	UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education
CANDIDATE NAME	
ENTRE IUMBER	CANDIDATE NUMBER
HYSICS	0625/06
aper 6 Alterr	ative to Practical October/November 2007
	1 hour swer on the Question Paper. Materials are required.

Write in dark blue or black pen.

You may use a soft pencil for any diagrams, graphs or rough working. Do not use staples, paper clips, highlighters, glue or correction fluid. DO NOT WRITE IN ANY BARCODES.

Answer all questions.

At the end of the examination, fasten all your work securely together. The number of marks is given in brackets [] at the end of each question or part question.

For Examiner's Use			
1			
2			
3			
4			
5			
Total			

This document consists of **11** printed pages and **1** blank page.



- 2
- 1 The IGCSE class is investigating the rate of cooling of hot water.
 - (a) A student measures room temperature. Write down the value of room temperature shown on the thermometer in Fig. 1.1.

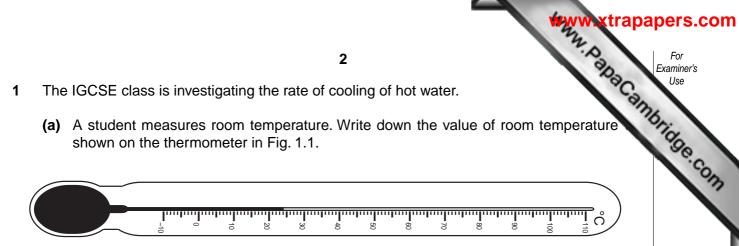


Fig. 1.1

*θ*₀ =[1]

(b) He then pours hot water into a beaker until it is about two-thirds full. He measures and records the temperature θ of the hot water and at the same time starts a stopwatch. As the water cools, he records the temperature every 30s for a total of five minutes. His readings are shown in the table below.

t/	θ/
0	68.0
30	53.0
60	45.0
90	40.0
120	36.5
150	33.5
180	32.0
210	30.0
240	29.0
270	28.5
300	28.0

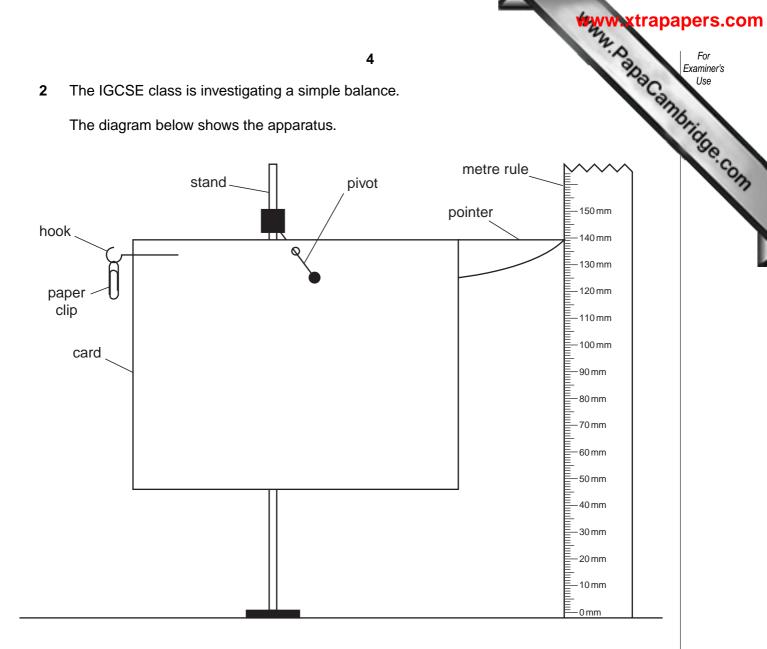
- (i) Complete the column headings in the table.
- (ii) Calculate the temperature fall T_1 in the first minute of the experiment.

*T*₁ =

(iii) Calculate the temperature fall T_2 in the final minute of the experiment.

