



UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

CANDIDATE NAME				
CENTRE NUMBER		CANDIDATE NUMBER		

CO-ORDINATED SCIENCES

0654/23

Paper 2 (Core)

October/November 2013

2 hours

Candidates answer on the Question Paper.

No Additional Materials are required.

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a pencil for any diagrams or graphs.

Do not use staples, paper clips, glue or correction fluid.

DO **NOT** WRITE IN ANY BARCODES.

Answer all questions.

Electronic calculators may be used.

You may lose marks if you do not show your working or if you do not use appropriate units.

A copy of the Periodic Table is printed on page 28.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

UNIVERSITY of CAMBRIDGE **International Examinations**

1 Fig. 1.1 shows a root hair cell.

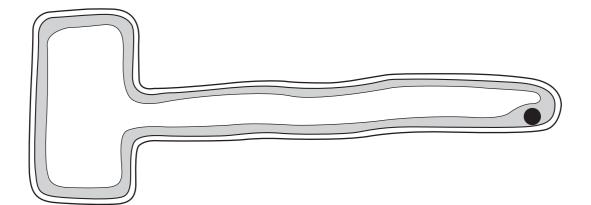


Fig. 1.1

- (a) Use the letters A, B and C to label these parts of the root hair cell in Fig. 1.1.
 - A the cell membrane
 - B the part that contains chromosomes
 - **C** a structure that is **not** present in animal cells [3]
- **(b)** Name **two** substances that are absorbed by root hair cells.

1	

2 ______[2]

(c) Fig. 1.2 shows part of a plant stem from which the outer layer, including the phloem, has been removed.

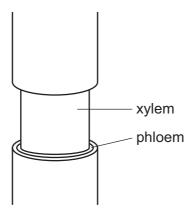


Fig. 1.2

(i)	State the function of phloem.
	[2]
(ii)	Suggest why this treatment would cause the roots of the plant to die.
	[2]

2 (a) Table 2.1 shows information about some chemical elements and their positions in the Periodic Table.

For Examiner's Use

Table 2.1

	<u> </u>
element	group number in the Periodic Table
oxygen	6
calcium	2
lithium	1
sulfur	6
fluorine	7

(i)	State the noble (inert) gas that is in the same period of the Periodic Table as sulfur.
	[1]
(ii)	Select two elements from Table 2.1 whose atoms form ionic chemical bonds with each other and explain your answer.
	and
	explanation
	[2]

(b) Fig. 2.1 shows a diagram of an atom.

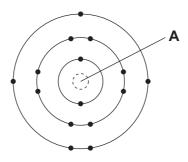


Fig. 2.1

(i) Name structure **A** in Fig. 2.1. [1]

(ii) State the proton number of the atom in Fig. 2.1.	For
Explain your answer briefly.	Examiner's Use
proton number	
explanation	
[2]	
(c) A student added excess acidified barium chloride solution to a solution of a magnesium compound to produce mixture W .	
Fig. 2.2 shows the procedure followed.	
acidified barium chloride solution mixture W solution of a magnesium compound white precipitate settled on the bottom of the beaker	
Fig. 2.2	
(i) Suggest the full name of the magnesium compound in the original solution.	
[1]	
(ii) Describe briefly what the student should do to find the mass of the white precipitate in mixture W.	
[3]	
[e]	

3 (a) Fig. 3.1 shows a circuit used to measure the current passing through a resistor when the voltage across it is changed.

For Examiner's Use

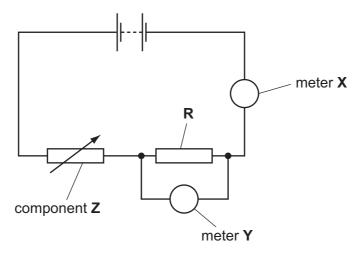


Fig. 3.1

		
(i)	Describe the purpose of component Z in the circuit.	
		[1]
(ii)	The meters shown in the circuit give readings of 0.6A and 8.0V.	
	State which meter, X or Y , gives the reading of 0.6 A.	
	Explain your answer.	
	meter	
	explanation	
		[1]
(iii)	Calculate the resistance of resistor R .	
	State the formula that you use and show your working.	
	formula	
	working	
	Ω	[2]
	72	r—1

