

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS
International General Certificate of Secondary Education

CO-ORDINATED SCIENCES

0654/01

Paper 1 Multiple Choice

May/June 2005

45 minutes

Additional Materials: Multiple Choice Answer Sheet
Soft clean eraser
Soft pencil (type B or HB is recommended)

READ THESE INSTRUCTIONS FIRST

Write in soft pencil.

Do not use staples, paper clips, highlighters, glue or correction fluid.

Write your name, Centre number and candidate number on the answer sheet in the spaces provided unless this has been done for you.

There are **forty** questions on this paper. Answer **all** questions.

For each question there are four possible answers **A, B, C** and **D**. Choose the **one** you consider correct and record your choice in **soft pencil** on the separate answer sheet.

Read the instructions on the answer sheet very carefully.

Each correct answer will score one mark. A mark will not be deducted for a wrong answer.

Any rough working should be done in this booklet.

A copy of the Periodic Table is printed on page 20.

This document consists of **17** printed pages and **3** blank pages.

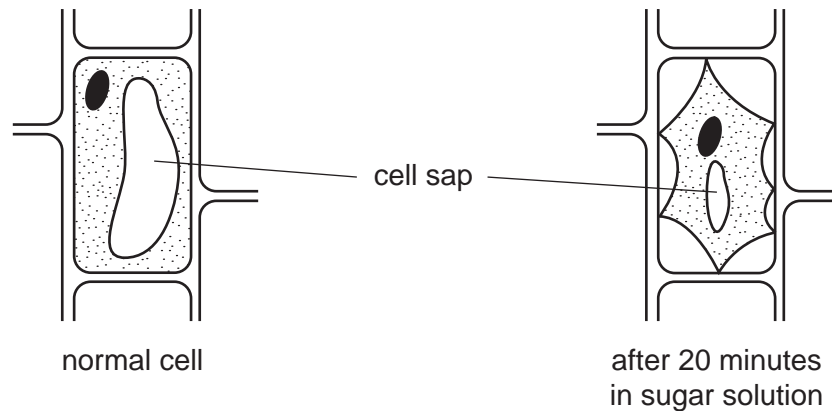


2

- 1 An animal is observed swimming in a river. It has legs, but no fins. Its skin is scaly.

To which class of vertebrates does this animal belong?

- A amphibians
 - B fish
 - C mammals
 - D reptiles
- 2 The diagrams show a normal plant cell, and a cell from the same plant, which has been in a sugar solution for 20 minutes.

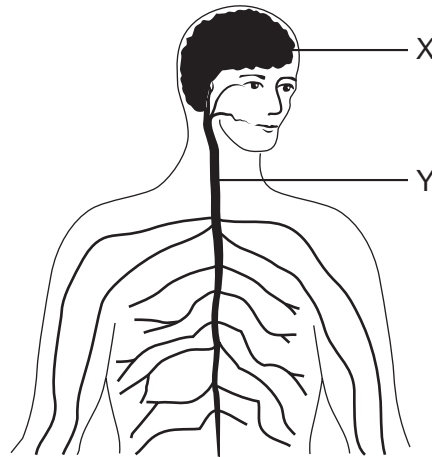


What explains this change?

- A The sugar solution is less concentrated than the cell sap.
- B The sugar solution is more concentrated than the cell sap.
- C The sugar solution is the same concentration as the cell sap.
- D The sugar solution has killed the cell.

3

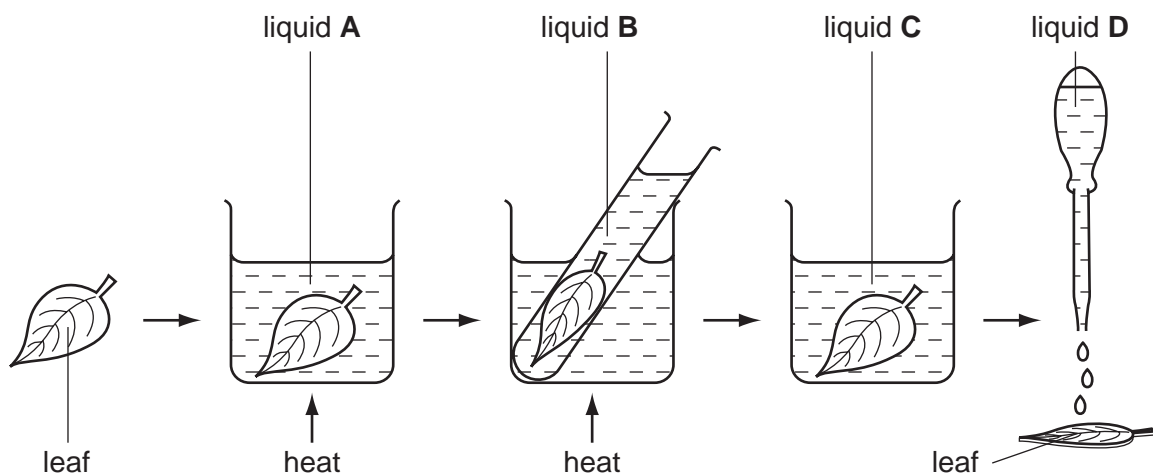
3 The diagram represents part of the human nervous system.



