## **UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2012 question paper for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/21

Paper 2 (Core Theory), maximum raw mark 120

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2012 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

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- 1 (a) (speed) = distance/time; = 25/2 = 12.5 (km/h);
  - (b) (i) chemical;
    - (ii) heating engine/heating surroundings/light/sound;

(c) metal track expands in summer/hot weather; metal can expand into gap;

prevents track from buckling/bending/being damaged/becomes unsafe for trains to move;

(d) (i) ethanol is renewable source/no sulfur dioxide produced/does not deplete fossil fuel supplies;

(ii) carbon, hydrogen, oxygen; [1]

(e) (i) 5 (km/h); (accept 4/reference to greater than 3, but equal to or less than 5) [1]

(ii) 1200 (W); [1]

(iii) dependent on wind/wind speed unreliable or variable/owtte; [1]

(iv) kinetic; electrical; [2

[Total: 13]

- 2 (a) (i) nitrogen 78% oxygen 21% others 1%;; [2] (all three correct 2 marks, two correct 1 mark)
  - (ii) carbon dioxide/water (vapour); [1
  - (b) (i) contain only one type of atom/all atoms have same proton number; [1]
    - (ii) covalent; bonded elements are both non-metals/compounds are gases;
    - (iii) pure water is neutral; rain water (generally) is more acidic (than pure water); rainwater during thunderstorm is the most acidic; likely to be caused by nitrogen oxides; which have reacted with the rain to form an acid;

[Total: 9]

[max 3]

[1]

[1]

[max 2]

[2]

[1]

Page 3	Mark Scheme: Teachers' version	Syllabus	ľ
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- 3 (a) label to root hair cell;
  - (b) water; minerals/ions/named mineral ion;

(c) (i) xylem; [1]

(ii) A in central area of root; [1]

(iii)

structure	plant cells	animal cells
cell membrane	✓	✓
cell wall	✓	×
nucleus	✓	✓
vacuole containing sap	✓	×

one mark for any row correct ;;;; [4]

[Total: 9]

- **4** (a) (i) between 10 and 20 Hz to between 20 000 and 25 000 Hz; [1]
  - (ii) frequency –
    number of waves produced/passing a point per second;
    wavelength –
    distance between two consecutive peaks/troughs on two consecutive waves; [2]
  - (b) (i) sound longitudinal; light transverse; [2]
    - (ii) no medium/material for waves to travel through; [1]
    - (iii) quieter; [1]
    - (iv) microwaves; [1]

[Total: 8]

	Page 4		Mark Scheme: Teachers' version	Syllabus	V.
	-		IGCSE – May/June 2012	0654	2
5	only plai		nal that) eats plants ; nts/not meat ; e to getting energy from, its food/plants ;		Cambridge.
	(b) (i)	eat/	a lot ; take in, more energy than they use ; ess, carbohydrate/protein, converted to fat ;		[max 2]
	(ii)	idea mas	greater the body mass, the greater the chance of su that effect is greater at lower body masses/leve ses;		. 01
		use	of figures ;		[max 2]
	(iii)	insu	lator/poor conductor/reduces conduction;		[1]
	(c) car	bon d thane			[2]
	1110	uiuio	' 1		[~]
	(d) (i)	mea	n body mass is increasing;		[1]
	(ii)		mots have more time to feed (from spring onwards)		
			mots lose less weight during hibernation (as winters rence to more food available ;	are shorter) ;	[max 1]
		10101	is the terminal to the terminal to the terminal		[max 1]
				[	Γotal: 11]
6		d mag ert bu	nesium to acid ;		
	me	asure	e time for known volume of gas to collect ; e volume of gas at fixed intervals ;		[max 3]
	bec	cause	(same volume) of gas to collect is shorter; rate of reaction is greater/greater collision frequencies;	ency/faster reaction/	[2]
	(c) (i)	(ato	m) loses electrons/achieves complete outer shell ;		[1]
	(ii)	MgC	$\mathcal{C}l_2$ ;		[1]
	(d) (i)	(pov	crease reaction/burning rate ; vder has) large surface area ; rate needed to get all light at once/so brighter ;		[max 2]
	(ii)		/low density; uces weight of aircraft/less energy or fuel needed fo	r flight ;	

strong ; reference to maintenance of the integrity of airframe/safety ;  $\,$ 

[Total: 13]

