## Cambridge International Examinations

Cambridge International General Certificate of Secondary Education

CO-ORDINATED SCIENCES
0654/61
Paper 6 Alternative to Practical
May/June 2016
MARK SCHEME
Maximum Mark: 60


This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

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1 (a) time/minutes; volume/ $\mathrm{cm}^{3}$;
(b) $6.8 ; 0.5$;
(c) both axes labelled at least one with units ;
linear scale covering $>1 / 2$ paper;
at least 4 plots correct $\pm$ half square ;
best fit line ;
(d) increases amount of juice produced/more juice per unit time ;
(e) keeps volume in each beaker constant/show that the water of enzyme solution does not have an effect/no effect without enzyme ;

2 (a) (i) 124;
(ii) C is $2.00 \mathrm{moldm}^{-3}$
$D$ is $0.50 \mathrm{moldm}^{-3}$
$E$ is $1.00 \mathrm{~mol} \mathrm{dm}^{-3}$;;
one correct $=1$ mark, three correct $=2$ marks
(b) add marble chip/add $\mathrm{UI} /$ add Mg ;
(marble chips or magnesium) count bubbles/collect gas/measure volume of gas ;
in a certain time ;
OR
(for marble chips) time ;
for limewater to go milky ;
OR
add NaOH from measuring cylinder/burette ;
until UI just green ;
the more bubbles or gas the more concentrated/the shorter the time (for limewater) the more concentrated/the more NaOH the more concentrated ;
equal volumes of the acids (in test-tubes) ;
(c) (acidified) silver nitrate/ $\mathrm{AgNO}_{3}$ AND white ppt. ;
(d) too long for magnesium to disappear/reaction too slow ;
[Total: 10]

| Page 3 | Mark Scheme | Syllabus | Paper |
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3 (a) $p=29.5 \mathrm{~cm}$;
(b) $x$ values correct (e.c.f. p)
24.5 ecf, (21.8), 19.1, 16.4, 13.6 ;
$y$ values correct
20.5 ecf, (18.2), 15.9, 13.6, 11.4 ;
(c) (i) suitable choice of scales $\geqslant 1 / 2$ the grid (can plot the 5 points) used AND linear ; minimum 4 plots correct to $1 / 2$ small square on easy to read scale ; good best fit straight line judgement ;
(ii) indication on graph of how the data were obtained AND more than half the line ; calculation correct ;
(d) $m$ correct to $2 / 3$ significant figures ;
(e) Any one from:
difficulty in obtaining balance ;
centre of mass of rule not at the 50.0 cm mark ;
load not uniform ;
difficulty in placing the centre of load over the mark on the rule ;
[Total: 10]

4 (a) placed in the dark;
at least 24 hours ;
(b) (i) potassium hydroxide/sodium hydroxide/soda lime;
(ii) any in the same state as (i) that does not absorb $\mathrm{CO}_{2}$;
(c) (i) iodine solution;
boiling/hot water ;
hot alcohol/ethanol ;
rinse with water ;
(safety) water bath/not naked flame ;
(ii) G is blue-black AND F is brown/orange ;
(because) G can photosynthesise and F cannot (photosynthesise) ;

## OR

$F$ is brown no photosynthesis ;
G is blue-black can photosynthesise ;
[Total: 10]

| Page 4 | Mark Scheme | Syllabus | Paper |
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5 (a) (i) limewater;
white ppt. ;
(ii) diagram showing filter funnel and paper ;
two relevant labels ;
(iii) blue ppt. AND blue ppt. ;
(deep) blue solution ; blue ppt. ;
(b) copper carbonate $/ \mathrm{CuCO}_{3}$;
(c) use of (acidified aqueous) barium chloride/barium nitrate ; white ppt. etc. ;

6 (a) (i) 112 ;
(ii) correct symbol for ammeter and voltmeter ; ammeter in series and voltmeter in parallel ; correct symbols for lamp and switch in series ; workable circuit (no short circuits, no gaps) ;
(iii) 54 and 21 ;

33 (ecf) ;
(iv) 112 (ecf) $\times 33(\mathrm{ecf}) \times 4.2 / 1000=15.5 / 16$;
(b) air/surroundings;
wires/leads/(heater) casing / circuit ;
AVP e.g. heat transferred to: beaker/used in evaporation ;
[Total: 10]