What name is given to X and Y together?

- A brain
 - B central nervous system
 - C nerve
 - D spinal cord
- 4 The diagram shows the stages in testing a green leaf for starch.

Which liquid is alcohol (methylated spirits)?

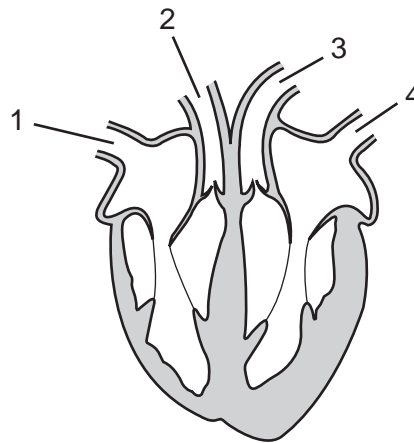


5 Which word equation represents aerobic respiration?

- A glucose \rightarrow carbon dioxide + ethanol
- B glucose \rightarrow lactic acid
- C glucose + oxygen \rightarrow carbon dioxide + water
- D glucose + oxygen \rightarrow lactic acid

4

6 The diagram shows a section through the heart.



Which two blood vessels are arteries?

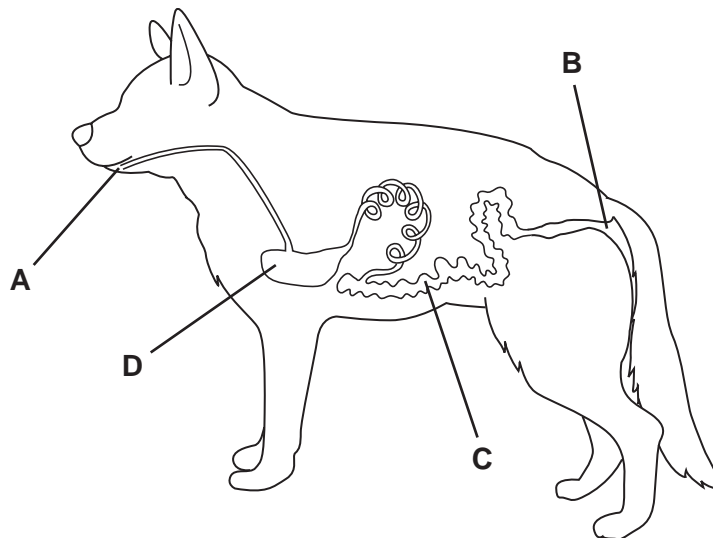
- A** 1 and 2 **B** 2 and 3 **C** 3 and 4 **D** 4 and 1

7 How do bacteria cause tooth decay?

- A** They release alkalis that dissolve enamel.
B They release ethanol that digests enamel.
C They release acids that dissolve enamel.
D They release enzymes that digest enamel.

