



# Cambridge IGCSE™

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**PHYSICS**

**0625/42**

Paper 4 Extended Theory

**February/March 2022**

MARK SCHEME

Maximum Mark: 80

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**Published**

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the February/March 2022 series for most Cambridge IGCSE™, Cambridge International A and AS Level components and some Cambridge O Level components.

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This document consists of **17** printed pages.

**PUBLISHED****Generic Marking Principles**

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

**GENERIC MARKING PRINCIPLE 1:**

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

**GENERIC MARKING PRINCIPLE 2:**

Marks awarded are always **whole marks** (not half marks, or other fractions).

**GENERIC MARKING PRINCIPLE 3:**

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

**GENERIC MARKING PRINCIPLE 4:**

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

**GENERIC MARKING PRINCIPLE 5:**

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

**GENERIC MARKING PRINCIPLE 6:**

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

**Science-Specific Marking Principles**

1	Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
2	The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
3	Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
4	The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
5	<p><u>'List rule' guidance</u></p> <p>For questions that require <i>n</i> responses (e.g. State <b>two</b> reasons ...):</p> <ul style="list-style-type: none"><li>• The response should be read as continuous prose, even when numbered answer spaces are provided.</li><li>• Any response marked <i>ignore</i> in the mark scheme should not count towards <i>n</i>.</li><li>• Incorrect responses should not be awarded credit but will still count towards <i>n</i>.</li><li>• Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should <b>not</b> be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.</li><li>• Non-contradictory responses after the first <i>n</i> responses may be ignored even if they include incorrect science.</li></ul>

**6** Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g.  $a \times 10^n$ ) in which the convention of restricting the value of the coefficient ( $a$ ) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

**7** Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

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0625/42 Preamble

## NOTES ABOUT MARK SCHEME SYMBOLS AND OTHER MATTERS

B marks	are independent marks, which do not depend on other marks. For a B mark to be scored, the point to which it refers must be seen specifically in the candidate's answer.
M marks	are method marks upon which accuracy marks (A marks) later depend. For an M mark to be scored, the point to which it refers <b>must</b> be seen in a candidate's answer. If a candidate fails to score a particular M mark, then none of the dependent A marks can be scored.
C marks	are compensatory marks in general applicable to numerical questions. These can be scored even if the point to which they refer are not written down by the candidate, <b>provided subsequent working gives evidence that they must have known it</b> . For example, if an equation carries a C mark and the candidate does not write down the actual equation but does correct substitution or working which shows he knew the equation, then the C mark is scored. A C mark is not awarded if a candidate makes two points which contradict each other. Points which are wrong but irrelevant are ignored.
A marks	A marks are accuracy or answer marks which either depend on an M mark, or which are one of the ways which allow a C mark to be scored. A marks are commonly awarded for final answers to numerical questions. If a final numerical answer, eligible for A marks, is correct, with the correct unit and an acceptable number of significant figures, all the marks for that question are awarded. However, an A mark following an M mark is a dependent mark and is only awarded if the M mark has been awarded.
Brackets ( )	Brackets around words or units in the mark scheme are intended to indicate wording used to clarify the mark scheme, but the marks do not depend on seeing the words or units in brackets, e.g. 10 (J) means that the mark is scored for 10, regardless of the unit given. However, if a word in brackets is replaced with another word that is clearly wrong then the mark should not be awarded.
<u>Underlining</u>	Underlining indicates that the underlined word (or a synonym) <b>must</b> be seen in the answer offered. If the word is a technical scientific term the word must be there.
OR / or	This indicates alternative answers, any one of which is satisfactory for scoring the marks.
eeoo.	This means 'each error or omission'.
owtte.	This means "or words to that effect".
Ignore	This indicates that something which is not correct or irrelevant i.e. it is not a contradiction (CON) is to be disregarded and does not incur a penalty.

