck back (of water)	1

Page 3	Mark Scheme	Syllabus	Paper
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Question	Answer	Mark
2(a)	table of results for Experiment 1 temperature boxes completed correctly 20, 20, 25, 28, 31, 33, 34, 35, 36, 36	2
2(b)	table of results for Experiment 2 temperature boxes completed correctly 22, 22, 21, 71, 76, 75, 72, 70, 67, 65, 64	2
2(c)	all points correctly plotted $\pm$ half a small square smooth line graphs labelled	2 1 1
2(d)(i)	working shown clearly as construction lines or cross value from graph (29–30 $^{\circ}\text{C}\pm0.5^{\circ}\text{C})$	1
2(d)(ii)	value from graph (72s) -60s	1
2(e)	room temperature or initial temperature from table (20–22 °C) reaction has finished/stopped	1
2(f)	more readings/points/data smoother curve/better or more accurate graph	1
2(g)	polystyrene is an insulator/copper is a (good) conductor reduced heat losses	1

Page 4	Mark Scheme	Syllabus	Paper
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Question	Answer	Mark
3(a)(i)	pH 1–3	1
3(a)(ii)	effervescence/fizzing/bubbling/solid disappears/dissolves lighted splint 'pops'	1 1 1
3(a)(iii)	effervescence/fizzing/bubbling/solid disappears/dissolves limewater milky	1 1 1
3(a)(iv)	white precipitate	1
3(b)	calcium/Ca <sup>2+</sup> hydroxide/OH <sup>-</sup>	1

Page 5	Mark Scheme	Syllabus	Paper
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Question	Answer	Mark
4		6
	silica	
	filter (the cleaner)	
	wash the residue	
	dry the residue	
	water	
	heat (the filtrate/cleaner)	
	condense the vapour	
	sodium carbonate	
	heat to dryness/no liquid left	
	(then solid) sodium carbonate is left	
	OR	
	heat until saturated	
	then cool to crystallise / leave to crystallise	