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0442 CO-ORDINATED SCIENCES (US)

0442/23

Paper 2 (Core), maximum raw mark 120

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

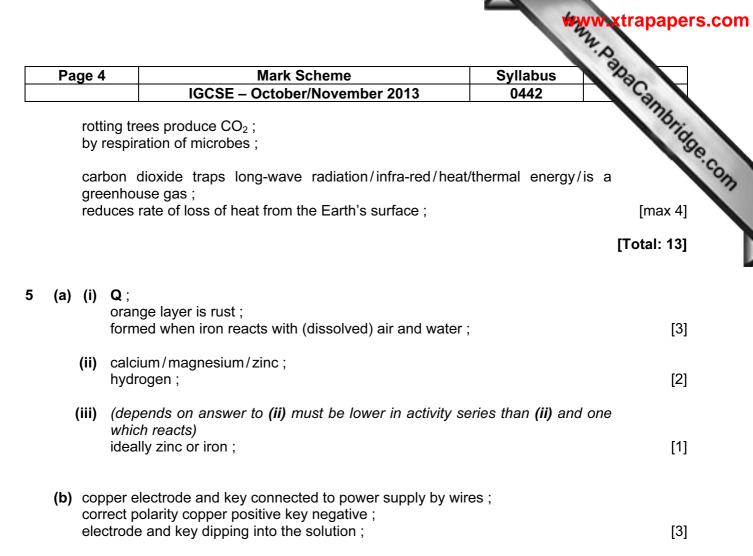
Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2013 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.

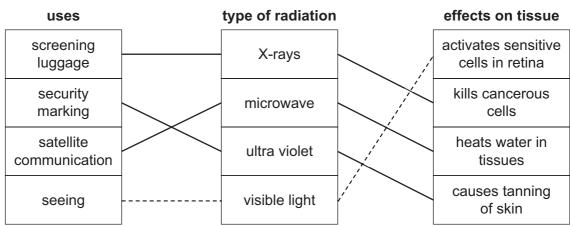
				us cambridge.com
Page 2				is Phan
			IGCSE – October/November 2013 0442	any
1	(a)		o cell membrane ; o nucleus ;	Stick
			o cell wall/large vacuole ;	Se.C.
				ST.
	(b)	wate		
		min	neral salts/named mineral ;	[2]
	(c)	(i)	transport ;	
	(-)	(-)	of sugars/substances made in the leaves ;	[2]
		(ii)	roots have no sucrose/short of nutrients;	
		-	no source of energy/cannot respire ;	[2]
				[Total: 9]
2	(a)	(i)	argon/Ar ;	[1]
		(ii)	calcium/lithium and oxygen/sulfur/fluorine ; metal with non-metal ;	[2]
	(b)	(i)	nucleus ;	[1]
		(ii)	15;	
			same as number of electrons/3 shells = Period 3, 5 outer electror Group V so must be phosphorus which has proton number 15 ;	ons means [2]
	(c)	(i)	magnesium sulphate ;	[1]
		(ii)	filter mixture (W) ;	
		-	dry the solid ; use balance to find mass/weigh it ;	[3]
				[Total: 10]
3	(a)	(i)	change resistance (of circuit)/change current through resistor ;	[1]
		(ii)	X – ammeters need to be in series in a circuit ;	[1]
		(iii)	R = V/I ;	
			$= 8/0.6 = 13.3\Omega$;	[2]

Page 3	A Mark Scheme		102		
	IGCSE – October/November 2013	0442	·Car		
stay	reases reases /s the same orrect = 2 marks, 2 correct = 1 mark ;;		ww.xtrapape w.pabacambridg [2]		
(c) leng dian	gth ; neter/cross sectional area ;		[2]		
mix pour	sh/chop ; with, ethanol/alcohol ; r into water ; ay appearance indicates presence of fat ;		[Total: 8] [max 3]		
	growth ; repair ; er use of protein ;		[max 2]		
	eases surface area ; a of making it easier for enzymes to make contact ;		[2]		
to a	eins not digested ; mino acids ; eins cannot be absorbed/amino acids cannot be absorbe	ed ;	[max 2]		
area	uction of habitat ; a too small to support populations/reduction in bi		tion /		
flood due	species become endangered/lack of opportunity to find new medicines ; flooding/leaching of minerals ; due to rain falling directly on soil/lack of protection of tree canopy/increased runoff ;				
	erosion ; to lack of tree roots ;				
	drought ; due to lack of transpiration by trees to form rain leading to desertification ;				
	er trees to photosynthesise/less photosynthesis ; emove carbon dioxide ;				