(b)	Complete the sentences phrase can be used once	•	•	e from the list. Each word o	r
	decreases	increases	is zero	stays the same	
	When the voltage across	the resistor is red	uced, the curr	ent through the resistor	
	When the voltage of the s	upply is reduced,	the voltage a	cross the resistor	
			•		
	When the voltage across	the resistor is red	uced, the resi	stance of the wire	
			·		
				[2]	[]
(c)	The resistance of a piece temperature of the wire an	-		nber of variables such as the made.	Э
	State two other factors w	hich affect the re	sistance of a p	piece of wire.	
	1				
	2			[2]

4

	va beans are an important crop in Brazil. Soya beans contain a lot of protein, plus aller quantities of starch and fat.
(a)	Describe how you could test a sample of soya beans to find out if they contain fat.
	[3]
	[الا
(b)	Explain why protein is an important part of a balanced diet.
	[2]
	[2]
(c)	When a person eats soya beans, the beans are chewed in the mouth.
	Explain why this makes it easier for enzymes in the digestive system to digest the beans.
	[2]
(d)	Raw soya beans contain substances that stop protease enzymes from working. Cooking destroys these substances.
	Suggest how eating uncooked soya beans could prevent the absorption of some of the nutrients from them.
	[2]

(e)	Large areas of rainforest have been cleared in Brazil, to provide more land for growing soya beans.
	Explain how cutting down the rainforest can harm the environment.
	[4]

10 5 (a) A student placed four equally-sized pieces of different metals into colourless liquids contained in four test-tubes P, Q, R and S. Fig. 5.1 shows what the student observed. Ρ R S gas produced gas produced no reaction no gas produced slowly quickly orange layer forms on metal surface after several days Fig. 5.1 (i) Suggest which of the test-tubes in Fig. 5.1 contained water to which a piece of iron was added. Explain your answer. test-tube explanation

(ii) The colourless liquid in test-tube **R** was dilute hydrochloric acid.

Suggest the name of the metal that was added to test-tube ${\bf R}$ and name the gas that was produced.

[3]

metal

gas [2]

(iii) Test-tube **P** contained the same concentration of dilute hydrochloric acid at the same temperature as test-tube **R**.

Suggest the name of the metal that was added to test-tube **P**.

[1]

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(b) In the process of copper plating, a thin layer of copper is formed on the surface of a metal object.

For Examiner's Use

Fig. 5.2 shows the apparatus and materials that are needed to copper plate a metal key.

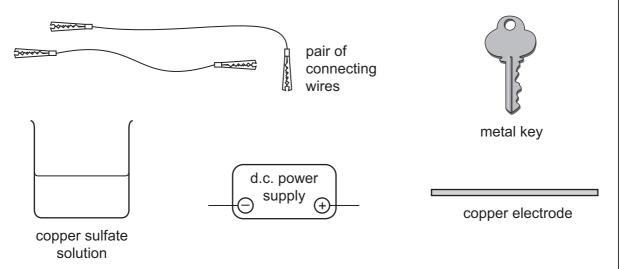


Fig. 5.2

Draw a diagram which shows how the apparatus and materials in Fig. 5.2 should be assembled so that the metal key will be copper plated.

[3]

6 (a) Fig. 6.1 gives information about the uses of different types of electromagnetic waves and their effects on living tissue.

For Examiner's Use

Draw lines to link each electromagnetic wave with its effect on living tissue and its use. One has been completed as an example.

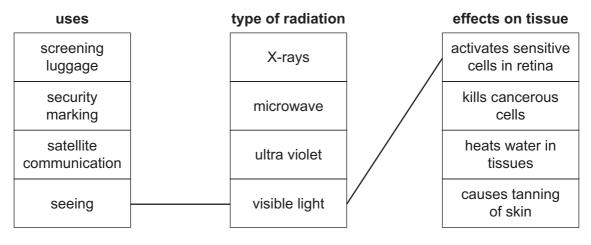


Fig. 6.1

[4]

(b) Electromagnetic waves are transverse waves. Water waves are also transverse.

Draw a diagram of a transverse wave on the axes below. Label the amplitude and **one** wavelength on your diagram.

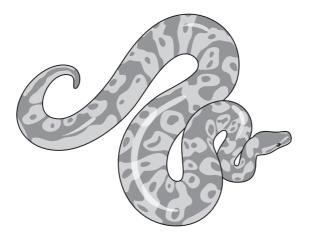


[3]

Please turn over for Question 7.

7 Ball pythons (royal pythons) are snakes that are kept as pets in many parts of the world.

For Examiner's Use



The colour of a ball python is determined by its genes.

Some ball pythons are albino (white). This is caused by a recessive allele, ${\bf a}$. The dominant allele, ${\bf A}$, gives normal colouring.