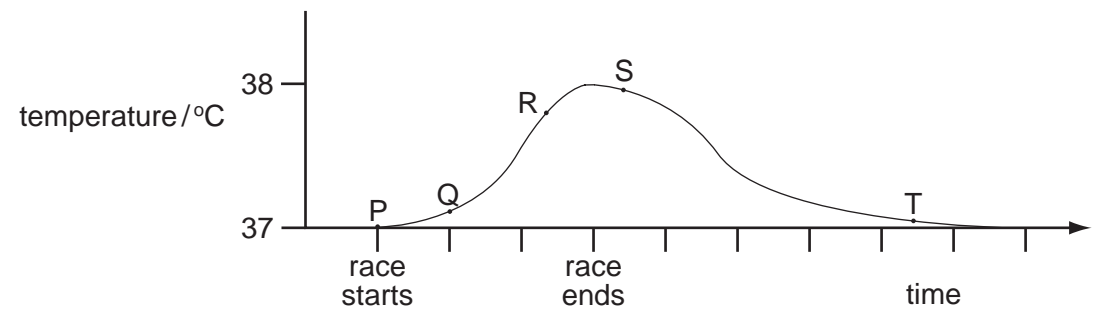
8 The diagram shows the alimentary canal of a dog.

Where does egestion occur?



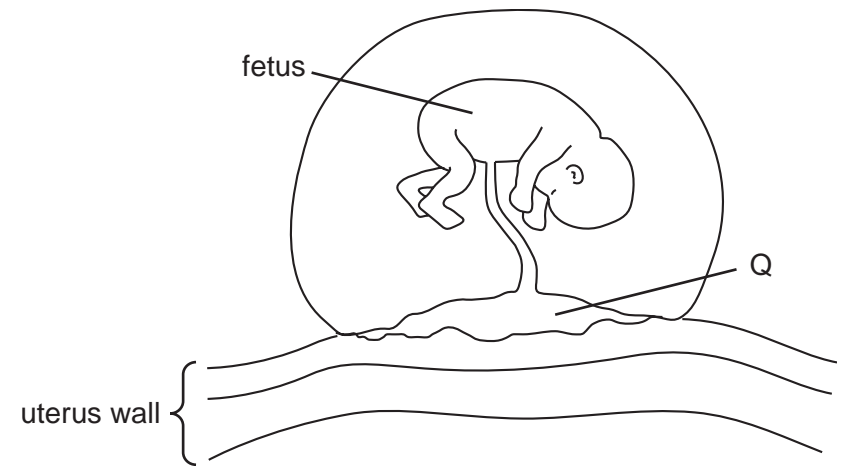
- 9 Which shows the sequence that occurs when a person becomes aware of light?
- A impulse → stimulus → receptor → spinal cord
 - B receptor → stimulus → impulse → brain
 - C stimulus → impulse → receptor → spinal cord
 - D stimulus → receptor → impulse → brain

10 The graph shows body temperature before, during and after running a race on a hot day.



Which stage of the graph occurs as a result of homeostasis?

- A P to Q
 - B Q to R
 - C R to S
 - D S to T
- 11 The diagram shows a developing fetus attached to the uterus wall.



What is the function of Q?

- A draining amniotic fluid
- B passing blood from the mother to the fetus
- C supplying carbon dioxide to the fetus
- D supplying oxygen to the fetus

12 Cystic fibrosis is an inherited disease.

Only people who are homozygous recessive, ff, suffer from this disease.

Which cross could **not** give rise to a child suffering from cystic fibrosis?

- A** FF x ff **B** Ff x Ff **C** Ff x ff **D** ff x ff

13 What is an ecosystem?

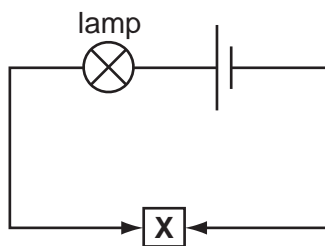
- A** a community and its habitat
B a group of organisms and their predators
C all organisms in a food chain
D where an organism lives and breeds

14 What do the chemical symbols N₂ and Ni represent?

	N ₂	Ni
A	a compound	a compound
B	a compound	an element
C	an element	a compound
D	an element	an element

15 The diagram shows a circuit.

Solid **X** makes the lamp light.



What is solid **X**?

- A** copper
B rubber
C silicon(IV) oxide
D sulphur

16 Large hydrocarbons can beX..... to make smaller, more useful molecules.

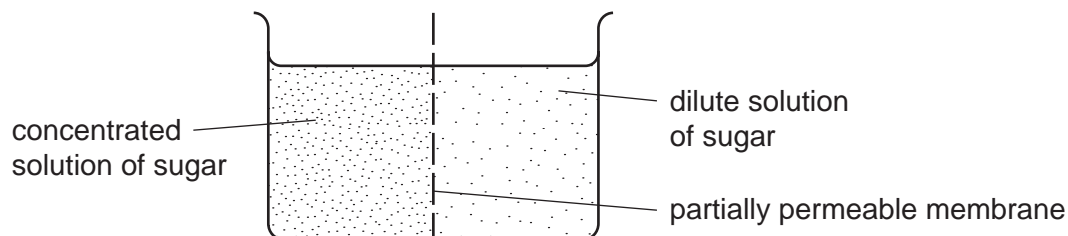
Small molecules can beY..... to make long molecules.

