CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2014 series

0653 COMBINED SCIENCE

0653/23

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2014 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



Page 2	Mark Scheme	Syllabus	Paper
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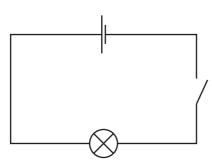
(a)	(i)	hydrogen;	[1]
	(ii)	flame; pops; (ecf for (a) (ii))	[2]
	(iii)	magnesium X copper; (i.e. X below magnesium and above copper) magnesium X G copper;	
		(i.e. G below magnesium and X in any order, and above copper)	[2]
	(iv)	${\sf zinc/iron/A}$ other metals with electronegativity between that of magnesium and iron ;	[1]
(b)	(i)	removal/loss of oxygen; gain of electrons;	[max 1]
	(ii)	carbon dioxide ;	[1]
(c)	(i)	P at or near negative electrode within electrolyte;	[1]
	(ii)	brom <u>ine</u> ;	[1]
			[Total 10]

Page 3	Mark Scheme	Syllabus	Paper
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2 (a) Sun; [1] [1] (b) (i) oak tree; (ii) beetles/greenfly/rabbits/squirrels; [1] (c) oak tree → beetles → blackbirds → hawks ;; oak tree → greenfly → frogs → hawks ;; [2] (1 mark correct sequence of organisms, 1 mark correct arrows) (d) their numbers may decrease; food supplies may become scarce; [2] (e) (concentration) increases; because less (carbon dioxide taken in for) photosynthesis; [2] [Total 9]

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3 (a)

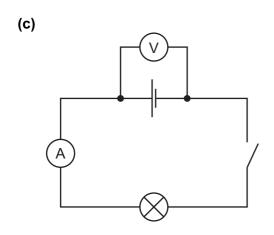


symbols all correct;

circuit connected correctly; (either one or two cells used)

[2]

- (b) (i) (2) lamp needs (p.d. of) 3V (to light), so needs 2 × 1.5=3V cells (owtte); [1]
 - (ii) lamp takes <u>current</u> of 1.2A when lit (owtte); [1]



voltmeter connected correctly; ammeter connected correctly;

[2]

[Total 6]

Page 5	Mark Scheme	Mark Scheme Syllabus F	
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[1] 4 (a) (i) fractional distillation/fractionation; (ii) the lower the boiling point, the higher up the tower it is released/ condenses; [1] (iii) gasoline (petrol)/diesel/fuel oil/A kerosene; used as <u>fue</u>l for transport/heating; [2] (b) nitrogen: 78%; [2] oxygen: 21%; (c) (i) increase in water (vapour); increase in carbon dioxide; decrease in oxygen; [max 2] temperature increases; (ii) heat energy released/temperature increases; new substance(s) are formed; [2]

Page 6	Mark Scheme	Syllabus	Paper
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5

correct name; correct box;

(a)		ht hand) <i>no mark</i> ige laterally inverte	ed (owtte);			[1]
(b)	(i)	electrical (energy	/) → sound (energy)			[1]
	(ii)		within human range above 20 Hz and lo			[1]
(c)	(i)	speed = distance speed = 25/40 = metres/second/	= 0.625/0.63;			[3]
	(ii)	(100 N) (forces) are <u>equa</u>	<u>al</u> ;			[1]
	(iii)	one complete wa	avelength correctly m	narked and labelled	;	[1]
	(iv)	amplitude/freque	ency;			[1]
(d)						1
		X- rays	visible light	infra- red	microwaves	

[Total 11]

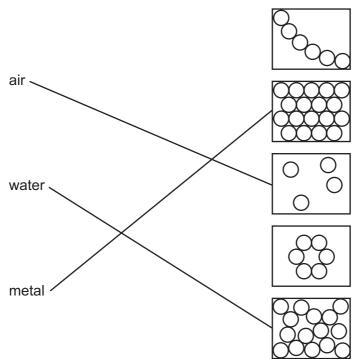
[2]

Page 7	Mark Scheme	e Syllabus	
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6	(a)	(i)	zygote/one of the ball of cells;	[1]
		(ii)	fertilization;	[1]
	(b)		terus/womb ; olants/embeds) in wall/lining of uterus ;	[2]
	(c)	(i)	vitamin D A A/B/E/K; correct use of named vitamin;	[2]
	(d)		× 37 ; 40.6/141 ;	[2]
				[Total 8]
7	(a)		ser in colour/gas to solid/increasing, mp/bp / density, down the group ; st state trend and direction)	[1]
	(b)	(i)	yellow/orange colouration;	[1]
		(ii)	chlorine + potassium bromide → potassium chloride + bromine LHS ;	
			RHS;	[2]
	(c)	cov	alent;	[1]
	(d)		tes water safe for consumption ; bacteria ;	[2]
				[Total 7]

Page 8	Page 8 Mark Scheme		Paper
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8 (a)



[2]

- (b) (i) more energetic water molecules escape into air; remaining water has less (thermal) energy (so cooler) (owtte); [2]
 - (ii) cooler water takes heat from air/water takes heat from warmer air; [1]
- (c) allow space for (thermal) expansion; [1]

(d) (i)
$$30 \times 15 \times 10 = 4500 \text{ (cm}^3)$$
; [1]

(ii) (density =) mass/volume/(d =) m/V; d = 7500/4500 = 1.7/1.67 (g/cm³); (ecf) [2]

[Total 9]

Page 9	Page 9 Mark Scheme		Paper
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9 (a)

diagram	name of cell	function of cell
	red blood cell	transport of oxygen;
	white blood cell	defence against disease / phagocytosis;

[4]

(b) right;
 pulmonary artery;

valves;

[3]

(c) (i) oxygen; [1]

(ii) glucose/sugar/amino acids/(any named) vitamin/(named) mineral/water/carbon dioxide;; [2]

[Total: 10]