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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2010 question paper for the guidance of teachers

0653 COMBINED SCIENCE

0653/22

Paper 2 (Core Theory), maximum raw mark 80

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Mark schemes must be read in conjunction with the question papers and the report on the examination.

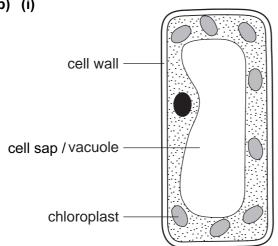
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Page 2		Syllabus
	IGCSE – October/November 2010	0653
(a) 1. 2.	traps (layer of) air; acts as insulator/reduces conduction/reduces convectors	Syllabus 0653 ction ;
(b) (i)	increased risk of skin cancer/eye damage/sun burn;	[1]
(ii)	1. wave; 2. use;	[2]
		[Total: 5]
(a) (i)	lead oxide + carbon \rightarrow lead + carbon dioxide LHS ; RHS ;	[2]
(ii)		
(b) (i)	(dc) power supply/battery/cell;	[1]
(ii)	 chlorine; anode, non-metals form at the anode/chlorine negative (and anode is positive); 	is a non-metal/chloride ions are [2]
		[Total: 8]
(a) (i)	transpiration / evapotranspiration / diffusion;	[1]
(ii)	stomata;	[1]
(iii)	 condensation; water vapour cooled / temperature fell; gas changed to, liquid / water; ref. to particles and (kinetic) energy; 	[max 2]
	(a) 1. 2. (b) (i) (ii) (ii) (ii) (ii)	 (a) 1. traps (layer of) air; 2. acts as insulator / reduces conduction / reduces convection. (b) (i) increased risk of skin cancer / eye damage / sun burn; (ii) 1. wave; 2. use; (a) (i) lead oxide + carbon → lead + carbon dioxide LHS; RHS; (ii) 1. lead oxide / carbon dioxide; ecf 2. compounds contain more than one type of elements. 3. reference to (different) elements / atoms in composition. (b) (i) (dc) power supply / battery / cell; (ii) 1. chlorine; 2. anode, non-metals form at the anode / chlorine negative (and anode is positive); (ii) transpiration / evapotranspiration / diffusion; (iii) stomata; (iii) 1. condensation; 2. water vapour cooled / temperature fell; 3. gas changed to, liquid / water;

Page 3	Mark Scheme: Teachers' version	Syllabus
	IGCSE – October/November 2010	0653
(b) (i)	cell wall	Cambridge.com



[max 2]

(ii) palisade (mesophyll);

1

[2]

- (iii) 1. carbon dioxide + water;
 - 2. glucose / carbohydrate / starch / sugar + oxygen;

[Total: 9]

- (a) 28(s); [1]
 - **(b) (i)** 1. distance = speed × time; $= 10 \times 60 \times 9 = 5400 \,\mathrm{m}$; [2]
 - (ii) 1. work done = force × distance; $= 10\ 000 \times 5400 = 54\ 000\ 000\ J$; [2]

[Total: 5]

- (a) idea of restoring full number of chromosomes in the zygote; 5 [1]
 - [1] **(b) (i)** ovary;
 - (ii) oviduct / Fallopian tube; [1]
 - (c) 1. produces / contains, (amniotic) fluid;
 - protects/supports, embryo; [2]
 - (d) 1. idea that mother's body needs to make substances for both herself and the fetus;
 - iron for haemoglobin;
 - calcium for, bones / teeth; [3]

[Total: 8]

		2.
Page 4	Mark Scheme: Teachers' version	Syllabus r
	IGCSE – October/November 2010	0653
		S

- 6 (a) water conducts electricity;
 - (b) all symbols in correct circuit ;;; all four correct for 3 marks any three correct for 2 marks all two correct for 1 mark

[3]

(c) (i) K and L;

[1]

(ii) 1. J lights up;2. K & L go off;

[2]

(d) add one 12 Ω to one 8 Ω (in series);

[1]

(e) B F C D E A

B F;

DE;

[2]

[Total: 10]

7 (a) (i) O and S;

[1]

(ii)

element name	protons	neutrons
(oxygen)	8	8
phosphorus	(15)	(16)

one mark for each row;;

[2]

(b) (i) copper oxide / copper carbonate / other correct;

[1]

(ii) magnesium; (allow aluminium) [1]

(c) (i) reaction 1

combustion / oxidation;

reaction 2

polymerisation;

[2]

(ii) (molecules) join together / form chains;

[1]

[Total: 8]

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Page 5	Mark Scheme: Teachers' version	Syllabus	
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- **8** (a) (i) 2000 (kg per hectare);
 - (ii) any two values with a range between 6 and 7.25;
 - (iii) 1. calcium carbonate is a base;
 - 2. neutralise (acid in the soil);
 - 3. raise pH;
 - 4. above 5.5 / closer to 6.0;

[2 max]

- (b) 1. terracing/walls (qualified);
 - 2. bunds/embankment/ditch;
 - 3. plough along slope (not up and down);
 - 4. keep crop cover (at all times)/plant trees;
 - 5. other valid points;;

[2 max]

(c) (i) insects; allow self-pollination

[1]

- (ii) 1. pollen contains male gamete;
 - 2. fertilisation must take place;
 - 3. male gamete (in pollen grain) fuses with female gamete;
 - 4. seed develops from (fertilised) ovule;

[2 max]

(iii) 1. biuret test/add biuret reagent/add copper sulfate and KOH solution;

purple / lilac / mauve ;

[2]

[Total: 11]

9 (a)

	description	charge	range in air	ionising ability
alpha	helium nucleus	positive	5 cm	very strong
beta	electron	negative	50 cm	medium
gamma	wave	none	many kilometres	weak

;;;; [4]

(b) alpha particles have low penetration in air/ will not reach people living in house/ smoke detectors are a long way from people;

[1]

[Total: 5]

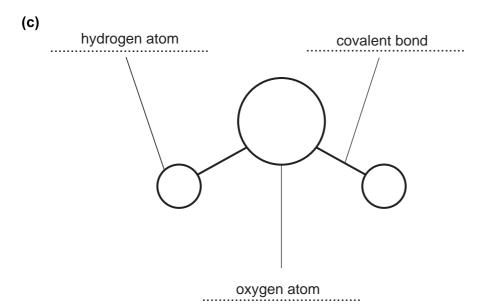
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Page 6	Mark Scheme: Teachers' version	Syllabus	.0	r
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- 10 (a) 1. adding chlorine;
 - 2. kills (harmful) bacteria / microorganisms / germs;
 - 3. filtration;
 - 4. removes solid / insoluble materials / dirt;
 - (b) (i) 1. dissolves in/reacts with rain water;
 - 2. produces acidic solution / acid rain;
 - 3. ref. to sulfurous / sulfuric acid;
 - 4. acidic rain collects in rivers / lakes;
 - 5. reference to harmful effects of acidity, e.g. kills organisms;
 - (ii) removal of sulfur compounds from fuel / removal of sulfur dioxide from waste gases / reduce demand for energy / burn less fuel;

max 1

max 3



3

[Total: 11]