What are **X** and **Y**?

	X	Y
A	cracked	distilled
B	cracked	polymerised
C	distilled	polymerised
D	distilled	cracked

17 A concentrated solution of a sugar is separated from a dilute solution of this sugar by a partially permeable membrane.

Sugar molecules are bigger than water molecules.



After one hour, the concentration of each solution has changed.

The reason is that more1..... molecules pass to the2..... than to the3.....

Which words correctly fill gaps 1, 2 and 3?

	1	2	3
A	sugar	left	right
B	sugar	right	left
C	water	left	right
D	water	right	left

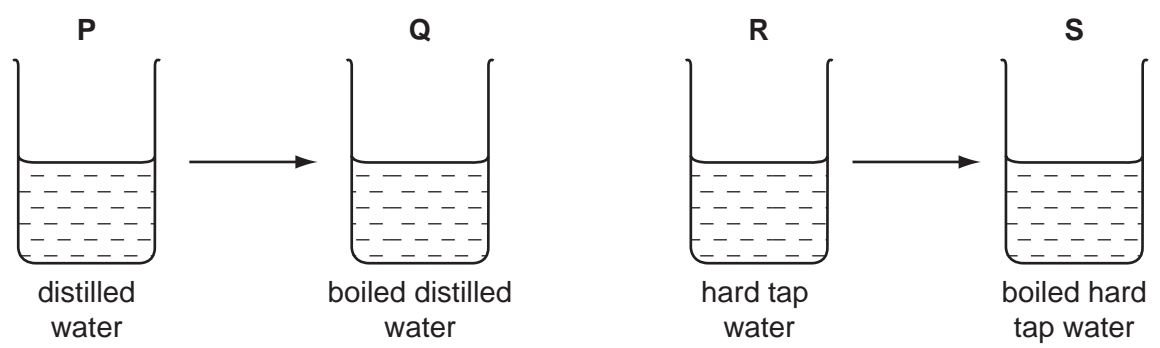
18 Carbon is used in the extraction of some metals from their ores because

- 1 carbon forms strong alloys with metals,
- 2 carbon reacts with oxygen in the ore.

Which of these statements are correct?

- A 1 only
- B 2 only
- C both 1 and 2
- D neither 1 or 2

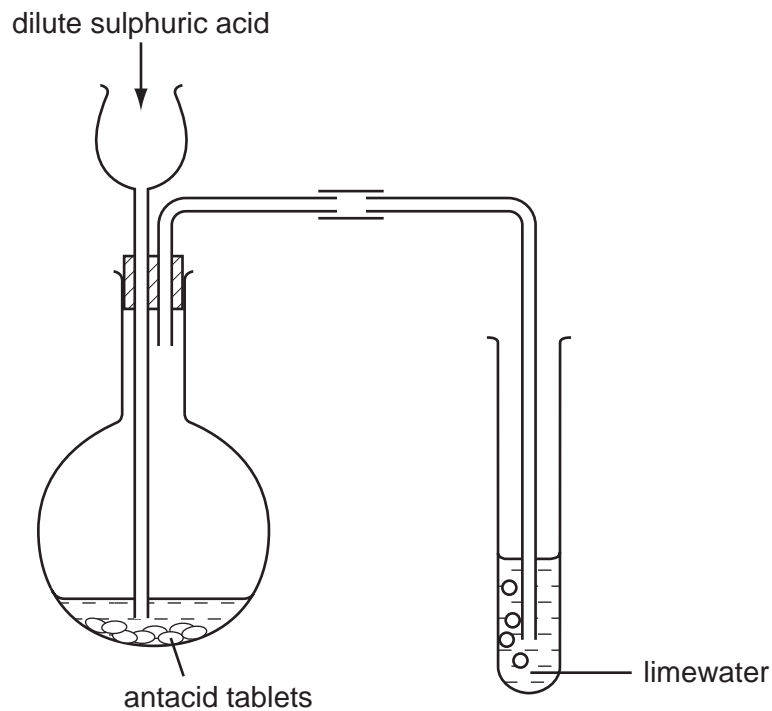
19 Soap solution is gradually added to separate samples of water **P**, **Q**, **R** and **S** until a lather forms.