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Spelling	Be generous about spelling and use of English. If an answer can be understood to mean what we want, give credit. However, do not allow ambiguities, e.g. spelling which suggests confusion between reflection / refraction / diffraction or thermistor / transistor / transformer.
Not/NOT	This indicates that an incorrect answer is not to be disregarded, but cancels another otherwise correct alternative offered by the candidate, i.e. right plus wrong penalty applies.
ecf	meaning ‘error carried forward’ is mainly applicable to numerical questions, but may in particular circumstances be applied in non-numerical questions. This indicates that if a candidate has made an earlier mistake and has carried an incorrect value forward to subsequent stages of working, marks indicated by ecf may be awarded, provided the subsequent working is correct, bearing in mind the earlier mistake. This prevents a candidate from being penalised more than once for a particular mistake, but <b>only</b> applies to marks annotated ecf in the mark scheme. <b><u>Always annotate ecf if applied.</u></b>
cao	correct answer only
Significant Figures	Answers are normally acceptable to any number of significant figures $\geq 2$ . Any exceptions to this general rule will be specified in the mark scheme. Annotate with SF from the toolbar. A second (or further) sig. fig. error in a single question is not penalised; annotate with SF SF. It is normally acceptable to quote just 1 s.f. for answers, which are exact to 1 s.f
Units	Deduct one mark for each incorrect or missing unit from an answer that would otherwise gain all the marks available for that answer: <b>maximum 1 per question</b> . No deduction is incurred if the unit is missing from the final answer but is shown correctly in the working. Annotate with U. For more than one unit error in a question, annotate UU to indicate an error which has not been penalised. Unless listed here or stated in the mark scheme for the question, do not accept derived units e.g. $\text{kg m s}^{-2}$ for N is NOT acceptable. The following are acceptable alternatives: Nm for J, $\text{Js}^{-1}$ or $\text{Nms}^{-1}$ for W, $\text{Nm}^{-2}$ for Pa, Ns and $\text{kg m s}^{-1}$ are both acceptable for both momentum and impulse. Beware : J NOT acceptable for moments. Condone wrong use of upper and lower case symbols, e.g. pA for Pa.
Arithmetic errors	If the <b>only</b> error in arriving at a final answer is clearly an arithmetic one, then the mark awarded will be one mark lower than the maximum mark. Regard a power-of-ten error as an arithmetic error unless otherwise specified in the mark scheme. Annotate with POT. Do not penalise the same POT error more than once in each whole question. Annotate POT POT. However if the power-of-ten error is due to the wrong omission or inclusion of $g$ ( $= 10 \text{ N / kg}$ ) this rule does not apply. The use of a wrong SI prefix in the final answer is counted as a power-of-ten error rather than a unit error.

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Transcription errors	If the only error in arriving at a final answer is because previously calculated data has clearly been misread, but used correctly, then for that part question the mark will be one less than the maximum mark.
Fractions	Allow these only where specified in the mark scheme; they are a form of sig. fig. error; annotate with SF. Consequently, when a sig. fig. error and a fraction is used in the same question, the second answer may still be awarded full marks.
Crossed out	Work which has been crossed out <b>and not replaced but can easily be read</b> , should be marked as if it had not been crossed out. Look to see if it has been replaced on a blank page or another part of the same page.
Use of <b>NR</b>	(# or / key on the keyboard). Use this if the answer space for a question is completely blank or contains no readable words, figures or symbols.

**RM Assessor 3**

Please note that 0625 papers are now marked using RM assessor3. Videos and documents are available by using the Help icon in the top right hand corner when logged in or from the RM support site. Familiarisation mode is also available on RM Assessor 3. The tool bar is now located on the left of the screen and you drag items used frequently to the right hand side of the tool bar. Note – the tool bar won't be visible until you have scripts to mark rather than just browse.

**RM Assessor3 annotations:**

annotation	suggested use
tick	mark awarded (note the ticks are added up next to the tick annotation, check the total you enter agrees)
cross	no mark awarded
SEEN	indicates page seen
BOD	benefit of doubt given
NBOD	no benefit of doubt given
on page comment	gives a text box to write comment – much easier to use than in the previous version of RM assessor
ECF	error carried forward
^	omission mark
?	unclear
U UU	unit penalty applied unit penalty not applied because already applied earlier in same question

annotation	suggested use
wavy line (horizontal or vertical)	used to highlight a particular point
CON	contradiction
NAQ	not answered question
PD	poor diagram
SF SFSF	error in number of significant figures significant figure error not penalized.
POT POT POT	power-of-ten error POT penalty not applied as already applied
TV	too vague
I	ignore
SC	special case



**PUBLISHED****Guidance on the expression : words, symbols or numbers**

Accept numbers if, in the context of the question, that number can reasonably be a value for the quantity in the equation/formula.

