CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the October/November 2014 series

0653 COMBINED SCIENCE

0653/22

Paper 2 (Core Theory), maximum raw mark 80

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Pa	ge 2	2	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2014	0653	22
1	(a)	cop car	copper oxide (loses oxygen so) is reduced / copper ions gain electrons ; carbon (gains oxygen so) is oxidised ;		[2]
	(b)	(i)	electrodes correctly labelled anode and cathode ; electrolyte labelled ;		[2]
		(ii)	at the positive electrode brom <u>in</u> e and at the negative electrode lead lead appears as a, grey / metallic, deposit / bead of molten metal ; brom <u>in</u> e, is a brown gas / causes a brown colouration of electrolyte	d; ;;	[3]
					[Total: 7]
2	(a)	(i)	speed = distance/time / (time =)distance / speed ; 200/40 = 5 (s) ;		[2]
		(ii)	$\begin{array}{l} 40\text{m/s}=40\times60\times60\text{m/h}\ (=144000\text{m/h})\ ;\\ 40\times60\times60\text{m/h}=40\times60\times60\text{/}1000\text{km/h}=144\ (\text{km/h})\ ; \end{array}$		[2]
	(b)	sp	eed fime		
		hor foll	izontal straight line ; owed by descending line, straight or curved, to meet time axis ;		[2]
	(c)	(i)	(400 N – no mark) for constant speed, forces must be equal and opposite (owtte) ;		[1]
		(ii)	chemical energy in the rider ; heat/thermal energy during braking ; allow sound		[2]
					[Total: 9]

Page 3		Mark Scheme	Syllabus	Paper
		Cambridge IGCSE – October/November 2014	0653	22
3 (a) /	At Bb	rachea ; pronchiole ;		[2]
(b)	brea volu	athing rate increase ; ime / depth (of breathing) increased ;		[2]
(c)	(i)	more carbon dioxide in exhaled air / less carbon dioxide in inhaled	air ;	[1]
	(ii)	(after exercise) exhaled air contains more carbon dioxide / ora ; use of numbers from data (e.g. exhaled air contains about four time as much carbon dioxide) ;	es	[2]
	(iii)	no carbon dioxide present ; not enough carbon dioxide in air to show a result ;		[2]
				[Total: 9]

4 (a) (i)

complete series circuit ; battery of 4 cells connected correctly ;

[2]

[1]

(ii)



symbol with correct connections (both required)	[1]

- (b) (i) number of vibrations / waves per unit time ;
 - (ii) amplitude increased ; frequency unchanged ; [2]

Page 4		Mark	Scheme		Syllabus	Paper
v		Cambridge IGCSE –	October/November 2014		0653	22
(c)	(i)	resistance = $6 / 2 = 3$; (units) ohms / Ω ;				[2]
	(ii)	current increased / doubled ; in parallel circuits, the current owtte) :	from the source is larger tha	n the cu	rrent in each	h branch /
		resistance is lower ;				[max 2]
						[Total: 10]
5 (a)	(i)					
			in nucleus	outside	nucleus	
		number of protons	6	0		
		number of neutrons	6	0		
		number of electrons	0	6		
			column correct ;	(column corr	ect ; [2]
	(ii)	equal numbers of protons and equal numbers of positive and protons are positive and elect	d electrons ; d negative charges ; trons are negative ;			[max 2]
(b)	(i)	natural gas / petroleum / refinery gas / rice fields / from biodegradation / digestive activity of ruminants ;			tive [1]	
	(ii)	methane + oxygen → carbon LHS ; RHS ;	dioxide + water			[2]
(c)	(i)	CH ₄ ;				[1]
	(ii)	covalent ;				[1]
	(iii)	2;				[1]
						[Total: 10]

P	age 5	5	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2014	0653	22
6	(a)	infr	a-red ;		[1]
	(b)	mo mo	lecules have more energy so more of them are moving faster (owtte re molecules have enough KE / moving fast enough to escape (from) ; i surface) ;	[2]
	(c)	nee	ed a medium for conduction & convection / no medium in space (owt	te) ;	[1]
	(d)		<i>i</i> upper face of plastic		
			r		
		lo	ower face of plastic		
			Fig 2.2	2	
		refr nor em	acted ray in plastic bent towards normal ; mal drawn at upper face with angles of incidence and refraction corr ergent ray parallel to incident ray ;	ectly marke	d ; [3]
					[Total: 7]
					[1010117]
7	(a)	(i)	phototropism ;		[1]
		(ii)	more / better absorption of light ; more / better photosynthesis ; any statement about light hitting leaves at right angles / not at an a	ngle ;	[max 2]
		(iii)	sensitivity ; movement ; growth ;		[max 2]
	(b)	(i)	shoot X bends towards the light / responds ; shoots Y and Z do not ;		[2]
		(ii)	the tip of the shoot detects the light / controls the response ; because no response occurs when tip is covered/removed ;		[2]
	(c)	give	es more glucose into blood ;		
		incr ma	eases pulse rate ; kes more energy available from respiration / speeds up metabolism	;	[max 2]
				I	Total: 11]

P	age 6	6	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2014	0653	22
8	(a)	wo col eva	rkable filtration equipment ; llection of filtrate ; aporation ;		[3]
	(b)	Г			
			cannot be broken down into simpler substances		
			contains different atoms chemically bonded together	pound	
			ele	ement	
			contains more than one type of atom which are not		
			bonded together m	xture	
			contains one type of atom.		
					[4]
	(c)	(i)	aluminium (atoms) lose electrons ; sulfur (atoms) gain electrons ; electrons are transferred from aluminium to sulfur (atoms) ;;		[max 2]
		(ii)	Al ₂ S ₃ ;		[1]
				[Total: 10]
9	(a)	(i)	cervix correctly labelled ; vagina correctly labelled ;		[2]
		(ii)	ovary correctly labelled ;		[1]
	(b)	(i)	oviduct / fallopian tube ;		[1]
		(ii)	uterus ; (embedded) in lining ;		[2]
	(c)	sha	aring needles / blood transfusions / avp ;		[1]
					[Total: 7]