CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the May/June 2015 series

0654 CO-ORDINATED SCIENCES

0654/61 Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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Page 2	Mark Scheme	Syllabus	Paper	
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1 (a) (i) outline concave on one side and projections on the other; [2]

2 circles shaded and labelled;

(ii) xylem; [2]

transport of water;

(b)

test solution	observation	conclusion
Benedict's solution	orange	reducing sugar/glucose (present);
biuret solution	blue	protein absent ;
iodine solution	orange	starch absent ;

[3]

(c) Any 3 from 4 [3]

(celery in dyed water and) measure distance dye moves;

minimum 3 different temperatures;

time for coloured water to appear at top of (cut) stalk/set time and measure distance moved for each T;

all other conditions/named condition kept constant;

[Total: 10]

[1]

- **2** (a) 14 and 16;
 - **(b) (i)** 0.7(0) 0.8(0); [3]

0.49 and 0.64;

 T^2 to 2 d.p.;

Allow ecf

- (ii) 4 plots correct \pm 1/2 small square ; [2] best fit straight line through origin \pm 1/2 small square ;
- (iii) gradient shown clearly on graph (triangle at least 1/2 of graph); [2] 1.6;
- (iv) 39.5/gradient from (b)(iii) = 25; quoted to 2 sig figs; [2]

[Total: 10]

Page 3	Mark Scheme	Syllabus	Paper
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(a) (i)	blue/purple AND (pH between) 8 to 14;		[1
(ii)	calcium hydroxide/limewater;		[2
	calcium oxide ;		
(b) (i)	(sodium hydroxide) (light) blue ppt;		[3
	(ammonia) (light) blue ppt;		
	(ammonia) dark blue solution (in excess);		

(c) react with (e.g.) sulphuric acid; add sodium hydroxide (soln)/ammonia (soln); white ppt (dissolves in excess);

[Total: 10]

[1]

[3]

4 (a) (i) A white blood cell;
B red blood cell;
C platelet;

D plasma ;

(ii) CuO (not name);

(ii) 8; [1]

(iii) 0.008;; ecf

(b) (i)

activity	average pulse rate for 15 seconds	average heart rate (beats per minute)
resting	17	68
jogging	35	140

[1]

(ii) heart rate increases; [max 1]

increased or faster blood flow;

need more oxygen/respiration/removal of carbon dioxide;

(iii) average calculated/identify anomalies/confirms similar values/repeats; [1]

[Total: 10]

Page 4	Mark Scheme	Syllabus	Paper	
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5	(a)	use of cell/battery/power supply and connections;							[3]		
		connect in circuit;									
		(first two marks can be from a diagram)									
		lamp works if lamp lights ;									
	(b)	ammeter symbol correct and in series with lamp;								[3]	
	(6)	voltmeter sym									[0]
		-	DOI COITE	st and in	paranei	withiam	ρ,				
		circuit ;									
	(c)		T	1	1			7			[3]
	-	(lamp)	eg A	В	С	D	E				
	-	current/A									
		potential difference/V									
		table with head	dings (all	ow p.d.)							
		correct units (allow name or symbol); room for 5 lamps may be labelled with letters, numbers or not at all;									
	(d)	resistance = potential difference (voltage)/current;								[1]	
		[Total: 10]							otal: 10]		
6	(a)	hydrogen; lighted splint; pop (etc.);									[3]
	(b)	conical flask with delivery tube; (connected to) syringe or measuring cylinder over water;								[2]	
	(c)) (i) rate decreases ; (then) stops ;								[2]	
		(ii) Mg or acid or reactant(s) used up/all Mg or acid or reactant reacted;							[1]		
	(d)	line T to left of line T reaches									[2]
										[Te	otal: 10]