Accept alternative symbols that can reasonably represent the quantity in the equation but not if there is confusion with another quantity that is represented by that symbol in the syllabus e.g. Q is often acceptable for thermal energy but not if, as has happened, if there is confusion with charge.

**Linking pages to other questions RM Assessor3**

Sometimes candidates answer questions outside the marking zone for a particular question. Examiners must ensure that they identify and mark the whole response for each item. This may mean

- Viewing the whole page so that you can read the entire answer and put ticks where marks are awarded
- Linking text to a question by using “view full response” and then “Link to Question”

**Blank Pages and Blank AOs**

Page 16 is blank. It is attached to Q1(a) and should be annotated “seen”.

Annotate with “seen” that you have noted any blank Additional Objects.

**Annotation**

To increase marking transparency, reduce the number of enquiries about results and assist team leaders, the following is mandatory :

- For **all** questions with two or more marks, tick to indicate where each mark is awarded (or near final answer if a calculation is all correct)
- For questions with one mark, examiners annotations are not mandatory.
- Annotations and comments must never suggest or imply that a mark has been deducted, e.g. –1

Question	Answer	Marks
1(a)	2.2 s	<b>B1</b>
1(b)	<p>Any two from:</p> <ul style="list-style-type: none"> <li>Line on graph is horizontal / gradient is zero</li> <li>(therefore) no acceleration / speed is constant</li> <li>(resultant) force causes / is proportional to acceleration</li> </ul>	<b>B2</b>
1(c)	8.5 ms <sup>-2</sup>	<b>A3</b>
	(a =) $\Delta v / t$ in any form OR gradient of graph OR 12.8 / 1.5 OR other suitable values from graph	<b>(C1)</b>
	(1.5, 12.8) both seen OR alternative suitable points on the line identified	<b>(C1)</b>
1(d)	$0.5 \times 12.8 \times 1.5 (= 9.56 / 9.6 \text{ m})$ OR $6.4 \times 1.5 (= 9.6)$	<b>A2</b>
	(length of ramp) = area under graph (between 0–1.5 s) OR <u>average</u> velocity $\times$ time	<b>(C1)</b>


Question	Answer	Marks
2(a)(i)	extension (of the spring) is (directly) proportional to the force / load (applied to the spring, up to the limit of proportionality)	<b>B1</b>
2(a)(ii)	$W=mg$ in any form OR force is (directly) proportional to mass	<b>B1</b>
2(a)(iii)	80 N	<b>B1</b>
2(b)	straight line through / from origin with positive gradient up to 175 N	<b>B1</b>
	smooth curve after 175 N with increasing positive gradient	<b>B1</b>
2(c)	$(80 \text{ N} \times 3.5 \text{ m} =) 280 \text{ J}$	<b>A2</b>
	$\Delta E = Fxd$ in any form OR $GPE = mgh$ in any form	<b>(C1)</b>