How does boiling affect the volume of soap solution needed for a lather?

	P to Q	R to S
A	no change	no change
B	no change	S needs less
C	Q needs more	no change
D	Q needs more	S needs less

20 Dilute sulphuric acid is added to antacid tablets in the apparatus shown.

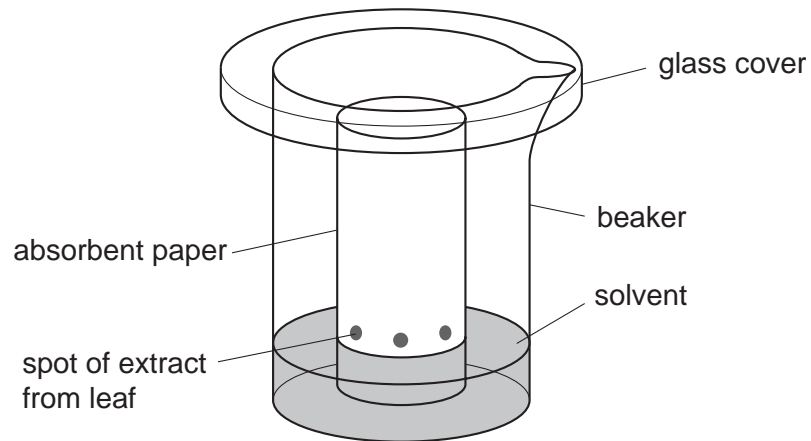


The limewater turns milky.

What do the antacid tablets contain?

- A magnesium
 - B magnesium carbonate
 - C magnesium hydroxide
 - D magnesium oxide
- 21 Which unit of time is most useful in describing the ages of rocks?
- A tens of years
 - B hundreds of years
 - C thousands of years
 - D millions of years
- 22 An increase in the world's population increases the demand for food.
- Which industrial process helps to increase food production?
- A chlorination of water
 - B distillation of petroleum to form petrol
 - C manufacture of ammonium sulphate
 - D recycling of glass bottles

23 A student uses the apparatus shown to find out how many different pigments are in leaf



What is this separation method called?

- A chromatography
- B distillation
- C evaporation
- D filtration

24 The contents of a beaker scatter a beam of light

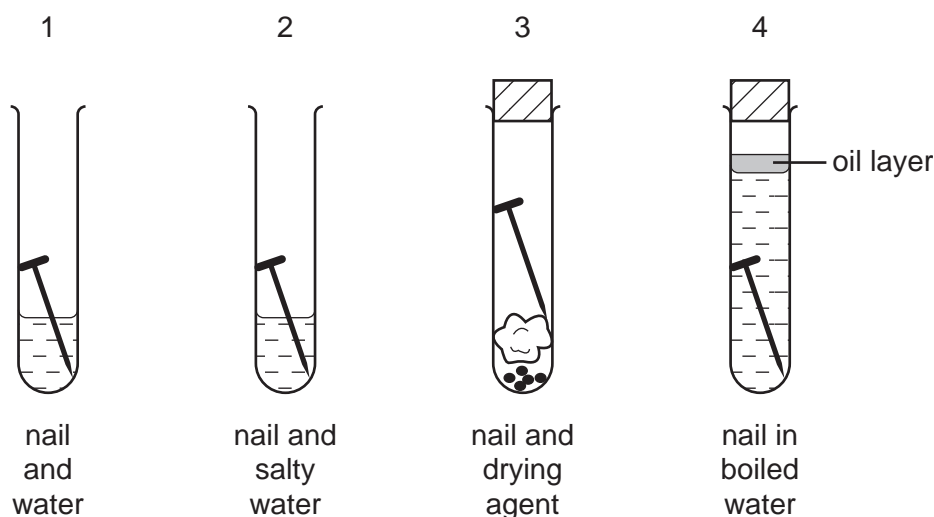
What does the beaker contain?

