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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2011 question paper for the guidance of teachers

0653 COMBINED SCIENCE

0653/32

Paper 3 (Extended Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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Syllabus 0653

				Con 1
1	(a)	(i)	carbon dioxide;	Brig
		(ii)	(there is not enough evidence) result shows that:	Cambridge
			carbonate present ; but not calcium/need to show it is calcium carbonate ;	[2]
	(b)	(i)	carbon dioxide dissolves in/reacts with (sea)water/rain; makes water more acidic/less alkaline;	
			non-metal oxides are acidic ;	[max 2]
		(ii)	accept any reasonable attempt at a scientific answer: e.g. calcium carbonate may react with more acidic water/lower pH makes it	
			more difficult for coral to extract ions from sea/coral (polyps) does not survive in more acidic water;	[1]
				[Total: 6]
				[10141110]
2	(a)		cose + oxygen → carbon dioxide + water ; marks for all correct, one mark if any mistake)	[2]
		(2 1	marks for all correct, one mark if any mistake	[2]
	(b)			
		combined with haemoglobin/as oxyhaemoglobin; in red blood cells;		[max 2]
	(c)	(i)	evaporation ; (evaporation) takes heat from body ;	[2]
		(ii)	(assume answer refers to not drinking fluid unless otherwise stated)	
			rose higher ; rose faster ;	
			use of comparative figures, e.g. 40.0 °C and 38.7 °C;	[max 2]
		(iii)	less sweat produced when no fluids drunk/or reverse argument; to maintain water content of body/ref. to homeostasis;	[2]
				[Total: 10]
3	(a)	-	os layer of air ; · is a good) insulator ;	[2]
		(411	33/	[-]
	(b) does not deplete fossil fuel reserves/non-renewable;			
			a that dung is carbon neutral/renewable ; osene is a hydrocarbon fuel ;	[max 2]

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[Total: 9]

Page 3		3	Mark Scheme: Teachers' version	Syllabus			
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	(c) (i)	Syllabus 7. Day 7. O. Day					
	(,	from	ation passes ; n particle to particle ; rence to rarefaction and compression/diagram ;				
			ries of (compressions and rarefactions)/diagram;	[max 2]			
				[Total: 7]			
4	(a) irc	on ;		[1]			
	(b) (i)	SnO	$0_2 + 2C \rightarrow Sn + 2CO$;; (symbols and balanced)	[2]			
	(ii)		ninium more reactive than carbon ; ess reactive than carbon ;				
		Alm	nore strongly bonded to oxygen; w max 1 for the simple statement: aluminium is mor	[max 2] re reactive)			
	(iii)	alum alum	rence to use of carbon electrodes; ninium oxide is melted/dissolved in cryolite; ninium ions are positive/are cations; attracted move to negative electrode/cathode;				
			gain electrons from/are discharged at negative ele	ctrode; [max 3]			
	(c) (i)		- 56 + 32 × 2/184 ; w 183.5)	[1]			
	(ii)	7.80	$0 \times 0.89 = 6.9(42)g$ (unit required);	[1]			
				[Total: 10]			
5	(a) (i)		stigma ; anther/stamen ;	[2]			
	(ii)	stan	ma, feathery/outside flower/large surface area; nen, dangling/outside flower; netals; (allow small petals)	[max 2]			
	(iii)	invo	tume answer refers to sexual reproducation unless of lives gametes; lives fertilisation;	otherwise stated)			
		zygo	ote produced; oring genetically different/not clones;	[max 2]			
(b) acid rain; caused when nitrogen oxides, react with/dissolve in, (rain) water;							
damages plants ; damages aquatic animals ;							

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[1]

[2]

(a) grou	IGCSE – October/November 2011	0653			
(a) arou					
	up of cells ; rying out a particular or specific function/are similar cells	Syllabus 170 August 17			
(b) (i)	proteins; amino acids;	[2			
(ii)	diffusion; ref to concentration gradient/from high concentration to	o low concentration ; [2			
(c) (assume answer refers to animal cells unless otherwise stated) look at cells (as opposed to whole organism); no cell walls;					
	no large vacuoles ; no chloroplasts ;				
		[Total: 8			
(a) (i)	switch 1 and switch 2;	[
(ii)	voltmeter in parallel and ammeter in series ;	[
(b) (i)	to reduce energy losses;	[1			
(ii)	$Vp \div Vs = Np \div Ns \; ; \\ 5000 \div 400000 = 10000 \div Ns / (Ns=) 800000 \; (turns) \; ; \\ (1 \; mark \; for \; formula \; and \; 1 \; mark \; for \; substitution \; and \; ans \; (turns) \; ; \\ (1 \; mark \; for \; formula \; for \; for \; for \; for \; formula \; for \; for \; for \; for \;$	wer)			
(iii)	ref. to alternating or changing voltage or current; (in primary coil) produces alternating or changing magnetic field in coinduces (alternating) voltage in secondary coil;				
	idea that size of voltage change depends on (ratio of) to	urns; [max 3			
		[Total: 8			

(a) (i) formed as fossil fuel / decomposition of organic matter / from digestive

systems of ruminants / sources related to volcanism;

four covalent bonds means four pairs of electrons; (correct bonding diagram alone gains both marks)

(ii) 8;

Syllabus

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(b)	(i)	fractional distillation/fractionation;	and	
	(ii)	the larger/heavier/greater surface area of/greater number molecules; the higher the boiling point; OR unsaturation; lowers boiling point (for similar molecular size);	r of atoms in	
	(iii)	(shake liquid with) bromine/potassium manganate(VII); mixture goes colourless if liquid is D ; because D is unsaturated; (or reverse argument for A)	[3] [Total: 9]	
9 (a)	(a) (force =) mass × acceleration; acceleration = 1200000/400000; = 3 m/s ² ;			
(b)	(i)	to stop potato snacks oxidizing/reacting;	[1]	
	(ii)	pressure inside packet is greater than airplane pressure;	[1]	
(c)	(i)	speed has magnitude only/velocity has magnitude and dir	ection; [1]	
	(ii)	A to B / C to D;	[1]	
	(iii)	(no) not a straight line;	[1]	
	(iv)	C ;	[1]	
	(v)	50 m/s;	[1]	
	(vi)	the faster the skydiver travels the greater the air resistance eventually the air resistance balances the gravitational force		
((vii)	parachute increases air resistance;	[1]	
			[Total: 13]	

Mark Scheme: Teachers' version

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