CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the October/November 2015 series

0654 CO-ORDINATED SCIENCES

0654/21 Paper 2 (Core Theory), maximum raw mark 120

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Р	age 2	Mark Scheme	Syllabus	Paper
		Cambridge IGCSE – October/November 2015	0654	21
1	(a) (i)	5;		[1]
	(ii)	neutron;		[1]
	(iii)	electron;		[1]
		creases ; cause elements change from metals to non-metals ;		[2]
	(c) (i)	covalent ;		[1]
	(ii)	it would be slower/it would not work;		[1]
	, n, , m			
	(d) (i)	nitrogen;		[1]
	(ii)	phosphorus and potassium ;		[1]
				[Total: 9]
2		= chloroplast ; = nucleus ;		[2]
	` '	rbon dioxide/CO ₂ ; ter ;		[2]
	(c) (o)	(xygen) from photosynthesis ; O_2) from respiration ;		[2]
	(d) (i)	transport (of water/minerals)/support;		[1]
	(ii)	dead/no chloroplasts ;		[1]
				[Total: 8]
3		int has small area and therefore high pressure therefore sinks in ; c/ski has large area therefore small pressure therefore doesn't sink	in ;	[2]
	(b) (i)	sound waves are reflected ;		[1]
	(ii)	166 (m);		[1]
	(iii)	(speed =) $\frac{\text{distance}}{\text{time}}$; = $\frac{166}{0.5}$ = 332 (m/s); (allow ecf from (ii))		[2]
		0.0		

Pa	age :	3	Mark Scheme	Syllabus	Paper
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	(c)	(i)	(B – no mark) particles are touching and randomly arranged ;		[1]
		(ii)	(A – no mark) particles are touching and regularly arranged;		[1]
	(d)	fas	id particles gain kinetic energy and move faster; test particles are able to overcome forces of attraction forces; test/most energetic/with most KE particles escape from liquid;		[max 2]
					[Total: 10]
4	(a)		expressed in the presence of the dominant allele/not expressed when	nen	
			erozygous ; ele requires the identical allele to be seen ;		[max1]
	(b)	(i)	NN, nn ;		[1]
		(ii)	Nn;		[1]
	(c)	Nn N, i	mal, normal ; , Nn ; n, N, n ; , Nn , Nn , nn (in Punnett square) ;		
		3:1	;		[max 4]
	(d)		s able to find food/find a mate/escape predators; less likely to survive/reproduce;		[2]
					[Total: 9]
5	(a)	(i)	W ; contains nitrogen (with C, H and O) ;		[2]
		(ii)	Y and Z; contain only hydrogen and carbon;		[2]
	(b)	(i)	ethene molecules contain double bond and/or ethane all single bo	nds;	[1]
		(ii)	(with ethane) no change / no reaction; (with ethene) bromine solution decolourised;		[2]
	(c)	(i)	(addition) polymerisation/self addition;		[1]

(ii) poly(ethene)/polyethene/polythene;

[1]

Pa	ge 4	1	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2015	0654	21
	(d)	(i)	carbon dioxide ;		[1]
		(ii)	goes milky/goes milky then clears;		[1]
					[Total: 11]
6	(a)		s of light pass through optical fibre ; gles approximately correct and all reflections occurring on fibre wall ;		[2]
	(b)		emical ; etic ;		[2]
	(c)		s ionising so less damage caused to tissue/can pass through tissue/corbed by tissue;	not /	[max 1]
	(d)		ays; wing bones in the body ; ow any correct electromagnetic wave and use)		[2]
					[Total: 7]
7	(a)	any	part of the nervous system <u>except</u> brain/spinal cord;		[1]
	(b)	(i)	response to a stimulus/response to protect body; immediate/automatic/without conscious thought;		[2]
		(ii)	carry impulses/AW from <u>receptors</u> to <u>CNS</u> ; carry impulses/AW from <u>CNS</u> to effectors/muscle; reference to sensory neurons/motor neurons;		[max 2]
	(c)		rried) in the blood ; stroyed by the liver ;		[2]
					[Total: 7]
8	(a)	(i)	weaker/less attraction where filler is;		[1]
		(ii)	no – aluminium is not magnetic ;		[1]
		(iii)	positive paint droplets attracted to negative panel because opposite attract; paint droplets repel each other and spread out because like charge		[2]