- A aqueous copper(II) sulphate
- B ethanol
- C milk
- D water

25 Which of the following is a solid fossil fuel?

- A coal
- B oil
- C sugar
- D wood

26 The diagrams show an investigation into the conditions needed for rusting of iron nails.



The nails in tubes 1 and 2 rust within a few days.

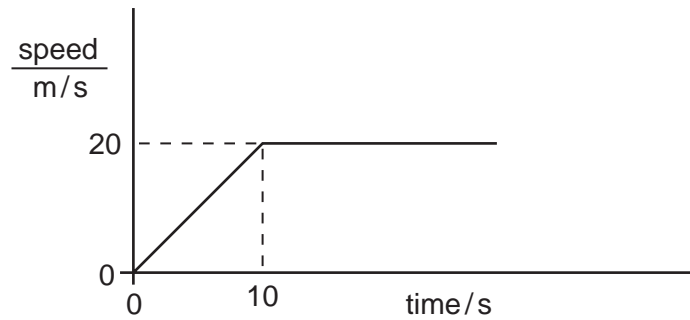
Which conditions are required for rusting?

- A air alone
 - B air and water
 - C salt and water
 - D water alone
- 27 Which ion gives a white precipitate **both** with aqueous sodium hydroxide **and** with aqueous ammonia?
- A $\text{Cu}^{2+}(\text{aq})$
 - B $\text{Fe}^{2+}(\text{aq})$
 - C $\text{Fe}^{3+}(\text{aq})$
 - D $\text{Zn}^{2+}(\text{aq})$
- 28 A decorator wishes to calculate the area of a bathroom tile so that he can estimate the amount of adhesive which he needs to buy.

What must he use?

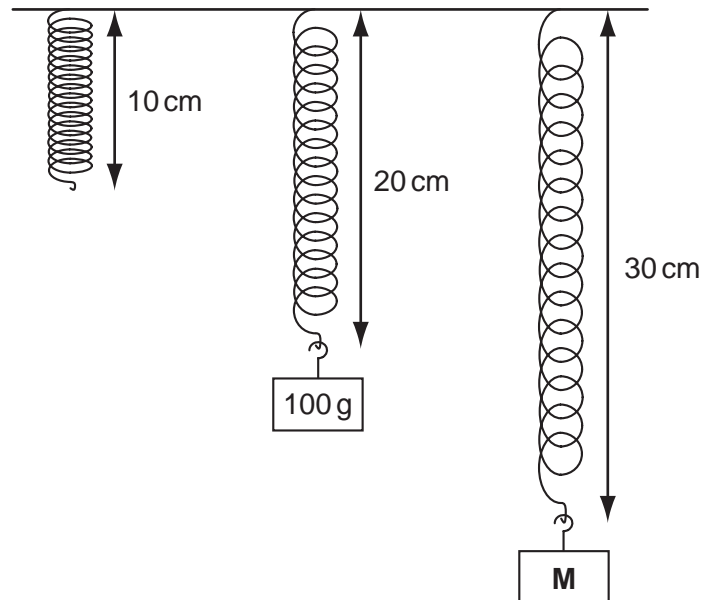
- A a measuring cylinder only
- B a ruler only
- C a measuring cylinder and a clock only
- D a measuring cylinder and a ruler only

- 29 A car accelerates from traffic lights. The graph shows how the car's speed changes with time.



How far does the car travel before it reaches a steady speed?

- A 10 m B 20 m C 100 m D 200 m
- 30 Objects with different masses are hung on a 10 cm spring. The diagram shows how much the spring stretches.



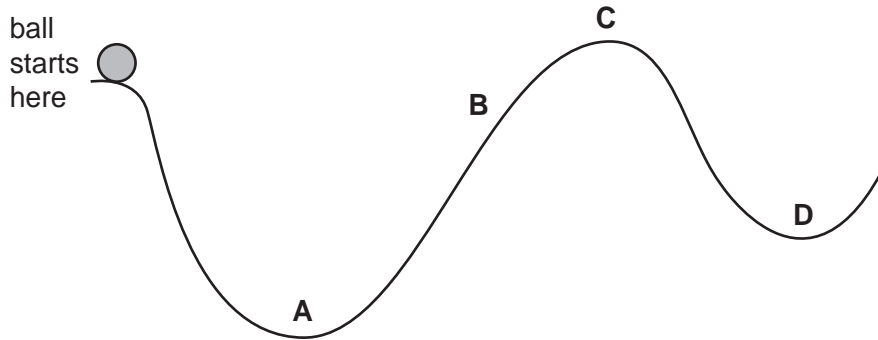
The extension of the spring is directly proportional to the mass hung on it.