*T*₂ =

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		3
	-1	Par Par
с) Т d	neo liffe	ory suggests that the rate of cooling of the hot water at any time depends rence between the temperature of the water at that time and room temperature
(i)	State and explain whether your answers in (b) support this theory.
		3 bry suggests that the rate of cooling of the hot water at any time depends rence between the temperature of the water at that time and room temperature State and explain whether your answers in (b) support this theory.
		Explanation
(i	i)	Suggest three variables that you would attempt to keep constant if this theory were
		to be investigated further.
		1
		2
		<i>L</i>
		3
d) S	Sug	gest one addition you could make to the apparatus provided in order to reduce the
ra	ate	of evaporation of the water in the beaker.
		[1]
		[Total: 9]

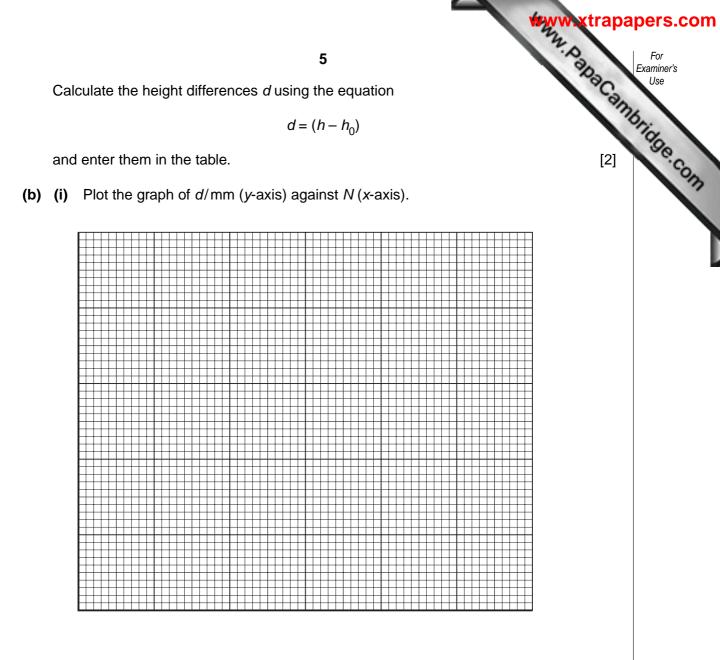




(a) A student records the height h_0 of the pointer above the bench. She then hangs a paper clip on the hook and records the new height *h* of the pointer above the bench. Next she records the heights of the pointer above the bench using different numbers *N* of paper clips. The readings are shown in the table below.

Ν	<i>h</i> /mm	d/mm	
1	108		
2	114		
3	120		
4	125		
5	134		
6	141		

 $h_0 = 100 \,\mathrm{mm}$

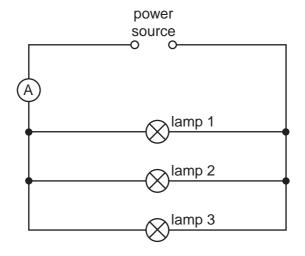


(ii) Use your graph to predict the value of *d* if a nail with the same mass as 4.6 paper clips were to be hung from the hook in place of the paper clips. Show clearly on the graph how you obtained your value.

d =[6]

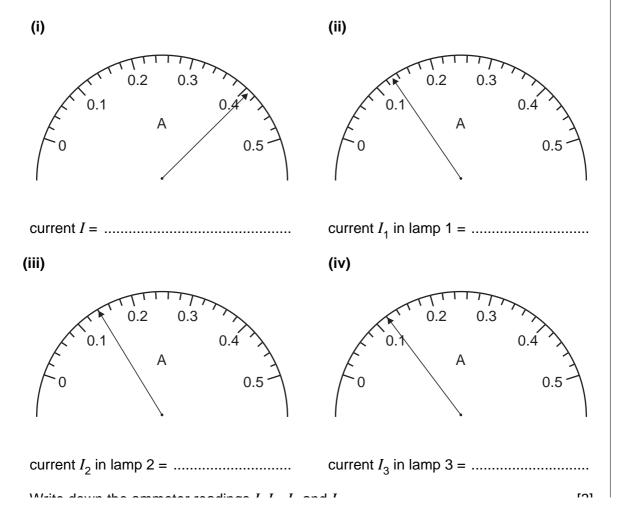
[Total: 8]

- Www.papacambridge.com 3 The IGCSE class is investigating the potential difference across lamps and the cul the lamps.
 - Fig. 3.1 shows the circuit that is being used.





(a) A student uses the ammeter to record the current I in the wire connecting the power source to the rest of the circuit. He then moves the ammeter to new positions in the circuit and measures the current in each lamp in turn. The positions of the pointer on the ammeter scale are shown below.