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(b) (i) all symbols correct;

two lamps connected in parallel with battery; switch in correct place;

[3]

(ii) voltmeter drawn across battery;

[1]

(iii) $(I =) \frac{V}{R}$;

$$= \frac{12}{2.5} = 4.8 \text{ (A)};$$

[2]

(iv) 1.25Ω ;

[1]

(v) charge/electrons;

[1]

(c) conduction/convection;

[1]

[2]

(d) to allow for expansion in hot weather;

flexible material fills gap but can be squashed;

[Total: 15]

9 (a) (i) electrolysis;

[1]

(ii) orange/brown vapour;

[1]

(iii) ions no longer mobile;

[1]

(b) reference to electron loss;

[1]

(c) (i) hydrogen;

[1]

(ii) lighted splint; 'pops';

[2]

(iii) solution becomes alkaline/sodium hydroxide is made;

[1]

(d) (i) alloys are stronger/less easily broken;

[1]

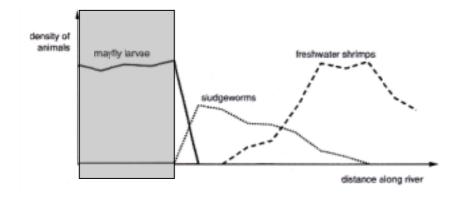
(ii) reduces the mass/weight of the aircraft;

[1]

[Total: 10]

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- 10 (a) mayfly larvae/caddis flies/freshwater shrimps/water lice/bloodworms; [max 1]
 - (b) (i) arrow anywhere in the shaded area;



[1]

(ii) (sewage) microorganisms; respiration deoxygenates water; which prevents respiration;

(chemical waste)

toxic;

heavy metal bioaccumulation;

[max 3]

(c) (i) increased temperature of the Earth; on average/at the Earth's surface;

(ii) burning of (carbon containing) fuels/named example;

[1]

[2]

(iii) reduced use of fossil fuels; public transport; alternative energy sources; planting trees/controlling defo

planting trees/controlling deforestation;

education/taxation/public awareness measures;

[max 2]

[Total: 10]

11 (a) force/weight; (vertical) distance; [2]

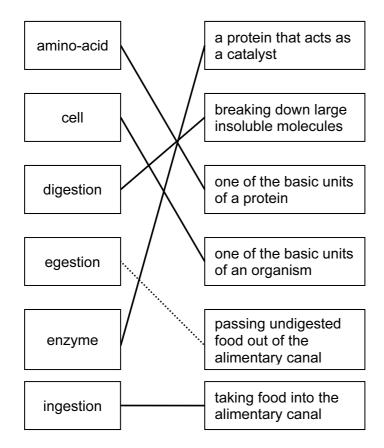
(b) (density =)
$$\frac{\text{mass}}{\text{volume}}$$
;
= $\frac{4000}{3.9}$ = 1026 (kg/m³);

[Total: 10]

Page 1	7	Mark Scheme	Syllabus	Paper
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(c)	(i)	(10Hz – no mark) lowest frequency detected is 20Hz;		[1]
	(ii)	number of waves produced/passing a point per second;		[1]
	(iii)	sound wave – arrow in same direction as wave movement; water wave – arrow perpendicular to wave movement;		[2]
				[Total: 8]
12 (a)	-			
	•	lrochloric acid ; bon dioxide + water ;		[3]
(b)	(i)	endothermic;		[1]
	(ii)	reference to use of a suitable indicator/pH meter; correct neutral colour/pH 7;		[2]
(c)	dec	crease acid concentration ;		
, ,	dec	crease (acid) temperature ; crease surface area / use larger pieces of calcium carbonate ;		[3]
(d)	refe	erence to decreasing acidity/neutralising acidic lake water;		[1]

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13 (a)



5 **or** 4correct = 4 marks, 3 correct = 3 marks, 2 correct = 2 marks, 1 correct = 1 mark [max 4]

(b) (i) anus; [1]

(ii) fibre/roughage/cellulose; [1]

[Total: 6]