CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2013 series

0653 COMBINED SCIENCE

0653/62

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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- 1 (a) (i) brown colour around where the seeds/grains were, (blue/black elsewhere);
 - (ii) enzyme/seeds/grains breaking down/use/need/converted the starch;
 - (iii) control/shows that breakdown depends on living seeds, ORA;
 - (b) (i) (reducing) sugar is made (around the seeds/when the starch is broken down); [1]
 - (ii) <u>starch</u> digested/changed/converted to <u>sugar</u>; [1]
 - (c) (improved) reliability/because one seed might not be active/owtte; [1]
 - (d) (i) larger brown areas; [1]
 - (ii) smaller brown areas (accept no brown area/all blue-black); [1]
 - (e) (use different varieties of) seeds on different dishes/(different parts) of the same dish;

dish ;
keep (named) conditions constant ;
compare diameters/sizes of brown areas ;

[Total: 10]

- **2** (a) (i) 68.0; 86.2 (±0.1); [2]
 - (ii) 8(0), 16.2 (ecf);; [2]
 - (b) (i) points plotted correctly; (allow 1 error) suitable straight line drawn; [2]
 - (ii) clear evidence on graph; allow 1.2 to 1.3 inclusive; [2]
 - (c) 150/candidates answer = between 125 and 115g (ecf); [1]
 - (d) density = $m/l \times t \times w$ (any order); [1]

[Total: 10]

[max 2]

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	. age o		IGCSE – October/November 2013 0653		8.
3	(a) (i)	obse	DaCambridge		
	(ii)	milk	y/white/cloudy chalky ;		13
	(iii)	obse	ervation: white precipitate;		[1]
	(b) (i)	hydr	roxide		[1]
	(ii)		litmus (ignore colour) or Universal Indicator ; ervation: blue ;		[2]
	(iii)	brow	vn, yellow or orange ;		[1]
	OR	c) iron(III) chloride FeC l ₃ and copper carbonate CuCO ₃ ;; OR			
			carbonate $Fe_2(CO_3)_3$ and copper chloride $CuCl_2$;; for two correct names or formulae)		[max 2]
					[Total: 10]
4	(a) (i)		cm (no tolerance) ; cm (no tolerance) ;		[2]
	(ii)	not s	able scale and label on <i>x</i> axis; starting <i>y</i> axis at 0 ; oth curve drawn ;		[3]
	(iii)		17 or 0.042 cm/g; or 0.010 cm/g;		[2]
	(b) (i)	•	ws aorta to stretch) to allow surge of blood through/ ween beats/smoothes out blood flow/ <u>change</u> in press		[1]
	(ii)	resis	stant to bursting/breaking/tearing;		[1]
	(c) e.g.	. sam	e width of sample taken/same part of body of animal	/same animal ;	[1]

[Total: 10]

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5 (a) (i) 22.5; 25.1; 27.8;

(ii) C is the most concentrated, A is the least concentrated;

(b) (i) sodium ethanoate + water;

[1]

(ii) orange or yellow, (reject red);

[1]

(c) same volume of acid and alkali;

without indicator;

evaporate;

filter;

dry crystals with e.g. filter paper;

(any 4)

OR

evaporate;

(heat/boil etc.) to concentrate/crystallisation point/saturation etc.;

leave or cool;

filter;

dry crystals with e.g. filter paper;

(any 4)

[max 4]

[Total: 10]

6 (a) (i) the amplitude decreases/gets smaller;

[1]

(ii) 4.0 cm (±0.1 cm);

[1]

(iii) frequency = speed/wavelength, 10/4;

= 2.5 (Hz);

[2]

(b) (i) 1.1 cm (±0.1 cm);

[1]

(ii) (1.1/0.25) = 4.4 (ecf);

cm/s;

[2]

(c) (i) distance = $2.2 \, \text{cm} (\pm 0.1 \, \text{cm})$;

[1]

(ii) speed = 2.2/0.25 = 8.8 (ignore units, ecf);

[1]

(d) speed at 9th wave is greater than speed at 4th wave, owtte (must refer to two things, wave numbers, part numbers or speeds);

[1]

[Total: 10]