(a) Complete Table 7.1 to show the possible genotypes and colours arising from this gene.

Table 7.1

genotype	colour
AA	
Aa	normal
	albino

[2]

(b)	State the correct biological term for the visible appearance produced by the genotype
	in this case the colour of the snake.

[1]

(c)	(i)	Complete the genetic diagram to explain the results of crossing two snakes that are heterozygous for these alleles.	For Examiner's Use
		genotype of parents Aa and	
		gametes and and	
		gametes from one parent	
		gametes from the other parent	
		[3]	
	(ii)	State the ratio of offspring that you would expect from this cross.	
		ratio of normal : albino offspring = : [1]	
(d)	A b	reeder has several snakes with normal colouring.	
(,		To the control of the	
		ggest how she can find out whether a particular snake is homozygous or erozygous.	
		[2]	

8 (a) Fig. 8.1 shows apparatus a student used to investigate the reaction between dilute nitric acid and excess calcium carbonate.

For Examiner's Use

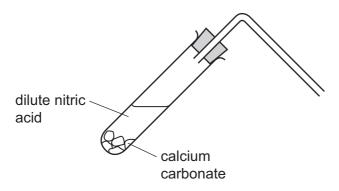


Fig. 8.1

	119. 0.1
(i)	Name the gas that is given off in this reaction.
	[1]
(ii)	Describe how the student could test for the gas you named in (i). You may wish to complete the diagram in Fig. 8.1 to help you to answer this question.
	[2]
(iii)	At the end of the reaction the test-tube in Fig. 8.1 contains a solution of the compound calcium nitrate.
	State the general name for compounds like calcium nitrate which are produced when an acid reacts with a metal carbonate.
	[1]
(iv)	The chemical formula of calcium nitrate is Ca(NO ₃) ₂ .
	State the total number of atoms and the number of different elements that are shown combined together in this formula.
	total number of atoms
	number of different elements [2]

(b) The student then carried out an investigation into the way that the rate of the reaction in (a) changed when he varied the concentration of the nitric acid.

For Examiner's Use

Fig. 8.2 shows the apparatus the student used to measure the rate of reaction.

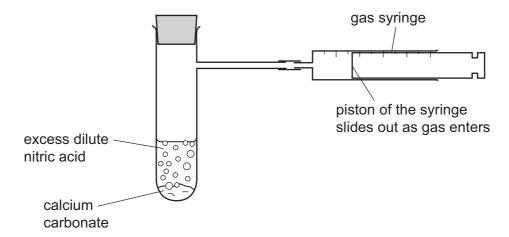


Fig. 8.2

The student measured the rate of reaction by finding how long it took for the gas syringe to fill with gas.

(i) After he had completed several measurements, the student wrote the following correct conclusion in his notebook.

Conclusion
The higher the pH of the dilute nitric acid
the longer it took for the gas syringe to
fill with gas.

	Explain this conclusion briefly.
	[2]
(ii)	State two other variables that can affect the rate of reaction between dilute nitric acid and calcium carbonate.
	1
	2

9 Fig. 9.1 shows a solar-powered golf cart used to carry golfers around a golf course.



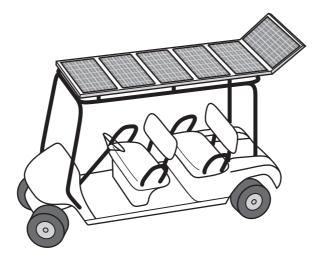


Fig. 9.1

(a) As the cart moves around the course, the motion of the cart is measured.

Fig. 9.2 shows a distance/time graph for a small part of the journey lasting 60 seconds.

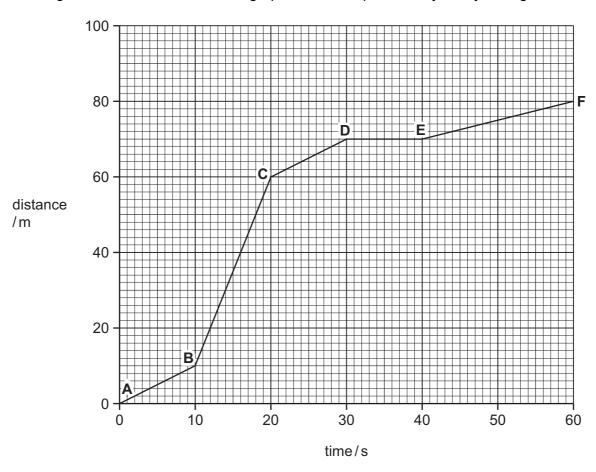


Fig. 9.2

(i) Write down the total distance covered in 60 s. m [1]

