CAMBRIDGE INTERNATIONAL EXAMINATIONS

Cambridge International General Certificate of Secondary Education

MARK SCHEME for the October/November 2015 series

0654 CO-ORDINATED SCIENCES

0654/51 Paper 5 (Practical), maximum raw mark 45

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(a)	(i)	(any) blue/no change;	[1]
	(ii)	colourless/like water/clear; (ignore: stayed the same)	[1]
(b)	(i)	turns white/pink AND indicates water is produced/present;	[1]
	(ii)	turns milky/cloudy/white ppt.; (allow: murky) (indicates) carbon dioxide/CO ₂ ;	[2]
(c)		nt produced/temperature increase ; t produced/glows/fire/flame/smoke ;	[2]
(d)		ontrol/show that water not already present/show that carbon dioxide not already sent;	[1]
(e)	(i)	respiration;	[1]
	(ii)	glucose/food/cheese + oxygen (not air) → carbon dioxide + water LHS correct = 1 ; RHS correct = 1 ;	[2]
(f)	bur	igles/hair tied back/Bunsen at safe distance/keep maximum distance from ning food/accept other sensible suggestions; nore: test-tube holders as in diagram)	[1]
(g)	(i)		[max 2]
	<i>(</i> 11)	(ignore: same time of burning)	F43
	(ii)	heat loss/incomplete burning;	[1]

[Total: 15]

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- 2 (a) (i) value of time greater than or equal to 10 s; [1] (allow: answers in minutes and seconds)
 - (ii) value within 10% of first value; both values to nearest second; [2]
 - **(b) (i)** Fe²⁺ value less than both values in **(a)**; [1]
 - (ii) Fe³⁺ value less than both values in (a), AND to nearest second; [1]
 - (iii) \mathbf{X}^{2+} value less than or equal to $5 \text{ s}/\mathbf{X}^{2+}$ value is 'instant'; [1]
 - (c) (i) at least four $\frac{1}{t}$ values calculated correctly (ignore s.f.); [1] (if t = 0 allow $\frac{1}{t}$ to be left blank or infinity but do **not** allow zero)
 - (ii) they are catalysts;
 - $\frac{1}{t}$ (rate) increased (with addition of metal ion)/time decreased (with addition of metal ion);
 - (d) reliable as within 10% (or other suitable percentage or comment)

OR

not reliable as greater than 10% difference (or other suitable percentage or comment);

(answer must demonstrate an understanding of reliability) (ignore: references to accuracy)

- (e) (i) add 1 cm³ water/add 5 drops + 1 cm³ starch;
 (do NOT allow: 0.5 cm³ more of A and 0.5 cm³ more of B)
 total volume should be same as in (b) / equivalent volume to metal ion/to
 keep concentrations the same;
 (mark independently)

 [2]
 - (ii) ppt. /white ppt. /cream ppt. /instant blue-black/instant reaction/more brown; [1]
- (f) blue ppt./dark blue solution;

 X is copper/Cu (depends on blue in first marking point);

 (allow: Cu²⁺ or copper(II) for second marking point)

 [2]

[Total: 15]

[2]

[1]

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3 (a) h AND D AND d recorded;

h > D > d;

all values to the nearest 0.1 cm;

 d_A calculation correct;

V calculation correct;

V given as whole number;

[6]

(b) (i) $V_{\rm w}$ correctly calculated with working shown, e.g. subtraction of two values; $V_{\rm w}$ is supervisor's value \pm 20 cm³ (can get this accuracy mark without calculation);

[2]

(ii) cup not completely full/measuring cylinder not read at eye level/measuring cylinder not read perpendicularly/measuring cylinder not read from bottom of meniscus/water spilled on transfer/R₂ off scale of measuring cylinder;

[max 1]

(iii) V_W since difficult to measure h/V_W since d (or D) not inside diameters V_W since it is a direct measurement V_W since V is an approximation V_W is actual measurement whereas V uses a formula;

[max 1]

(c) (i) evidence of at least 2 loops of string around cup; (this could be in words or from diagram and could be

(this could be in words or from diagram and could be in different positions or one position repeated)

correct averaging of two or more measurements for value of *C*; answer to 0.1 cm (*independent mark*);

[3]

(ii) diagram showing correct positioning of one loop, e.g. half way up/at top/at bottom;

[1]

(iiii) calculation correct to 2 or 3 s.f.;

[1]

[Total: 15]