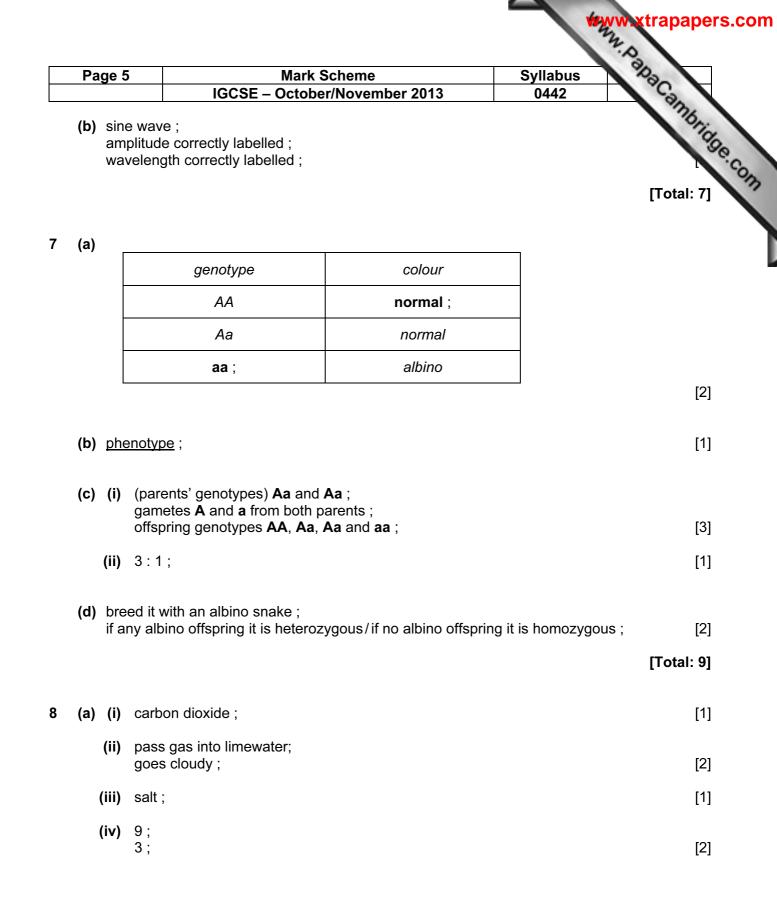
[Total: 9]

6 (a)



three correct linkages uses to type = two marks ;; one correct linkages uses to type = one mark ; three correct linkages type to effect = two marks ;; one correct linkages type to effect = one mark ;

[4]



Page	6 Mark Scheme Syllabus	20
	IGCSE – October/November 2013 0442	an,
(b) (i)	identifies higher pH with lower acid concentration ; lowering acid concentration decreases the rate ;	da Cambridg
(ii)	temperature ;	
	surface area of calcium carbonate ; degree of agitation of the mixture ;	
		[max 2]
		[Total: 10]
(a) (i)	80m ;	[1]
(ii)	(speed = distance/time = 50/10 =) 5 m/s ;	[1]
(iii)	not moving ;	[1]
(iv)	unbalanced because speed is changing ;	[1]
(b) ge	othermal/hydroelectricity/waves/wind/biomass;	[1]
(c) (i)	kinetic energy ;	[1]
(ii)	(gravitational) potential energy ;	[1]
	nsity = mass/volume; I5/36 = 1.25 g/cm ³ ;	[2]
		[~]
(e) (i)		
	irregular arrangement ;	[2]
(ii)	particles move faster therefore more collisions (with tyre wall);	[1]
(iii)	body/heat energy in body reduced by sweating ; kinetic energy of water molecules increase/water molecules move faster ; faster moving/more energetic (water) molecules escape/leave the surface/water molecules turn to gas/vapour ;	
	break bonds/break forces of attraction between molecules ; (KE) /energy of (remaining) water molecules (in sweat) decreases ;	[max 3]

Page	e 7 Mark Scheme Syl					12
			ctober/Novembe	r 2013	0442	C.
	 (a) A trachea ; B lung ; 					WWW xtrapape
(b) (i)	from re	ent of molecules gion of high cond concentration gi	centration to low c	oncentration ;		[max 2]
(ii)) plasma	;				[1]
(iii)	referen so more	ce to respiration, e carbon dioxide	e muscle contracti / oxidation of gluco produced by cells diffuses into the b	ose; s;		[max 2]
(iv)			gradient (from blo	ood to alveolus);	[2]
						[Total: 9]
(a) co	oal/peat;					[1]
(b) (i)) <u>fraction</u>	al_distillation/fra	ctionation ;			[1]
(ii)			nuch energy when	ı burnt ;		[2]
(c) (i)	۲ ۱ ۲ ۲ ۲ ۲ ۲ 2 C joir	I H ├ ─ H I H ned by single bor				
		single bonded to				[2]
(ii)		+ oxygen → carl or 1 mark and RF	bon dioxide + wate IS for 1 mark)	er ;;		[2]
(d) (i)) crackin	g;				[1]
(ii)		ains oxygen ; t would burn inst	tead of crack/owt	to ·		[2]
	Teactar			le,		ر~ ا