Question	Answer	Marks			
3(a)	<p><i>Any two from:</i></p> <ul style="list-style-type: none"> <li>(Amount of water in the pool decreases) as water evaporates / becomes water vapour / gas</li> <li>The (more) energetic molecules escape OR fast(er) molecules escape OR molecules with more (kinetic) energy escape</li> <li>From the <u>surface</u> of the water</li> </ul>	<b>B2</b>			
3(b)	<table border="1"> <tr> <td>lower temperatures / cold(er) day</td> <td>OR</td> <td>less windy weather</td> </tr> </table>	lower temperatures / cold(er) day	OR	less windy weather	<b>B1</b>
	lower temperatures / cold(er) day	OR	less windy weather		
<table border="1"> <tr> <td>produces smaller pool more slowly because rate of evaporation decreases with decreasing temperature</td> <td>OR</td> <td>produces smaller pool more slowly as a draught over surface removes water vapour enabling faster rate of evaporation</td> </tr> </table>	produces smaller pool more slowly because rate of evaporation decreases with decreasing temperature	OR	produces smaller pool more slowly as a draught over surface removes water vapour enabling faster rate of evaporation	<b>B1</b>	
produces smaller pool more slowly because rate of evaporation decreases with decreasing temperature	OR	produces smaller pool more slowly as a draught over surface removes water vapour enabling faster rate of evaporation			
3(c)	<p><i>Any three from:</i></p> <ul style="list-style-type: none"> <li>(thermal) energy in the skin / body transferred to (molecules of) sweat</li> <li>These molecules (have enough KE to) escape from the skin / become water vapour</li> <li>Leaving behind molecules with lower energy</li> <li>Which leaves the skin / body at a lower temperature</li> </ul>	<b>B3</b>			

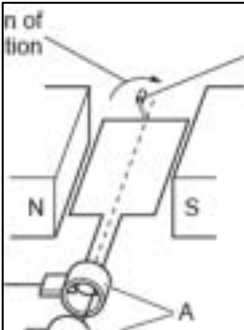
Question	Answer	Marks
4(a)(i)	(mass = $1900 \times 0.05$ ) = 95 kg	<b>A2</b>
	(m =) $\rho V$ in any form OR $1900 \times 0.05$	<b>(C1)</b>
4(a)(ii)	(= $95 \times 1500$ ) = 140 000 J / °C / $1.4 \times 10^5$ J / °C	<b>A2</b>
	(C =) $m \times c$	<b>(C1)</b>

Question	Answer	Marks
4(a)(iii)	34 000 s / 560 min / 9 h 20 min	<b>A3</b>
	Temperature rise = $12(^{\circ}\text{C}) / 19-7$ OR $E = C \times \Delta \theta$ in any form	<b>(C1)</b>
	$(t =) E / P$ in any form OR (thermal capacity $\times 12) / 50$	<b>(C1)</b>
4(b)(i)	sand molecules gain KE OR vibrate more OR hit (other) molecules (when heated) OR (thermal energy is transferred by) conduction	<b>B1</b>
	Energy is transferred to molecules of plastic pot in contact with sand (through collisions) OR Energy OR (lattice) vibrations transferred to neighbouring molecules	<b>B1</b>
4(b)(ii)	(sand is warmer than surroundings and so thermal) energy (constantly) is lost from the sand	<b>B1</b>
	(at a constant temperature) rate of (thermal) energy supplied to the sand is equal to rate of (thermal) energy lost from sand	<b>B1</b>

Question	Answer	Marks
5(a)(i)	straight line from clock to mirror AND from mirror to eye with correct arrows	<b>B1</b>
	Angle of incidence = angle of reflection	<b>B1</b>
5(a)(ii)	Correct position of image	<b>B1</b>
5(a)(iii)	Virtual (no mark) And Cannot be projected on a screen / light doesn't pass through image / AW	<b>B1</b>

Question	Answer	Marks
5(a)(iv)		<b>B1</b>
5(b)(i)	(monochromatic light) is light of a single frequency	<b>B1</b>
5(b)(ii)	$5.4 \times 10^{14}$ Hz	<b>A3</b>
	(speed of light =) $3 \times 10^8$ (m / s)	<b>(C1)</b>
	( $f = $ ) $v / \lambda$ in any form OR $3.0 \times 10^8 / 5.6 \times 10^{-7}$	<b>(C1)</b>

Question	Answer	Marks
6(a)	Minimum of one arrow on a field line pointing N → S and not contradicted	<b>B1</b>
	central line perpendicular to poles of magnets AND at least two other correct field lines	<b>B1</b>
6(b)(i)	(A are) slip rings	<b>B1</b>

