# Cambridge IGCSE<sup>™</sup>

#### **COMBINED SCIENCE**

0653/11

Paper 1 Multiple Choice (Core)

October/November 2020

45 minutes

You must answer on the multiple choice answer sheet.

You will need: Multiple choice answer sheet

Soft clean eraser

Soft pencil (type B or HB is recommended)

#### **INSTRUCTIONS**

- There are forty questions on this paper. Answer all questions.
- For each question there are four possible answers **A**, **B**, **C** and **D**. Choose the **one** you consider correct and record your choice in soft pencil on the multiple choice answer sheet.
- Follow the instructions on the multiple choice answer sheet.
- Write in soft pencil.
- Write your name, centre number and candidate number on the multiple choice answer sheet in the spaces provided unless this has been done for you.
- Do **not** use correction fluid.
- Do **not** write on any bar codes.
- You may use a calculator.

#### **INFORMATION**

- The total mark for this paper is 40.
- Each correct answer will score one mark. A mark will not be deducted for a wrong answer.
- Any rough working should be done on this question paper.
- The Periodic Table is printed in the question paper.

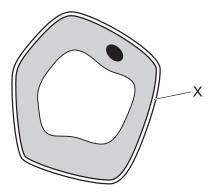
This document has 16 pages. Blank pages are indicated.

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1 A motor car has many features that are similar to the characteristics of living things.

Which characteristic of living organisms cannot be matched to a feature of a car?

- A release of energy by breaking down energy-rich molecules
- B release of waste materials
- C reproduction of a similar version of itself
- **D** takes in materials for energy
- 2 The diagram shows a cell.



What is the function of X?

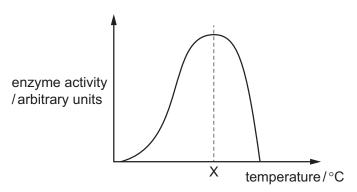
- A contains the genetic information
- B controls substances entering and leaving the cell
- C maintains the shape of the cell
- **D** photosynthesis
- 3 A sample of food contains only fat, protein and water.

Which food tests give a positive result when the sample is tested?

	Benedict's solution	biuret solution	ethanol emulsion	iodine solution
Α	✓	✓	X	X
В	✓	X	X	✓
С	X	✓	✓	X
D	X	✓	✓	✓

3

4 The diagram shows how the activity of an enzyme changes with temperature.



This enzyme works in the human body.

What is the most likely value of temperature X?

- **A** 10 °C
- **B** 40 °C
- **C** 70 °C
- **D** 100 °C

**5** What is necessary for photosynthesis?

- 1 carbon dioxide
- 2 chlorophyll
- 3 glucose
- 4 light
- 5 oxygen
- 6 water
- **A** 1, 2, 4 and 6
- **B** 1, 3, 4 and 6
- **C** 2, 3, 4 and 5
- **D** 3, 4, 5 and 6

6 Which breakdown processes occur inside cells, and which occur outside cells?

	large molecules to small molecules for absorption	breakdown of glucose to release energy		
Α	inside	inside		
В	inside	outside		
С	outside	inside		
D	outside	outside		

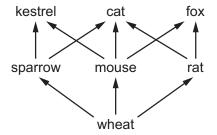
- 7 Which statement about the pulmonary vein is correct?
  - A It carries blood from the heart to the body.
  - **B** It carries blood from the heart to the lungs.
  - **C** It carries blood to the heart from the body.
  - **D** It carries blood to the heart from the lungs.
- **8** Which statement about adrenaline is correct?
  - **A** It is produced by a gland.
  - **B** It is transported in the red blood cells.
  - C It only has one target organ.
  - **D** It reduces the size of the pupils.
- **9** Which row about whether roots and shoots respond to gravity is correct?

	respond to gravity					
	roots shoots					
Α	no	no				
В	no	yes				
С	yes no					
D	yes	yes				

10 Which row describes asexual reproduction?

	number of parents	a zygote is produced	offspring identical to the parent
Α	1	no	yes
В	1	yes	no
С	2	no	yes
D	2	yes	no

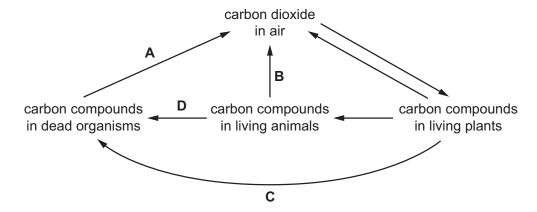
- 11 Which conditions are needed for the germination of seeds?
  - 1 oxygen
  - 2 suitable temperature
  - 3 water
  - **A** 1 and 2 only **B** 1 and 3 only **C** 2 and 3 only **D** 1, 2 and 3
- 12 The diagram shows a food web.