	(ii)	Calculate the speed of the cart between B and C .	For Examiner's
		Show your working.	Use
		m/o [1]	
	/III\	m/s [1]	
	(iii)	Describe the motion of the cart between D and E .	
		[1]	
	(iv)	During another part of the journey, the cart is accelerating.	
		State whether the forces acting on the cart are balanced or unbalanced.	
		Explain your answer.	
		[1]	
(b)		e cart is powered by solar cells on its roof. The solar cells produce electrical energy and to charge the rechargeable batteries in the cart.	
	Nar	me one other renewable energy resource that could produce electrical energy.	
		[1]	
(c)	The	e golfer hits a golf ball with his club. The ball flies through the air.	
	(i)	State the form of energy given to the golf ball when the ball is hit.	
		[1]	
	(ii)	State the form of energy gained by the golf ball as it rises into the air after being hit.	
		[1]	

(d)	The	e mass of a golf ball is 45 g. The volume of a golf ball is 36 cm ³ .		
	Cal	culate the density of the golf ball.		
	State the formula that you use and show your working.			
		formula		
		working		
		g/cm ³ [2]		
(e)	(i)	The head of the golf club is made of solid metal. The air that the golf ball is travelling through is a gas.		
		Complete Fig. 9.3 below to show the arrangement of particles in a gas. The diagram for a solid has been done for you.		
		solid gas		
		Fig. 9.3		
		[2]		
	(ii)	During the cart's journey, the temperature of the air in the tyres increases by 15 °C.		
		The volume of the air in the tyre remains the same.		
		Explain in terms of particles why the pressure of the air in the tyre increases when this happens.		
		[1]		

(iii)	Sometimes the golfer's hands begin to sweat.
	Explain in terms of particles how sweating cools his hands.
	থে

10 Fig. 10.1 shows the contents of the human thorax (chest).



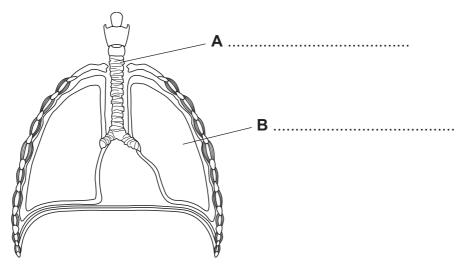


Fig. 10.1

(a) On F	⁻ig. ′	10.1,	name	struc	tures 🖊	A and	В
----	--------	--------	-------	------	-------	---------	--------------	---

(i) Define the term diffusion.

[2]

(b) Oxygen diffuses into the blood from the alveoli inside the lungs. Carbon dioxide diffuses into the alveoli from the blood.

	[2]
(ii)	Name the component of blood that transports dissolved carbon dioxide.
	[1]
(iii)	When a person is doing vigorous exercise, the concentration of carbon dioxide in the blood increases.
	Explain why this happens.
	ro

(iv)	Suggest how this will affect the rate of diffusion of carbon dioxide from the blood to the alveoli.
	Explain your answer.
	effect on rate of diffusion
	explanation
	[2]

11	Pet	role	um (crude oil) is a liquid fossil fuel.	For Examiner's
	(a)	Nar	me one solid fossil fuel. [1]	Use
	(b)	Gas	soline and diesel are mixtures of liquid hydrocarbons obtained from petroleum.	
		(i)	Name the process used to separate gasoline and diesel from petroleum.	
			[1]	
		(ii)	State the main use of gasoline and explain, in terms of its chemical properties, why it is suitable for this use.	
			use	
			explanation	
			[2]	
	(c)		cural gas is a gaseous fossil fuel, which contains mainly methane mixed with other appounds such as ethane.	
		(i)	Complete the diagram of the structure of one molecule of ethane.	
			—c	
			[2]	
		(ii)	Complete the word chemical equation for the complete combustion of ethane.	
	et	hane	e +	
			[2]	
	(d)	Eth	ene, C ₂ H ₄ , is an unsaturated hydrocarbon.	
			ene is manufactured by heating large hydrocarbon molecules in the presence of a alyst. During this process no air must be allowed into the reaction vessel.	
		(i)	Name the process used to manufacture ethene. [1]	

(ii)	Suggest one reason why air must be kept out of the reaction vessel.
	[2]

12 (a) Fig. 12.1 shows a light ray entering an optical fibre.

