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## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

MARK SCHEME for the October/November 2011 question paper for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/63

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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[1]

[1]

[1]

[3]

[1]

[1]

[Total: 10]

Syllabus

	9 -				O.
			IGCSE – October/November 2011	0654	200
1	(a) (i)	expa	nata/pores ; anding gas , oxygen/CO <sub>2</sub> ;		Mac ambride
	(ii)		between 42 and 45 ; between 20 and 24 ;		[2]
	(iii)	total	no. of squares for <b>C</b> e.g. 42 multiplied by 100; no. of squares for <b>P</b> e.g. 20 multiplied by 100;		[2]
	(iv)	in di more less more	mer; rect sun; e wind movement; humid; e water loss; e wilting;		[max 2]
	( <b>b)</b> all	bundle em ;		[2]	
					[Total: 10]
2	(a) (i)	gree to ye	en ; ellow/orange ;		[2]

Mark Scheme: Teachers' version

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(ii) carbonic acid; (allow H<sub>2</sub>CO<sub>3</sub>)

(iii) (aq) = aqueous/dissolved; (g) = gas/gaseous;

(s) = solid;

(iv) precipitate;

(c) **B** and **C**;

(b) (i) turns white/white precipitate/milky/cloudy/owtte;

(ii) white/milkiness disappears/owtte (reject dissolves/reacts);

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3	(a) (i)		A ; V (± 0.1) ;	Syllabus 0654 places);		
	(ii)	2.0/ 2.3/	places) ; [2]			
	(b) (i)	all p	sible scales chosen, axes labelled; coints plotted ± small square (e.c.f.); coth curve drawn;	[3]		
	(ii)		re extended to show five wires; ut 0.5 ohms (value from candidate's graph);	[2]		
		(c) repeat (the experiment (using 1 wire – with different voltages and average (tresults));				
				[Total: 10]		
4	(a) (i)		C rate = 0.77/min ; C rate = 0.50/min ;	[2]		
	(b) (i)		ect plotting ; eptable smooth curve drawn ;	[2]		
	(ii)	50°0	C;	[1]		
	(iii) cannot tell exactly the rate		not tell exactly the rate either side of 50°C/owtte;	[1]		
	(c) (i)	(rate	e speeds up due to) particles moving faster/more co	llisions; [1]		
	(ii)	prote	ein denatures (due to high temperatures);	[1]		
	(d) tuk		to check if acid is needed for the reaction; to see if pepsin is needed/see if acid could do react	ion ; [2]		
				[Total: 10]		
5	(a) (i)	wate	er, ethanol, propanone or any suitable named organi	c solvent ; [1]		
	(ii)	horiz	zontal line drawn below the start line ;	[1]		
	(iii)	to pr	revent paper drying out/solvent evaporating/owtte;	[1]		
	(iv)	any	reasonable length of time, e.g. between 30 and 180	minutes; [1]		

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	(b)	(i)	one	are mixtures/impure; contains two dyes the other three; e one common dye;	Da Cambridge	
		(ii)	one	is pure one a mixture/only 3 pure ; contains three dyes the other one ; e no common dye ;	[any 1]	
	(c)	(c) named acid ; named alkali (either order) ;				
	cut spot from paper/use of spot ; add acid or alkali to spot ; look for colour change ;					
					[Total: 10]	
6	(a)			ark labelled <b>Y</b> ; ark labelled <b>Z</b> ;	[2]	
	(b)	line	es <b>YO</b>	and <b>ZO</b> drawn (e.c.f.) ; (ruler straight)	[1]	
	(c)	(i)	66 m	nm (or as candidate's diagram) ;	[1]	
		(ii)	63 m	nm (or as candidate's diagram) ;	[1]	
		(iii)	87 m	nm (or as candidate's diagram) all ± 1 mm ;	[1]	
	(d)	(i)	87/6	6 = 1.3 (e.c.f);	[1]	
		(ii)	87/6	3 = 1.4 (e.c.f);	[1]	
	(e)	(i)	•	ow) because the fish is deeper/further away than he sees it/light is bent y from the normal as it leaves the surface/owtte;	[1]	
		(ii)	his a	aim must be deeper than in fresh water, because the light is bent more/e;	[1]	
					[Total: 10]	