[4]

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			S

7 (a) use – cancer treatment/tracers/kill bacteria/sterilising surgical equipment; correct description;

(b) gamma alpha gamma alpha

(2 marks for four correct, 1 mark for two or three correct);;

[2]

[1]

(c) destroys/damages cells; caused cancer; mutations/damages DNA;

[max 2]

[Total: 6]

- 8 (a) (i) nucleus;
  - /::\ 10 ·
  - (ii) 18;
  - (iii) (no) no mark reference to isotopes; which are atoms of same element with different number of neutrons; [2]
  - (iv) hydrogen; [1]
  - (b) (i) any typical metal property for **X** and corresponding non-metal for **Y**; [1] e.g. **X** conductor **Y** insulator
    - X malleable Y not malleable
      - X forms positive ions Y forms negative ions
      - X high fixed points Y low fixed points
    - (ii) Y is more reactive/Z is less reactive/unreactive/specific example; [1]
  - (c) (i) combustion of carbon; (thermal) decomposition/breakdown of calcium carbonate; [2]
    - (ii) soil is acidic; lime reacts with acid/reduces acidity/neutralises; to increase fertility; [max 2]

[Total: 11]

[Total: 10]

Syllabus

		J		IGCSE – May/June 2012	0654	StaCambridge.
9	(a)	nut	rition			DH.
	` '		cretion	ı		Ori
			piratio			Se .
			roduc	tion		i,
		_	wth	1		
			veme mark f	or any two correct) ;;		[2]
	(b)	pro		by a gland ;		
				y the blood;		
				specific) target organs ; d by the liver ;		[max 3]
		uce	stroyc	a by the liver,		[max o]
	(c)			piration ; ergy (for muscles) ;		
				can work, harder/faster;		[max 2]
	(d)			phototropism ;		ro1
		gro	ws to	wards the light ;		[2]
						[Total: 9]
10	(a)	(i)		neter in series ;		
				neter in parallel ;		ro1
			all e	se correct;		[3]
		(ii)	to ch	nange voltage across, current through lamp ;		[1]
		(iii)	R =	V/I:		
		()		$0.3 = 10 \; (\Omega) \; ;$		[2]
	(b)	(i)	D it's	s longer (resistance proportional to length) ;		[1]
		/i:\	۸ ۵۳	naller cross-sectional area ;		
		(ii)	A 511	ialici Gios-sectional alea ,		[1]
	(c)	(i)	posi	tive and negative ;		[1]
		(ii)	elec	tron ;		[1]

Mark Scheme: Teachers' version

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[2]

[Total: 9]

	Page 7		Mark Scheme: Teachers' version	Syllabus
			IGCSE – May/June 2012	0654
11	(a)	B t	larynx ; trachea ; bronchus/bronchiole ;	Syllabus 7.70 r 0654 r 0654
	(b)	less	re carbon dioxide ; s oxygen ; re water vapour ;	[max 2]
	(c)	(i)	diffusion;	[1]
		(ii)	pulmonary vein ;	[1]
	(d)	(i)	1 in 4/one quarter/0.25 ;	[1]
		(ii)	(parents' genotypes) both Ff; gametes F and f from both parents; offspring genotypes FF, Ff, Ff and ff; ff identified as having cystic fibrosis;	[4] [Total: 12]
12	(a)	peti	roleum/crude oil;	[1]
	(b)		goes cloudy/milky/opaque/white; carbon dioxide;	[2]
		(ii)	water;	[1]
	(	iii)	(no) – no mark ethanol's combustion products are the same as methan	ne's ; [1]
	(c)	wat	ter/steam ;	[1]
	(d)	(i)	polymerisation/addition/self addition;	[1]

(ii) link together;

into long chains/into chain molecules;