

MARK SCHEME for the October/November 2013 series

0652 PHYSICAL SCIENCE

0652/62

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2013 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.

Page 2	Mark Scheme	Syllabus Syllabus
T dge Z	IGCSE – October/November 2013	0652
(a) (i)	brown colour around where the seeds/grains were, (blue/black elsewhere) ;
(ii)	enzyme/seeds/grains breaking down/use/need/con	Syllabus 0652 blue/black elsewhere) ; werted the <u>starch</u> ; ds, ORA ;
(iii)	control/shows that breakdown depends on living seed	ds, ORA ; [1
	(reducing) sugar is made (around the seeds/when down);	n the starch is broken [1
(ii)	<u>starch</u> digested/changed/converted to <u>sugar</u> ;	[1
(c) (imp	proved) <u>reliability</u> /because one seed might not be activ	ve/owtte ; [1
(d) (i)	larger brown areas ;	[1
(ii)	smaller brown areas (accept no brown area/all blue-b	black); [1
dish keep	p (named) conditions constant ;	rent parts) of the same
com	npare diameters/sizes of brown areas ;	[max 2
		[Total: 10]
	68. <u>0</u> ; 86.2 (±0.1) ;	[2
(ii)	8(0), 16.2 (ecf) ; ;	[2
	points plotted correctly ; (allow 1 error) suitable straight line drawn ;	[2
	clear evidence on graph ; allow 1.2 to 1.3 inclusive ;	[2
(c) 150/	/candidates answer = between 125 and 115g (ecf) ;	[1
(d) den:	sity = $m/l \times t \times w$ (any order);	[1

· <u>- 9-</u>	3 Mark Scheme	Syllabus r
	IGCSE – October/November 2013	0652 22
(a) (i)	observation: bubbles ;	an
	<i>conclusion:</i> transition elements/metals ;	010
(ii)) milky/white/cloudy chalky ;	Syllabus 0652 [1]
(iii)	observation: white precipitate ;	[1]
(b) (i)) hydroxide	[1]
(::)	(()) () () () () () () () ()	
(ii)	<i>test:</i> litmus (ignore colour) or Universal Indicator <i>observation:</i> blue ;	; [2]
(iii)	brown, yellow or orange ;	[1]
	on(III) chloride FeC l_3 and copper carbonate CuCO ₃	
	R on(III) carbonate Fe₂(CO₃)₃ and copper chloride Cu Ilow 1 for two correct names or formulae)	Cl ₂ ;; [max 2]
		[Total: 10]
(a) (i)	11.7 cm (no tolerance) ; 13.9 cm (no tolerance) ;	[2]
(ii)		
	not starting <i>y</i> axis at 0 ; smooth curve drawn ;	[3]
(iii)	0.0417 or 0.042 cm/g;	[0]
	0.01 or 0.010 cm/g;	[2]
(b) (i)		•
	between beats/smoothes out blood flow/ <u>change</u>	in pressure ; [1]
(ii)	resistant to bursting/breaking/tearing ;	[1]
(c) e.ę	g. same width of sample taken/same part of body o	of animal/same animal; [1]

Page 4	Mark Scheme Syllab	ous & r
	IGCSE – October/November 2013 0652	2 100
2	25.1 ; 27.8 ;	bus 2 2 1 1 1 1
(ii) (C is the most concentrated, A is the least concentrated ;	[1
(b) (i) s	sodium ethanoate + water ;	[1
(ii) c	orange or yellow, (reject red) ;	[1
witho <u>evap</u> filter dry c (any OR <u>evap</u> (heat leave filter	rystals with e.g. filter paper ; 4) t/boil etc.) to concentrate/crystallisation point/saturation etc. ; e or cool ; ; rystals with e.g. filter paper ;	[max 4
		[Total: 10]
(a) (i) t	he amplitude decreases/gets smaller ;	[1
(ii) 4	4.0 cm (±0.1 cm) ;	[1
	requency = speed/wavelength, 10/4 ; = 2.5(Hz) ;	[2
(b) (i) 1	1.1 cm (±0.1 cm) ;	[1
	(1.1/0.25) = 4.4 (ecf) ; cm/s ;	[2
		[1
	distance = $2.2 \text{ cm} (\pm 0.1 \text{ cm})$;	['
(c) (i) (distance = 2.2 cm (±0.1 cm) ; speed = 2.2/0.25 = 8.8 (ignore units, ecf) ;	[1

[Total: 10]