UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

WANN, PapaCambridge.com MARK SCHEME for the October/November 2010 question paper

for the guidance of teachers

0653 COMBINED SCIENCE

0653/62

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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Page 2	Mark Scheme: Teachers' version Syllabus	2
	IGCSE – October/November 2010 0653	Day
(a) (i) 5.4 g		Vaha Cambridg
5.(0)	, g ;	Dri
		8
(ii) tube tube	1 0.2g; 2 0.3g;	
tube	3 <u>1.0 g</u> ;	
tube	4 0.8 g ; (1 mark each, (ecf))	[4]
(b) pineapple	· (allow cof)	
	ost greatest mass ;	[2]
	eighed) protein with acid (instead of juice) ; loss in / change of mass after 10 mins ;	[2]
	 /	
		[Total: 10]
(a) (i) corre	ct symbols for ammeter and lamp shown in circuit ;;	[2]
(ii) it is n	netallic / metal ;	[1]
(b) any ment	on of use of a magnet ;	[1]
(c) (i) heat	the mixture ;	
	am or mention of suitable apparatus, e.g. test-tube or metal container ;	[2]
(ii) heat	gives energy (so that atoms react) ;	[1]
(iii) exoth	ermic ;	[1]
	roperty mentioned ;	[0]
	n iron sulfide ; netic + non-magnetic/melting point + high mpt/electrical conductivity uctor)	+
		[Total: 10]

	3 Mark Scheme: Teachers' version Syllabus IGCSE – October/November 2010 0653	×	
	IGCSE – October/November 2010 0653	C.	
(a) (i)	8.6 cm (+/– 0.1 cm) ;	mbr.	
(ii)	3Mark Scheme: Teachers' versionSyllabus3IGCSE – October/November 20100653 $8.6 \text{ cm} (+/- 0.1 \text{ cm});$ $6.2 \text{ cm} (+/- 0.1 \text{ cm});$	3	
(iii)	8.6/6.2 = 1.4 (1.39) (no penalty for using more decimal points) (ecf);		
(b) (i)	$r_3 = 49$ degrees (+/- 2 degrees) ; $r_4 = 76$ degrees ;	[2]	
(ii)	sine $r_3 = 0.75$ / sine $r_4 = 0.97$ (ecf) (one or both correct) ;	[1]	
(iii)	both points correct (+/- half square) and straight line drawn through the origin ;	[1]	
(iv)	<i>x</i> - and <i>y</i> - distances used marked on the graph ; gradient = 1.5 (ecf) ;	[2]	
ì it i	Ilue (b)(iv) is more accurate) is derived from several values instead of just one/owtte/very difficult to easure through glass block ;	[1]	
	Т	otal: 10]	
(a) (i)	[T still air 1.8 cm ; windy air 14.7 cm ;	otal: 10] [2]	
(a) (i) (ii)	still air 1.8 cm ; windy air 14.7 cm ;	-	
(ii)	still air 1.8 cm ; windy air 14.7 cm ; 1.4 cm ;	[2]	
(ii) (iii) (b) mo (gra	still air 1.8 cm ; windy air 14.7 cm ; 1.4 cm ; 14.4 cm ; 1.4 / 4 = 0.35 ; (ecf)	[2]	
(ii) (iii) (b) mo (gra	<pre>still air 1.8 cm; windy air 14.7 cm; 1.4 cm; 14.4 cm; 1.4/4 = 0.35; (ecf) 14.4/4 = 3.6; (ecf)</pre>	[2] [2] [2]	
(ii) (iii) (b) mo (gr. <u>eva</u> (c) (i)	<pre>still air 1.8 cm; windy air 14.7 cm; 1.4 cm; 1.4 cm; 1.4/4 = 0.35; (ecf) 14.4/4 = 3.6; (ecf)</pre>	[2] [2] [2]	

Page 4	Mark Scheme: Teachers' version	Syllabus r
-	IGCSE – October/November 2010	0653
	 no change / no reaction / no bubbles / dissolves no change / no reaction / no bubbles / dissolves 	
	 B sodium chloride or hydrochloric acid C nitric acid or potassium nitrate 	[2]
solution solution (solution	A is nitric acid B is sodium chloride C is potassium nitrate D is hydrochloric acid ;;; rect 3 marks, 3 correct 2 marks, 2 correct 1 mark)	[3]
test gas litmus tur or carry	um hydroxide solution and aluminium foil and warm evolved using red litmus or by smell ; rns blue / ammonia is given off ; out flame test ;	
lilac flam	e seen ; (for a max of 2 marks)	[3]

Page 5		Syllabus Syllabus
	IGCSE – October/November 2010	0653
(a) any	dimensions to give an area of $5 \text{ cm}^2 \text{ e.g. } 5 \text{ cm} \times 1 \text{ cm}$;	ennth
(b) 0.75	5A, 0.90A (second decimal point must be shown) ;	Syllabus 0653 0654 0654 0654 0654 0654 0654 0654 0000 0000
• • •	increases the resistance so that) the current is decrease resistor/owtte ;	
• •	points plotted +/– half square ; ight line drawn ;	[2
(e) the I	hook / pan has a mass / owtte ;	['
	iron loses its magnetism when the current is switched o steel does not / owtte / steel retains its magnetism ;	off ; [2
	ent could leak from the wire (through the iron)/owtte/p ck if touched ;	revent short circuit / no [´
		[Total: 10