Question	Answer	Marks
6(b)(ii)	 <p>Minimum of one arrow, pointing clockwise, on the wire, not contradicted</p>	<b>B1</b>
	Field direction, motion of wire and induced current are mutually perpendicular OWTTE	<b>B1</b>
6(b)(iii)	(as coil rotates) it cuts (magnetic) field between the magnets	<b>B1</b>
	This <u>induces</u> an e.m.f. / voltage / p.d. (in the coil)	<b>B1</b>
	This produces a current in the (coil transferred to the) galvanometer (via the slip rings and carbon brushes)	<b>B1</b>
	Direction of current flow changes with each 180 degree rotation of coil	<b>B1</b>

Question	Answer	Marks
7(a)	8 (cells)	<b>B1</b>
7(b)(i)	$(12 / 2.4 =) 5.0 \Omega$	<b>A2</b>
	$R = V / I$ in any form	<b>(C1)</b>
7(b)(ii)	(Resistance of Q = $5 - 1.5$ ) = $3.5 \Omega$	<b>A2</b>
	$1 / R = 1 / R_1 + 1 / R_2$ OR $R = R_1 R_2 / (R_1 + R_2)$	<b>(C1)</b>

Question	Answer	Marks
7(c)(i)	correct voltmeter symbol connected correctly across both lamps	<b>B1</b>
7(c)(ii)	4.3 W	<b>A3</b>
	(P = ) IV in any form OR $1.2 \times 3.6$	<b>(C1)</b>
	3.6 V <b>OR</b> 1.2 A	<b>(C1)</b>

Question	Answer	Marks
8(a)	Labelled diagram showing <ul style="list-style-type: none"> <li>labelled <u>Iron</u> (core)</li> <li>two coils, labelled primary and secondary, (around core and connected to separate circuits)</li> <li>Fewer turns on labelled secondary coil</li> </ul>	<b>B3</b>
8(b)	(Primary voltage causes) an <u>alternating</u> current in primary coil	<b>B1</b>
	produces a <u>changing</u> magnetic field	<b>B1</b>
	(changing) field <u>induces</u> pd / e.m.f. (in secondary coil)	<b>B1</b>
8(c)	0.026	<b>A2</b>
	(Ns / Np =>) Vs / Vp in any form	<b>(C1)</b>

Question	Answer	Marks																				
9(a)(i)	(A signal that has one of) two possible states	<b>B1</b>																				
9(a)(ii)	<table border="1" style="display: inline-table; vertical-align: top;"> <thead> <tr> <th>A</th> <th>B</th> <th>C</th> <th>D</th> </tr> </thead> <tbody> <tr> <td>0</td> <td>0</td> <td>0</td> <td>1</td> </tr> <tr> <td>0</td> <td>1</td> <td>0</td> <td>1</td> </tr> <tr> <td>1</td> <td>0</td> <td>0</td> <td>1</td> </tr> <tr> <td>1</td> <td>1</td> <td>1</td> <td>0</td> </tr> </tbody> </table> <p>C all correct ; D all correct ;</p>	A	B	C	D	0	0	0	1	0	1	0	1	1	0	0	1	1	1	1	0	<b>B2</b>
A	B	C	D																			
0	0	0	1																			
0	1	0	1																			
1	0	0	1																			
1	1	1	0																			
9(b)	NAND	<b>B1</b>																				

Question	Answer	Marks
10(a)	${}_{95}^{241}\text{Am} \rightarrow {}_{93}^{237}\text{Np} + {}_2^4\alpha$	
	${}^{237}\text{Np}$ nucleon number correct for Np	<b>B1</b>
	${}_{93}\text{Np}$ proton number correct for Np	<b>B1</b>
	$+ {}_2^4\alpha$ alpha notation correct	<b>B1</b>
10(b)(i)	alpha (particles emitted from americium)	<b>B1</b>
	move close to / hit molecules in the air (between the metal plates)	<b>B1</b>
	removing electrons (out of the molecules)	<b>B1</b>



Question	Answer	Marks
10(b)(ii)	<i>Any two from:</i> <ul style="list-style-type: none"><li>• alpha not penetrating / short range AND alpha (particles) stopped by smoke particles</li><li>• alpha (particles) more highly ionising (than gamma) AND ionise air more easily</li><li>• range of alpha particles is short / alpha is not penetrating AND alpha less harmful (to humans)</li></ul>	<b>B2</b>