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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2012 question paper for the guidance of teachers

0653 COMBINED SCIENCE

0653/21

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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1 (a) (speed =) distance / time; = 25 / 2 = 12.5 (km / h);

(b) (i) chemical;

(ii) heating engine / heating surroundings / light / sound;

[1]

(c) metal track expands in summer / hot weather; metal can expand into gap;

prevents damage to tracks;

[max 2]

[1]

(d) ethanol is renewable source / no sulfur dioxide produced;

(e) (i) 5 (km/h); (accept 4 / reference to greater than 3 but equal to and less than 5) [1]

(ii) 1200 (W); [1]

(iii) wind speed variable / wind unreliable / owtte; [1]

[Total: 10]

2 (a) (i) nucleus; [1]

(ii) 18; evidence of neutrons = nucleon number minus proton number; [2]

(iii) hydrogen; [1]

(b) (i) any typical metal property for X and corresponding non-metal for Y;
 e.g. X conductor Y insulator
 X malleable Y not malleable

(ii) Y is reactive / specific example, Z is unreactive; [1]

- (c) (i) reference to oxygen (not air); which joins / reacts with carbon; [2]
 - (ii) soil (too) acidic; lime, reacts with / neutralises acid / reduces acidity; to increase fertility;

[Total: 10]

[max 2]

	Do	MA 2	1	Mark Scheme: Teachers' version	Syllabus	
	Pa	ige 3)	IGCSE – May/June 2012	0653	8
3	(a)	onl	y eats	al that eats plants ; s plants / does not eat meat ; e to getting energy from, its food / plants ;	3333	DaCambridge.
	(b)	eat		; e in, more energy than they use; carbohydrate / protein, converted to fat;		[max 2]
	(c)	(i)	idea mas	greater the body mass, the greater the chance of sur that effect is greater at lower body masses / levels of ses; of figures;		[max 2]
		(ii)	insul	lator / poor conductor / reduces conduction ;		[1]
	(d)		bon d thane	lioxide ;		[2]
	(e)	(i)	(mea	an) body mass is increasing ;		[1]
		(ii)	weig as s	mots have more time to feed (from spring onwards) and during hibernation (as winters are shorter); pring arrives earlier, plants grow faster and as marmar have more plants to eat, therefore increasing mass	nots are herbivores	[max 1] [Total: 11]
4	(a)	inse me	ert bu asure	nesium to acid ; ng ; time for known volume of gas to collect / measure v rvals / measure how long reaction lasts ;	volume of gas at fixed	[3]
	(b)	(i)	temp	perature / surface area of magnesium ;		[1]
		(ii)		for (same volume) of gas to collect is shorter; ause rate of reaction is greater;		[2]
	(c)	(i)	aton	s charged, atom is neutral / proton and electron numn, but differ in the ion / the atom is reactive, the ion is tron shells full, while atom outer shell is not full;		[1]
		(ii)	MgC	Cl_2 ;		[1]
		(iii)	hydr	rogen;		[1]
						[Total: 9]

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- **5** (a) (i) between 10 and 20 (Hz) to between 20 000 and 25 000 (Hz);
 - (ii) frequency number of waves produced / passing a point per second;wavelength distance between consecutive, peaks / troughs;
 - (b) (i) needs, air / particles / a medium; [1]
 - (ii) quieter / lower volume ; [1]
 - (iii) microwaves; [1]
 - (c) angle of incidence labelled; angle of refraction labelled; [2]

[Total: 8]

- 6 (a) label to root hair cell; [1]
 - (b) water; minerals / ions / named mineral ion; (allow two different mineral ions for 2 marks) [2]
 - (c) (i) xylem; [1]
 - (ii) A in central area of root; [1]

(d) (i)

structure	plant cells	animal cells
cell membrane	✓	✓
cell wall	✓	×
nucleus	✓	✓
vacuole containing sap	✓	×

1 mark for any row correct ;;;; [4]

(ii) no, as they do not get light / as they are underground; [1]

[Total: 10]

[Total: 6]

Page 5	Mark Scheme: Teachers' version	Syllabus
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7 (a) (i) ammeter in series with lamp; voltmeter in parallel with lamp; variable resistor in series with lamp and all else correct; (ii) to change, voltage across / current through, lamp / in circuit; (b) (i) positive and negative; [1] (ii) electrons; [1] (iii) metal / named metal / graphite; [1] [Total: 7] 8 (a) (i) nitrogen 78% oxygen 21% others 1%;; [2] (all correct 2, two correct 1) (ii) carbon dioxide / water (vapour) / any noble gas ; [1] (b) (i) elements all atoms have same proton number / are found in Periodic Table / cannot be broken down into anything simpler; compounds [2] contain different types of atom / elements (bonded); (ii) covalent; bonded elements are both non-metals / compounds are gases; [2] (iii) pure water is neutral; rain water (generally) is (more) acidic (than pure water); rainwater during thunderstorm is the most acidic / owtte; likely to be caused by nitrogen oxides; which have reacted with the rain to form an acid; [max 2] [Total: 9] 9 (a) nutrition excretion respiration reproduction growth movement (1 mark for any two correct);; [max 2] (b) (i) adrenaline; [1] (ii) increases pulse rate; increases blood pressure; increases glucose in blood; [max 2] (c) in the blood / blood vessels; [1]