Which sentence correctly describes an organism in the food web?

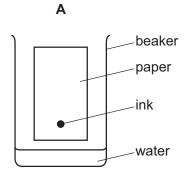
- **A** The cat is a herbivore and consumer.
- **B** The kestrel is a carnivore and consumer.
- **C** The mouse is a herbivore and producer.
- **D** The sparrow is a carnivore and consumer.
- 13 The diagram shows part of the carbon cycle.

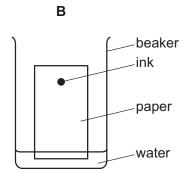
Which arrow represents respiration by decomposers?



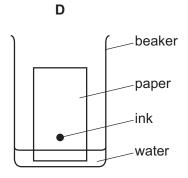
- 14 Which term describes ammonia, NH<sub>3</sub>?
  - **A** element
  - **B** ion
  - C atom
  - **D** molecule
- 15 Chromatography separates ink into different colours.

Which diagram shows how the apparatus is set up?





beaker paper ink water



**16** Sugar dissolves in water to form sugar solution.

Which word describes the sugar?

- **A** distillate
- **B** filtrate
- C solute
- **D** solvent



17 Which equation is balanced?

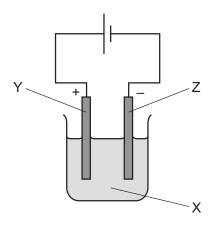
$$\textbf{A} \quad Cu \ + \ O_2 \ \rightarrow \ CuO$$

$$\textbf{B} \quad 2\text{CO} \, + \, \text{O}_2 \, \rightarrow \, 2\text{CO}_2$$

$$\textbf{C} \quad 2 \text{Mg} \ \textbf{+} \ 2 \text{HC} l \ \rightarrow \ \text{MgC} l_2 \ \textbf{+} \ \text{H}_2$$

$$\mathbf{D} \quad 2N_2 + 3H_2 \rightarrow 2NH_3$$

18 The diagram shows apparatus used for electrolysis.



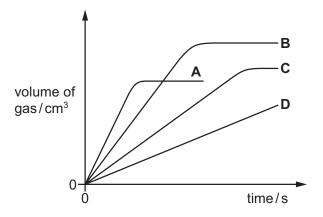
What are X, Y and Z?

	Х	Y	Z
Α	electrode	anode	cathode
В	electrode	cathode	anode
С	electrolyte	anode	cathode
D	electrolyte	cathode	anode

19 Which temperature changes occur during exothermic and endothermic reactions?

	exothermic	endothermic
Α	decreases increases	
В	decreases	no change
С	increases	decreases
D	increases	no change

20 Which line on the graph represents the reaction with the greatest rate?



- 21 Which substances react with dilute sulfuric acid to make copper sulfate?
  - 1 copper
  - 2 copper carbonate
  - 3 copper hydroxide
  - 4 copper nitrate
  - **A** 1 and 2 **B** 1 and 4 **C** 2 and 3 **D** 3 and 4
- 22 Acid X reacts with metal Y.

A colourless gas is given off and a pale green solution is produced.

Two tests are carried out on the solution.

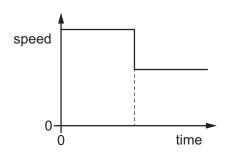
test	reagent(s) added	result
1	aqueous silver nitrate and nitric acid	white precipitate
2	aqueous sodium hydroxide	green precipitate

What are acid X and metal Y?

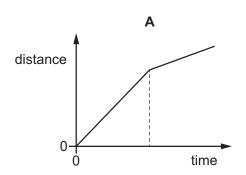
	acid	metal		
Α	hydrochloric	iron		
В	hydrochloric	zinc		
С	sulfuric	iron		
D	sulfuric	zinc		

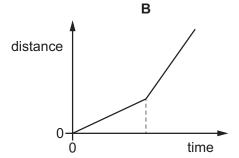
23	vvn	which statement about the Periodic Table is <b>not</b> correct?								
	Α	It lists all the known elements.								
	В	Non-metallic elements are placed on the right-hand side.								
	С	The position of an element helps us to predict its properties.								
	D	Vertical column	s in	the Periodic Tal	ole a	re called <sub>l</sub>	periods.			
24	Am	monia, NH₃, can	be	made by combir	ning '	the gases	nitrogen,	N <sub>2</sub> , and	hydroge	∙n, H <sub>2</sub> .
	Thi	s reaction is slow	<b>/</b> .							
	Wh	en element Y is	add	ed, the rate of re	actio	on increas	ses.			
	Wh	at is Y?								
	Α	Al	В	Fe	С	Rb	D	$I_2$		
25	Wh	ich method is us	ed t	o extract copper	fron	n copper(	II) oxide?			
	A	dissolving copp	er(I	I) oxide in hydro	chlo	ric acid ar	nd then filte	ering		
	В	dissolving copp	er(I	I) oxide in water	and	then filter	ring			
	С	heating the cop	per	(II) oxide						
	D	heating the cop	per	(II) oxide mixed	with	carbon				
26	Wh	ich process does	s no	t produce carbo	n dic	oxide?				
	Α	fractional distilla	atior	of petroleum						
	В	·		ustion of methar						
	С			en an acid and a						
	D	the thermal decomposition of calcium carbonate								
27	Wh	Vhich statement describes a hydrocarbon?								
	Α									
	В	·		ontains carbon a			, ,			
	С	·		nly contains ionic		, ,	,			
	D	·		acts easily with						
		•		•						