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		7	For Examiner's
(b)	The theo	eory suggests that $I = I_1 + I_2 + I_3$. State whether or not your readings support. Give a reason for your answer.	For Examiner's Use
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		[1	1 I
(c)	To f vary	test the theory further, you would need to vary the value of <i>I</i> . State how you would y <i>I</i> .	- k
	•••••		
		[1]
(d)	The	e student uses a voltmeter to measure the potential difference V across the lamps.	
	His	reading is $V = 1.6$ V.	
	(i)	Calculate the resistance R of the lamps arranged in parallel, using the equation	
		R = V/I,	
		where <i>I</i> is the value of the current in (a)(i) .	
		R =	
	(ii)	On Fig. 3.1, add the symbol for the voltmeter connected to measure the potential difference across the lamps. [3	
		[Total: 8	;]

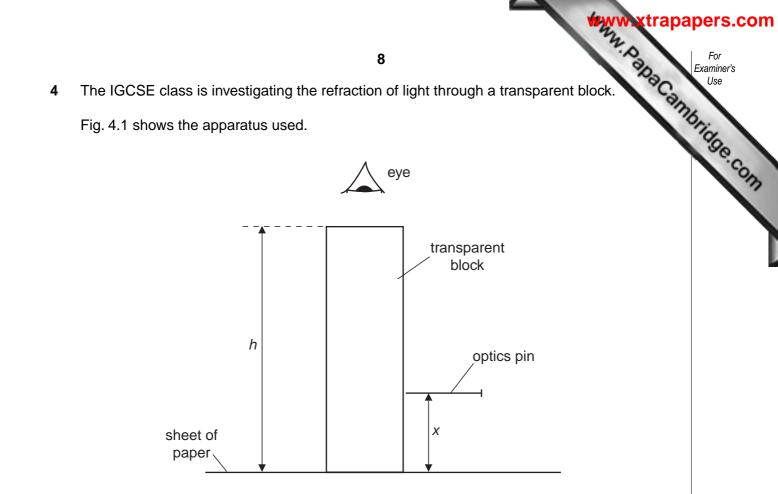


Fig. 4.1

- (a) A student looks down through the transparent block at the image of a line drawn on the sheet of paper. She carefully places the point of the optics pin exactly in line with the image.
 - (i) On Fig. 4.1, measure the vertical distance *x* between the paper and the pin.

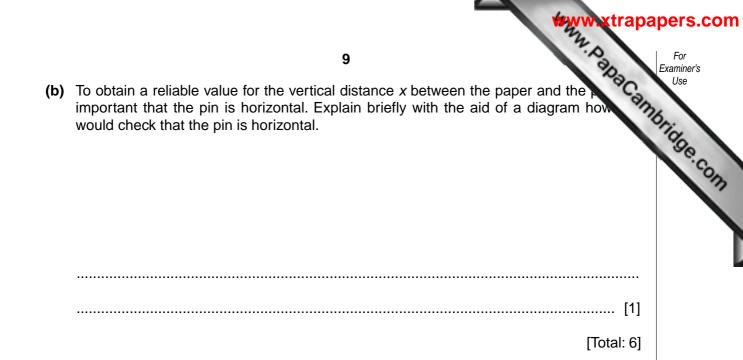
(ii) On Fig. 4.1, measure the height *h* of the transparent block.

h =

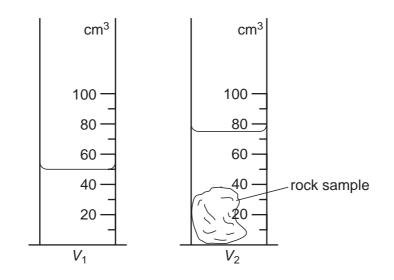
x =

(iii) Calculate the refractive index *n* of the material of the block using the equation

$$n=\frac{h}{h-x}.$$



www.papacambridge.com 5 (a) An IGCSE student is investigating the differences in density of small pieces of a rocks. She is using an electronic balance to measure the mass of each sample using the 'displacement method' to determine the volume of each sample. Fig. 5.1 show the displacement method.





Write down the volume shown in each measuring cylinder. (i)

> *V*₁ = V₂ =

Calculate the volume V of the rock sample. (ii)

V =

(iii) Calculate the density of sample A using the equation

density =
$$\frac{m}{V}$$
,

where the mass *m* of the sample of rock is 109 g.

density =

[4]

- (b) The table shows the readings that the student obtains for samples of rocks B Complete the table by
 - inserting the appropriate column headings with units, (i)
 - calculating the densities using the equation density = $\frac{m}{V}$. (ii)

	he table show omplete the ta		11 that the stude	ent obtains for	samples of rocks B	For Examiner's Use
(i) inserting the appropriate column headings with units, (ii) calculating the densities using the equation $density = \frac{m}{V}$.						Tide com
sample	<i>m</i> /g			V/	density/	
В	193	84	50	34		
С	130	93	50	43		

[4]

(c) Explain briefly how you would determine the density of sand grains.

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[Total: 9]



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12

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