UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

## WANN, PapaCambridge.com MARK SCHEME for the October/November 2011 question paper

## for the guidance of teachers

## 0653 COMBINED SCIENCE

0653/22

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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Page	2	Mark Scheme: Teachers' version Syllabus	2
		IGCSE – October/November 2011 0653	Da
(a) (i)	cart	oon dioxide ;	Alba Cambrid
(ii)	) HC	!;	19
(iii)		ction has stopped ; I has been used up/owtte ;	[2]
(iv)	) calo	sium ;	[1]
(b) (i)	mal	oon dioxide reacts with (sea)water ; kes water more acidic / less alkaline / pH decreases ; -metal oxides are acidic ;	[max 2]
(ii)	e.g. mor	ept any reasonable science based idea: calcium carbonate may react with more acidic water/lower pH makes i re difficult for coral to extract ions from sea/coral organism does no vive in more acidic water/enzymes (in coral) denatured/ref. to globa	ot
		ming ;	[max 1]
			[Total: 8]
(a) (i)	) air r	resistance / friction / drag ;	[1]
(ii)	) equ	al and opposite / cancel each other out ;	[1]
(iii)		stant speed ; ow constant velocity)	[1]
		e = speed × time ; 6600 = 288 000 m ;	[2]
(c) (i)	cau	ations / damage DNA ; se cancer ; cells ;	
	radi	ation burns ; ation sickness ;	[max 2]
(ii)	) (gra	anite) rocks ;	[1]
(d) na		iata una .	CI
aþ	propr	iate use ;	[2]
			[Total: 10

Page 3	Mark Scheme: Teachers' version	Syllabus
	IGCSE – October/November 2011	0653 23
(a) gluco wate	ose ; r + carbon dioxide ;	Syllabus 0653 Syllabus 0653
ref. to	e blood / in an artery / in a capillary ; o haemoglobin ; <u>d</u> blood cells ;	[max 2]
(c) (i) (	D.4 dm <sup>3</sup> ;	[1]
r	(assume answer refers to fast run unless otherwise state nore (oxygen used per minute) ; ncreases more rapidly ; ).9 dm <sup>3</sup> more ;	d) [max 2]
<u>r</u>	nore <u>energy</u> used when running faster ; <u>muscles</u> working harder ; herefore more <u>respiration</u> ;	[max 2]
(d) breal	kdown of walls of alveoli / reduction of surface area ;	[1]
		[Total: 10]
(a) (i) 🤅	switches 1 and 2/both ;	[1]
	voltmeter in parallel and ammeter in series ; everything else unchanged ;	[2]
(b) (i) (	coal / oil / gas ;	[1]
<b>(ii)</b> t	o reduce energy losses ;	[1]
<b>(</b> iii <b>)</b> (	(5000/400000 = 10000/Ns, so Ns =) 800000 (turns) ;	[1]
(iv) \	voltage needs to be lower ;	
• •	or safety ;	[2]

Pa	ige 4		Mark Scheme: Teachers' version IGCSE – October/November 2011		<u> </u>
(a)	.,		ŀ	Syllabus 0653	anbrid
(b)		ovary ;			[2]
	stat	ement	asexual reproduction	sexual reproduction	]
	gam	netes are involved	×	$\checkmark$	1
	new	individuals are produced	✓	~	1
	a zy	gote is produced	×	$\checkmark$	
	offs	pring are genetically identical	$\checkmark$	×	
(a)	(i)	not allow for anything where it i 89 (%);	is not clear whether it is a t		[3 otal: 7 [1
		metals are <u>melted</u> together ;			[1
	. ,	iron ;			[1
	(iv)	unreactive ; strong / hard / not easily bent of malleable ;	r deformed ;	[1	max 2
(b)	(i)	tin oxide + carbon $\rightarrow$ tin + carb	oon monoxide ;		[1
	(ii)	carbon ; gains / bonds with oxygen ;			[2
(c)	(i)	negative electrode ; compound in liquid form/soluti	ion/molton:		

- compound in liquid form / solution / molten ; which conducts a current / contains free ions ;
  (ii) group number = outer electrons / Al is in Group 3 ;
- [Total: 12]

[3]

[1]

Page 5	Mark Scheme: Teachers' version	Syllabus
	IGCSE – October/November 2011	0653
(a) work = 1	force × distance ;	22
	2000 = 200 000 J ;	01
		3
<b>(b) (i)</b> kine	etic / movement ;	Syllabus 0653 (1) (1)
(ii) hea	at/sound;	[1]
(iii) suri	roundings ;	[1]
<b>(c) (i)</b> 401	<ol> <li>(0)</li> </ol>	[1]
		[']
	ume = mass / density ; 0/1020 = 0.04 m <sup>3</sup> ;	[2]
	0/1020 - 0.04m ,	
		[Total: 8]
<b>(a) (i)</b> dige	estion ;	[1]
	nutrients / molecules, can be <u>absorbed</u> ;	
(11) 50,	nutrients / molecules, can be <u>absorbed</u> ,	[1]
	teins ;	
	/gen ; natured ;	[3]
(b) (i) the	number of different, species / types of organisms ;	[1]
(ii) affe	ect, food chains / food webs ;	
	dators of frogs may reduce in numbers ;	
Inst	ects / prey of frogs, may increase in numbers ;	[max 2]
		[Total: 8]

Page 6		Syllabus 7
	IGCSE – October/November 2011	0653 23
(a) (i)	formed as fossil fuel / decay of organic matter / digestive ruminants / vulcanism ;	Syllabus 0653 e system of
(ii)	H only other symbol ; H × 4 bonded to central C with all single bonds ;	[2
(iii)	(carbon dioxide) global warming / (runaway) greenhouse effect ; detail of mechanism e.g. reflects heat back to Earth ; causing climate change / or example of ;	[max 2
	(carbon monoxide) toxic (to humans) ;	[max 1
(b) (i)	fractional distillation / fractionation ;	[1
(ii)	the greater the molecular mass ; the higher the boiling point ; use of the data e.g. $C_{12}H_{26}$ most massive and has higher	est boiling point ; [max 2
		[Total: 9