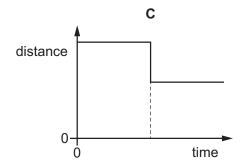
28 The diagram shows the speed-time graph for a car travelling along a horizontal road.

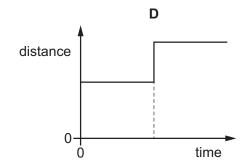


What is the distance-time graph for this part of the journey?









29 A solid metal cube of side 5.0 cm has a mass of 250 g.

What is the density of the metal?

- $\mathbf{A} \quad 0.50\,\mathrm{g/cm^3}$
- **B** 2.0 g/cm<sup>3</sup>
- **C** 10 g/cm<sup>3</sup>
- $D = 50 \,\mathrm{g/cm^3}$

30 A raindrop falls vertically at a constant speed.

What is the resultant force on the raindrop as it falls?

- A It is equal to the air pressure on the drop.
- **B** It is equal to the air resistance on the drop.
- **C** It is equal to the weight of the drop.
- D It is zero.

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31 An object is not moving. Work is done on the object and it moves a known distance.

What other single measurement is needed to calculate the amount of work done?

- A the acceleration of the object
- **B** the direction of movement of the object
- **C** the force applied to the object in the direction of movement
- **D** the speed of the object
- 32 An apple falls to the ground.

Which form of energy decreases as the apple falls?

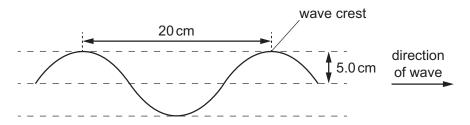
- A chemical potential
- **B** gravitational potential
- **C** kinetic
- **D** sound
- 33 What happens as a liquid starts to evaporate?
  - A The mass of the remaining liquid increases.
  - **B** The mass of the remaining liquid is constant.
  - **C** The temperature of the remaining liquid decreases.
  - **D** The temperature of the remaining liquid increases.
- 34 In which states of matter can convection occur?

	in a solid	in a liquid	in a gas
Α	no	no	yes
В	no	yes	yes
С	yes	no	no
D	yes	yes	no

## **35** The diagram shows a section of a rope.

Four wave crests pass a point on the rope every second.

Each wave crest travels 80 cm in one second.

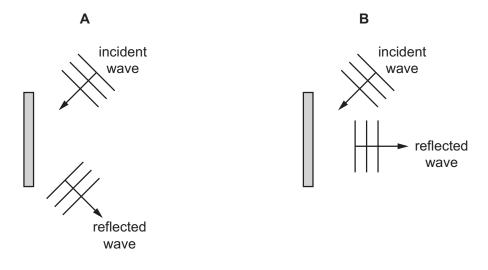


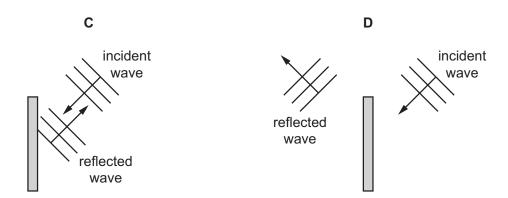
What is the speed of the wave?

- **A** 4.0 cm/s
- **B** 5.0 cm/s
- **C** 20 cm/s
- **D** 80 cm/s

## **36** A water wave strikes a plane barrier.

Which diagram shows the direction of travel of the reflected wave?





- 37 Which substance is an electrical insulator?
  - **A** aluminium
  - **B** copper
  - **C** plastic
  - **D** steel
- **38** A power supply causes a current in a circuit.

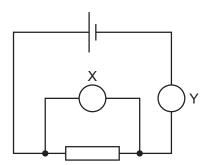
The electromotive force (e.m.f.) of the power supply and the resistance of the circuit are both changed.

