CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the October/November 2015 series

0654 CO-ORDINATED SCIENCES

0654/23

Paper 2 (Core Theory), maximum raw mark 120

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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Page 2		Mark Scheme	Syllabus	Paper
		Cambridge IGCSE – October/November 2015	0654	23
1	(a) () fat ; protein ; calcium ;		[max 2]
	(i) iron ;		[1]
	(ii) has more fat ;		[1]
	(b) () (1.50) 15; 6;		[2]
	(i) no, because large amount is needed to meet vitamin C requirement	;	[1]
	(ii) bleeding gums ; poor skin/bruising ; scurvy ;		[max 2]
	(c) () prevents constipation/promotes peristalsis ;		[1]
	(i) (named) cereal grain/fruit/vegetable ;		[1]
				[Total: 11]
2	(a) () idea of greater precision/accuracy;		[1]
	(i) neutralisation ;		[1]
	(ii) salt ; water ;		[2]
	(b) () (first 35 cm ³) decreased slowly/decreased from pH 13 to 12 ; (next 10 cm ³) decreased rapidly/more quickly/decreased from pH12	2 to 2 ;	[2]
	(i) 40 (cm ³) ; evidence of finding the volume at pH = 7 ;		[2]
	(ii) take same amount/20.0 cm ³ of alkali ; add 40 cm ³ of the acid <i>(allow ecf from (ii))</i> ;		[2]
	(iv) white solid/solid sodium chloride ;		[1]
				[Total: 11]

Page 3		3	Mark Scheme							Pap	Paper	
3	(a)	(i)		Campri		October	NOVEINDEL	2013	0054	23)	
	[(ga	mma)	X-rays	ultraviolet	(visible)	infra-red	(microwaves)	radio wa	ives]	
	_		(all fo	our correct	2 marks, any	two correc	:t 1 mark) ;;			[2]		
		(ii)	micro	waves;							[1]	
	(b)	(i)	label	line at bas	e of fire/labe	l line where	e both rays	meet ;		[1]		
		(ii)	55 (m	nm) ± 1 mm	ו;					[1]		
	(c)	par coll forc	ticles <u>c</u> lide wit ce of co	es <u>constantly</u> in motion ; with walls of container ; of collisions exerts a pressure ;							ax 2]	
	(d)	wei (su	ight (of rface) a	^f penguin) area of foc	; t/feet ;							
	(e)	dia (if J	gram E A or C	3 (no mark – 0 marks) particles are even with co	e touching rrect expla	and random nation)	nly arranged ;		[Total	[1] I: 10]	
4	(a)	(i)	magn	nesium + c	arbon dioxide	\rightarrow magne	sium oxide	+ carbon ;			[1]	
		(ii)	oxida magn carbo	tion is gair lesium gair on dioxide l	n of oxygen a ns oxygen an loses oxygen	nd reductio d is oxidise and is red	on is remova ed ; uced ;	al of oxygen ;		[m	ax 2]	
	(b)	(i)	anode	e clearly la	belled ;						[1]	
		(ii)	chlori Cl _{2;}	ne ;							[2]	
	(c)	(i)	carbo carbo	on ; on dioxide ;	;						[2]	
		(ii)	test th	he electrica	al conductivity	/ of the pro	duct/lead v	vill conduct elec	tricity ;		[1]	
										[Tota	al: 9]	

Page 4		4	Mark Scheme	Syllabus	Paper	
			Cambridge IGCSE – October/November 2015	0654	23	
5	(a)	(i)	asexual ;		[1]	
		(ii)	no gametes / fertilisation involved ;		r 41	
			genetically identical ;		[max 1]	
	(b)	(i)	photosynthesis ;		[1]	
		(ii)	sexual reproduction ;		[1]	
	(c)	(i)	anther/stamen;		[1]	
	. ,	(ii)	sepal :		[1]	
		(,			[.]	
	(d)	be	cause the fruits develop from the flowers ;		[1]	
					[Total: 7]	
6	(0)	(1)	(X) marked on graph at 12, 14 a, 71 a, 105 a and 150 a c		[4]	
0	(a)	(1)	crosses (X) marked on graph at 15–14 s, 71 s, 105 s and 150 s ,		[']	
		(ii)	13–14 (s) ;		[1]	
		(iii)	20 (s) ;		[1]	
		(iv)	C–D or G–H ; graph goes down ;		[2]	
	(b)	(i)	thermal energy produces increased particle vibration ;			
			particle vibration is passed on from particle to particle ; metals are good thermal conductors ;		[max 2]	
		(ii)	gas around filament heats up/gas expands ;		501	
			gas rises/gas less dense ;		[2]	
		(iii)	<i>wavelength:</i> distance between two waves ; but distance between two peaks/two troughs/two identical points of	on consecu	tive	
			waves ; frequency: number of waves produced per second/number of wave	es passing a	a	
			fixed point per second ;		[3]	
	<i>,</i> .	<i>(</i> 1)	voltage			
	(c)	(i)	(current) =; resistance			
			$=\frac{12}{4}=3$ (A);		[2]	
		(ii)	8 (Ω) ;		[1]	
					[Total: 15]	