What is the mass of object **M**?

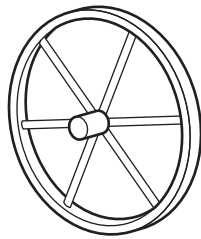
- A 110 g B 150 g C 200 g D 300 g

31 A ball is released from rest and rolls down a track from the position shown.

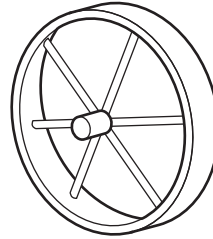
What is the furthest position the ball could reach?



32 A farmer has two carts. The carts have the same weight, but one has narrow wheels and the other has wide wheels.



narrow wheel

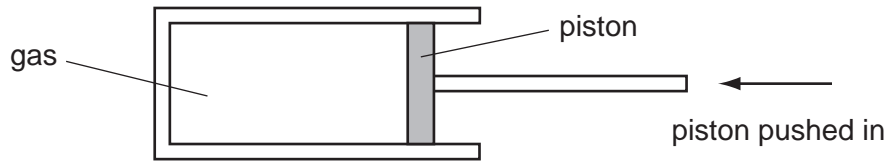


wide wheel

In rainy weather, which cart sinks **less** into soft ground, and why?

	cart wheels	why
A	narrow	greater pressure on the ground
B	narrow	less pressure on the ground
C	wide	greater pressure on the ground
D	wide	less pressure on the ground

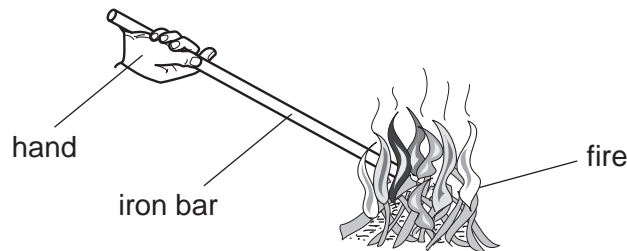
- 33 A measured mass of gas is placed in a cylinder at atmospheric pressure and is compressed.



The temperature of the gas does not change.

What happens to the pressure of the gas?

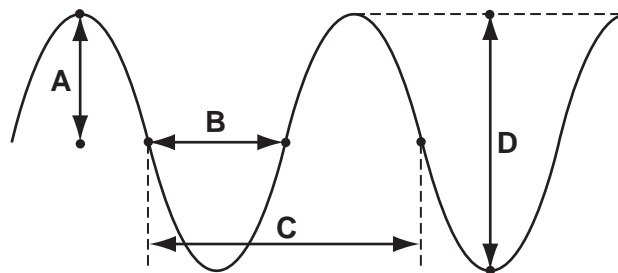
- A It drops to zero.
 - B It decreases, but not to zero.
 - C It stays the same.
 - D It increases.
- 34 An iron bar is held with one end in a fire. The other end soon becomes too hot to hold.



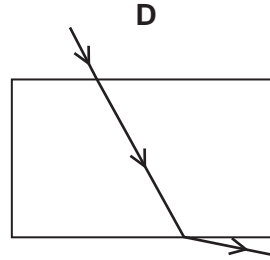
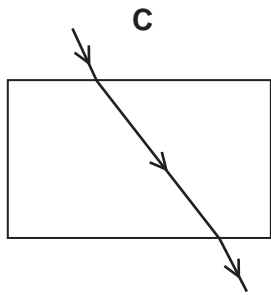
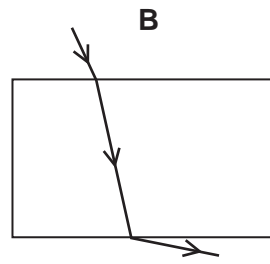
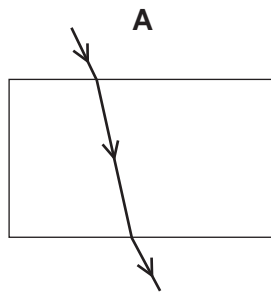
How has the heat travelled along the iron bar?

- A by conduction
 - B by convection
 - C by expansion
 - D by radiation
- 35 The drawing shows a wave.