Which pair of changes must result in a smaller current in the circuit?

	e.m.f.	resistance		
Α	decreased	decreased		
В	decreased	increased		
С	increased	decreased		
D	increased	increased		

**39** The diagram shows a cell connected to a resistor and two meters, X and Y.

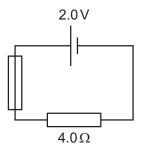
The circuit is used when determining the resistance of the resistor.



What are the quantities measured by meters X and Y, and what are their correct units?

	met	er X	meter Y		
	quantity	unit	quantity	unit	
Α	current	А	p.d.	V	
В	current	V	p.d.	Α	
С	p.d.	Α	current	V	
D	p.d.	V	current	Α	

**40** The diagram shows a circuit containing a 2.0 V cell, a fuse and a resistor of resistance 4.0  $\Omega$ .



What is the most appropriate rating for the fuse?

- **A** 0.4 A
- **B** 1A
- **C** 2A
- **D** 5A

15

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The Periodic Table of Elements

		2	Ε	helium 4	10	Ne	neon 20	18	Ā	argon 40	36	궃	krypton 84	54	Xe	xenon 131	98	格	radon			
	II/				6	ш	fluorine 19	17	Cl	chlorine 35.5	35	ğ	bromine 80	53	Н	iodine 127	85	¥	astatine -			
	<b>-</b>				80	0	oxygen 16	16	ഗ	sulfur 32	34	Se	selenium 79	52	<u>e</u>	tellurium 128	84	Ъ	polonium -	116		livermorium –
	>				7	z	nitrogen 14	15	۵	phosphorus 31	33	As	arsenic 75	51	Sp	antimony 122	83	Ξ	bismuth 209			
	2				9	ပ	carbon 12	14	S	silicon 28	32	Ge	germanium 73	50	S	tin 119	82	Ъ	lead 207	114	Εl	flerovium
	≡				2	Δ	boron 11	13	Αl	aluminium 27	31	Ga	gallium 70	49	I	indium 115	81	11	thallium 204			
		•									30	Zu	zinc 65	48	පි	cadmium 112	80	Рg	mercury 201	112	ပ်	copernicium
											29	J	copper 64	47	Ag	silver 108	62	Αu	gold 197	111	Rg	roentgenium -
dn											28	z	nickel 59	46	Pd	palladium 106	78	瓧	platinum 195	110	Ds	darmstadtium -
Group											27	ပိ	cobalt 59	45	뫈	rhodium 103	77	'n	iridium 192	109	¥	meitnerium -
		- 1	I	hydrogen 1							26	Ьe	iron 56	44	Ru	ruthenium 101	92	Os	osmium 190	108	Hs	hassium
					,						25	Mn	manganese 55	43	ည	technetium -	75	Re	rhenium 186	107	Bh	bohrium
						log	sse				24	ပ်	chromium 52	42	Mo	molybdenum 96	74	>	tungsten 184	106	Sg	seaborgium -
				Key	atomic number	atomic symbo	name relative atomic mass				23	>	vanadium 51	41	q	niobium 93	73	Б	tantalum 181	105	Ob	dubnium
						ato	rela				22	F	titanium 48	40	Zr	zirconium 91	72	Ξ	hafnium 178	104	꿒	rutherfordium -
											21	လွ	scandium 45	39	>	yttrium 89	57–71	lanthanoids		89–103	actinoids	
	=				4	Be	beryllium 9	12	Mg	magnesium 24	20	Ca	calcium 40	38	ഗ്	strontium 88	99	Ba	barium 137	88	Ra	radium
	_				က	<u></u>	lithium 7	7	Na	sodium 23	19	¥	potassium 39	37	8	rubidium 85	55	Cs	caesium 133	87	Ļ	francium -

rı Lu	lutetium 175	103	۲	lawrencium	1
° X	ytterbium 173	102	8	nobelium	1
e9 Tm	thulium 169	101	Md	mendelevium	1
88 Ē	erbium 167	100	Fm	fermium	1
67 Ho	holmium 165	66	Es	einsteinium	1
。 5	dysprosium 163	86	ŭ	californium	I
es Tb	terbium 159	26	益	berkelium	ı
64 Gd	gadolinium 157	96	Cm	curium	I
ез П	europium 152	92	Am	americium	1
Sm	samarium 150	94	Pn	plutonium	1
Pm	promethium -	93	δ	neptunium	1
9 PX	neodymium 144	92	$\supset$	uranium	238
59 Pr	praseodymium 141	91	Ра	protactinium	231
Se O	cerium 140	06	드	thorium	232
57 La	lanthanum 139	88	Ac	actinium	1

lanthanoids

actinoids

The volume of one mole of any gas is  $24\,\mathrm{dm}^3$  at room temperature and pressure (r.t.p.).