Page 5		5	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2015	0654	23
7	(a)	xyl	em ;		[1]
	(b)	eva fro foll by	aporation of <u>water</u> ; m surfaces of mesophyll cells ; owed by loss of <u>wate</u> r vapour ; diffusion ; t through stomata :		[max 4]
		50			[IIIIIAN F]
	(c)	(i)	(coloured) water does not move as far ;		[1]
					[Total: 6]
8	(a)	pet fra	troleum ; ctional distillation ;		[2]
	(b)	(i)	carbon dioxide ; water ;		[2]
		(ii)	reference to carbon monoxide/incomplete combustion ; which are toxic/which could poison people ;		[2]
	(c)	(i)	hydrocarbon will decolourise bromine ; if it is unsaturated ;		[2]
		(ii)	н Дин		
			$c = c$ H H $;$ carbon – carbon double bond ; $4 \times H$ and all else correct ;		[2]
					[Total: 10]
9	(a)	no	resultant force because constant speed ;		[1]
	(b)	thr hoi inte	ee straight lines ; rizontal lines from boat and into eye ; ernal reflection shown at both prisms ;		[3]
	(c)	(i)	lead/concrete/aluminium;		[1]
		(ii)	geiger counter/GM tube, etc.;		[1]
					[Total: 6]

Page 6		6	Mark Scheme	Syllabus	Paper
			Cambridge IGCSE – October/November 2015	0654	23
10	(a)	(i)	deforestation ;		[1]
		(ii)	logging ; building of roads/towns/factories ; farming ; fuel ;		[max 2]
	(b)	cor (ca alte AV	ntrol of hunting/nature reserve/conservation area ; ptive) breeding programmes ; ernatives to timber/control of deforestation/replanting ; P ;		[max 2]
	(c)	(i)	grow/photosynthesise more (because not eaten by okapis);		[1]
		(ii)	have less food/must find alternative food sources ; (accept: more competition for food/migration)		[1]
					[Total: 7]
11	(a)	(i)	neon ;		[1]
		(ii)	proton/atomic number/number of electrons;		[1]
		(iii)	9 protons ; 10 neutrons ;		[2]
	(b)	(i)	sodium chloride ;		[1]
		(ii)	reference to loss of electron(s)/loss of outer shell ;		[1]
		(iii)	balance of charge/protons and electrons in the atom ; excess of electrons in the ion/gains electrons ;		[2]
	(c)	silv whi	er nitrate ; te precipitate ;		[2]
					[Total: 10]
12	(a)	(i)	5000000 (N) ;		[1]
		(ii)	need positive resultant, for upward motion/acceleration;		[1]
		(iii)	chemical, thermal (heat), kinetic ;; (all three for 2 marks, any two for 1 mark)		[2]

Page 7	Mark Scheme Syllabus				
	Cambridge IGCSE – October/November 2015	0654	23		
(b) (i)	sound waves cannot travel through space/vacuum or sound waves medium ;	s need a	[1]		
(ii)	speed = $\frac{\text{distance}}{\text{time}}$;				
	$= \frac{225000000}{750} = 300000(\text{km/s});$		[2]		
(c) (i)	ionising radiation that humans are exposed to/radiation that is alwa	ays there ;	[1]		
(ii)	rocks ;		[1]		
			[Total: 9]		
13 (a) (i)	increased rate of breathing ; increased depth of breathing/volume of breaths ;		[2]		
(ii)	less oxygen/O ₂ ; <i>(reject: no oxygen)</i> more carbon dioxide/CO ₂ ;				
	more water vapour ;		[max 2]		
	warner,				
(b) (i)	increased heart/pulse rate ;				
	increased blood glucose ; AVP ;		[max 2]		
(ii)	chemical/substance produced by a gland ;				
	carried in the blood ; alters the activity of target organ(s) :				
	destroyed by the liver ;		[max 3]		
			[Total: 9]		