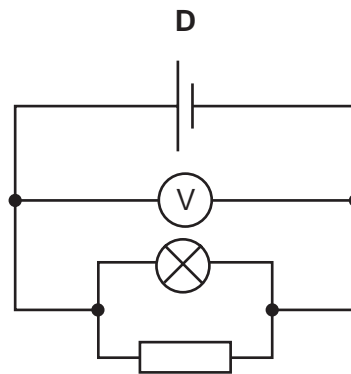
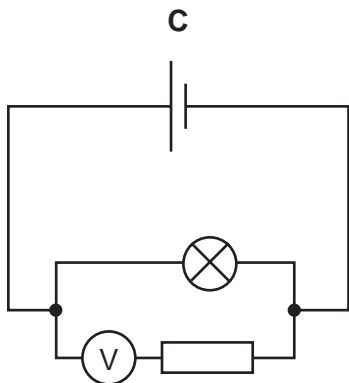
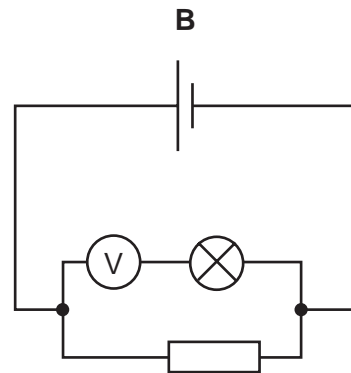
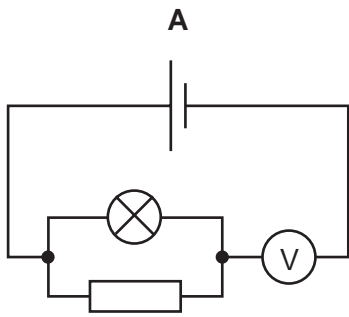
Which labelled distance is the wavelength?



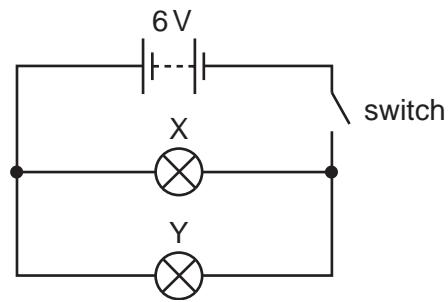
36 Which diagram correctly shows a ray of light passing through a rectangular glass block?



37 In which circuit does the voltmeter read the potential difference across the lamp?



38 In the circuit below, X and Y are identical 6 V lamps.



What happens when the switch is closed?

- A X lights more brightly than Y.
 - B Y lights more brightly than X.
 - C X and Y light with equal brightness.
 - D Neither X nor Y light.
- 39 Which type of radiation produces the most ionisation?
- A alpha-particles
 - B beta-particles
 - C gamma-rays
 - D all produce the same amount
- 40 A powder contains 400 mg of a radioactive material which emits alpha-particles.
- The half-life of the material is 5 days.
- What mass of that material remains after 10 days?
- A 0 mg
 - B 40 mg
 - C 100 mg
 - D 200 mg

DATA SHEET
The Periodic Table of the Elements

		Group															
I	II	III	IV	V	VI	VII	O										
1 H Hydrogen											2 He Helium						
3 Li Lithium	4 Be Beryllium											10 Ne Neon					
11 Na Sodium	12 Mg Magnesium	13 Al Aluminium	14 Si Silicon	15 P Phosphorus	16 S Sulphur	17 Cl Chlorine	18 Ar Argon										
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon
55 Cs Caesium	56 Ba Barium	57 La Lanthanum	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon
87 Fr Francium	88 Ra Radium	89 Ac Actinium															

140 Ce Cerium	141 Pr Praseodymium	144 Nd Neodymium	150 Sm Samarium	152 Eu Europium	157 Gd Gadolinium	162 Dy Dysprosium	165 Ho Holmium	167 Er Erbium	169 Tm Thulium	173 Yb Ytterbium	175 Lu Lutetium
90 Th Thorium	91 Pa Protactinium	92 U Uranium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium

*58-71 Lanthanoid series
90-103 Actinoid series

Key

a	X
b	

 a = relative atomic mass
 X = atomic symbol
 b = proton (atomic) number

The volume of one mole of any gas is 24 dm³ at room temperature and pressure (r.t.p.).