For Examiner's Use

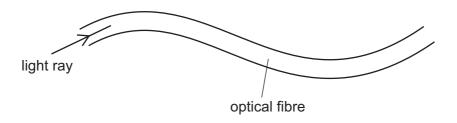


Fig. 12.1

The light ray travels all the way through the optical fibre.

Explain why the light ray is able to stay inside the optical fibre.

You may draw on the diagram if it helps your answer.

[2]

(b) White light is passed through a prism as shown in Fig. 12.2.

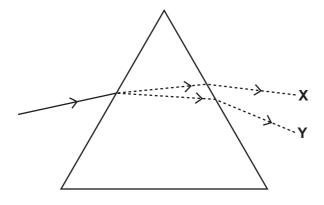


Fig. 12.2

(i) State the colours seen at positions X and Y.

X	
Υ	[2]

(ii) A rainbow is formed in a similar way. Suggest what is acting as a prism when forming a rainbow.

[1]

(c) Fig. 12.3 shows a person looking into a mirror and seeing an image.

mirror

For Examiner's Use



Fig. 12.3

- (i) Write the letter **X** on Fig. 12.3 to show the position of the image of the person's nose. [2]
- (ii) Select three words or phrases from the list that describe the image correctly.

larger than object	ct real	same size as o	object
smaller than object	upright	upside down	virtual
			[3]

DATA SHEET
The Periodic Table of the Elements

	0	4 Heium	20 Neon 10 At Argon	84 Kr Krypton 36	131 Xe Xenon 54	Radon 86		175 Lu Lutetium 71	Lr Lawrencium 103
Group	II/		19 Fluorine 9 35.5 C 1	80 Br Bromine 35	127 I lodine 53	At Astatine 85		173 Yb Ytterbium 70	Nobelium 102
			16 O 8 0xygen 32 S Sulfur 16	79 Se Selenium 34	128 Te Tellurium 52	Po Polonium 84		169 Tm Thulium 69	Md Mendelevium 101
	>		14 Nitrogen 7 31 P Phosphorus 15	75 AS Arsenic 33	122 Sb Antimony 51	209 Bi Bismuth		167 Er Erbium 68	Fm Fermium
	>		12 Carbon 6 28 Si Silicon	73 Ge Germanium 32	Sn Tin 50	207 Pb Lead 82		165 Ho Holmium 67	ES Einsteinium 99
	=		11 B Boron 5 A 1 A 1 A 1 A 1	70 Ga Gallium 31	115 In	204 T 1 T T T Thallium		162 Dy Dysprosium 66	Californium
				65 Zn Zinc 30	112 Cd Cadmium 48	201 Hg Mercury 80		159 Tb Terbium 65	BK Berkelium 97
				64 Cu Copper 29	108 Ag Silver 47	197 Au Gold 79		157 Gd Gadolinium 64	Cm Curium 96
				59 Nicke l Nickel 28	106 Pd Palladium 46	195 Pt Platinum 78		152 Eu Europium 63	Am Americium 95
Gre				59 Co Cobalt	103 Rh Rhodium 45	192 Ir Iridium		Sm Samarium 62	
		1 Hydrogen		56 Fe Iron	Ru Ruthenium 44	190 Os Osmium 76		Pm Promethium 61	Neptunium
				55 Mn Manganese 25	Tc Technetium 43	186 Re Rhenium 75		Neodymium 60	238 C Uranium
				52 Cr Chromium 24	96 Mo Molybdenum 42	184 W Tungsten 74		141 Pr Praseodymium 59	Pa Protactinium 91
				51 V Vanadium 23	93 Nb Niobium 41	181 Ta Tantalum 73		140 Ce Cerium 58	Th Thorium
				48 Ti tanium 22	2r Zramium 40	178 Hf Hafnium 72			nic mass bol nic) number
				Scandium 21	89 ×	139 La Lanthanum 57 *	227 Ac Actinium 89	l series eries	 a = relative atomic mass X = atomic symbol b = proton (atomic) number
	=		Berylium 4 24 Magnesium 12	40 Ca Calcium	Strontium	137 Ba Barium 56	226 Ra Radium	*58-71 Lanthanoid series 190-103 Actinoid series	<i>a</i> × <i>a</i>
	_		Lithium 3 Lithium 3 23 Na Sodium 11	39 K Potassium	Rb Rubidium 37	CS Caesium 55	Francium 87	*58-71 L	Key

The volume of one mole of any gas is 24 dm³ at room temperature and pressure (r.t.p.).

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